The Investigation of Entrainment Limit in Operating Heat Pipes Nancy K. Tsai University Undergraduate Fellow, 1989-90 Texas A&M University Department of Mechanical Engineering

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ABSTRACT

A heat pipe is a two-phase flow heat transfer device that utilizes the latent heat of vaporization of a fluid. It is a sealed container that is lined with a wick structure which is saturated with a selected working fluid. Heat input at one end raises the temperature, which creates an axial pressure gradient and causes the vapor to flow to the opposite end, where heat is rejected and the vapor recondenses. The capillary pumping action provided by the wick structure maintains the circulation of the fluid. At steady-state, the operation of the heat pipe is limited by the occurrence of several phenomena--boiling in the wick, the capillary limit, the sonic limit or the choking of the vapor flow, and the entrainment of liquid droplets in the vapor flow. Of these four limits, the entrainment limit is the least understood. The analytical studies available to date yielded four different equations that predict significantly different maximum heat transfer rates at the onset of entrainment. Therefore, it was the objective of this investigation to compare experimental results to the entrainment limits predicted by these four equations.

In this experiment, liquid droplet entrainment was monitored using a laser beam travelling down the length of the vapor channel to investigate qualitatively the occurrence of entrainment. A graph of the results indicated that the heat flux at the entrainment limit may be at least as great as the lowest value predicted by the four analytical equations. More tests are being conducted to determine the actual entrainment limit in the heat pipe.

The occurrence of slugging affected the heat pipe's operation. Further studies to analytically determine the slugging limit will help in understanding this behavior in an operating heat pipe.

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NOMENCLATURE

A	area (m)
D	diameter (m)
g	gravitational acceleration (m/s2)
hfg	heat of vaporization (kJ/kg)
K	thermal conductivity of the wick (W/mK)
L	length (m)
Р	pressure (kPa)
Q	heat flux (W)
r	radius (m)
Т	temperature (°K)
V	velocity (m/s2)
We	Weber number
Δ	change in, difference
λ	wavelength associated with the liquid surface (m)
ρ	density (kg/m3)
σ	Surface tension (N/m2)

Subscripts

с	capillary,		
С	condenser		
cm	capillary, maximum		
e	evaporator		
h	hydraulic		
1	liquid		
n	nucleation		
norm	normal, perpendicular		
S	sonic		
v	vapor		
w	wick		

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OBJECTIVE

It is the objective of this investigation to compare the experimental heat flux with the maximum heat transfer rate at the entrainment limit predicted by analytical equations for an operating heat pipe.

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INTRODUCTION

Two types of heat transfer devices that utilize the latent heat of vaporization of a fluid and two-phase flow of liquid are the thermosyphon and the heat pipe. In both thermosyphons and heat pipes the working fluid is chosen such that the operating temperature lies between the critical state and the triple state of the fluid. In the thermosyphon, the condenser is located above the evaporator. Heat input results in the upward flow of the vapor. The vapor then recondenses in the condenser section and is returned to the evaporator by gravity assistance. The use of the gravitational force as the liquid return mechanism in thermosyphons thus dictates that the thermosyphon must be oriented vertically.

Like the thermosyphon, heat input at the evaporator of the heat pipe results in a temperature rise and a corresponding axial pressure gradient, causing the vapor to flow from the evaporator to the condenser. At the condenser, the vapor recondenses and returns to the liquid stream. Unlike the thermosyphon, the heat pipe consists of a sealed container lined with a screen or wick structure. The heat pipe is charged with enough water to saturate the wick. Also, the heat pipe relies on the capillary pressure provided by the wick structure rather than gravitational force to pump the liquid back to the evaporator. Therefore, heat pipes are not limited to the vertical configuration of a thermosyphon.

While thermosyphons have been dated back to the nineteenth century, the first heat pipe patent was awarded to R. S. Gaugler (1944), and the first application of heat pipes was not implemented until twenty years later, at the Los Alamos Scientific Laboratory (Grover, et al., 1964). The heat pipe has several inherent advantages, the most important of which is that heat can be transported a long distance with only a small temperature drop. Therefore, the heat transfer becomes more efficient, from 40 to 1000 times that of a copper rod of the similar size. (Chi, 1976)

Heat pipes have several other advantages. They are much lighter than the solid metal conductors of the same size. In space applications, the light weight of a heat pipe becomes an advantage. Also, a heat pipe is essentially a self-contained, sealed unit that is maintenance-free and requires no external pump. Heat pipes can be used in a wide range of temperatures by selecting the proper working fluid, a cryogenic fluid for low temperature and liquid metal for high temperature applications. In addition, heat pipes can be fabricated into a variety of shapes and sizes.

The applications of heat pipes are varied and numerous. Some of their uses are thermal switches, thermal diodes, and thermal shields. For the Trans-Alaska Pipeline System, a maintenance-free heat transfer device was required to conduct heat away from the permafrost to protect the environment and to prevent the structure from sagging under the melting ice (Chi, 1976). Heat pipes became the best answer for this problem. Recently, Faghri, et al. (1989), developed gloves that use heat pipes to transfer heat from the elbow to the fingertips to prevent frost bite in an extremely cold environment. For their adaptability to different needs, low maintenance cost, and other advantages, heat pipes are being considered in a variety of applications which require a more efficient transfer of heat.

LITERATURE REVIEW

Heat Pipe Operation

The heat pipe typically is divided into three regions, evaporator, adiabatic, and condenser. Heat addition occurs in the evaporator and results in a temperature and pressure rise. In the adiabatic, no heat transfer with the surrounding occurs, and in the condenser, heat is rejected. The pressure rise in the evaporator as a result of the heat addition creates a pressure gradient along the length of the heat pipe. This pressure gradient drives the vapor flow from the evaporator to the condenser region where heat rejection results in the condensation of the vapor. The condensed liquid then returns to the evaporator by the capillary pumping force provided by the wick structure. Fig. 1 shows the diagram of a conventional heat pipe.

The capillary pressure difference in the heat pipe wick structure results from a difference in the capillary radius in the wick as shown in Fig. 2. In the evaporator, the vaporization of the working fluid results in the liquid meniscus receding into the wick structure. The capillary radius of the pores then decreases. At the condenser, the condensation process deposits more liquid at the interface, resulting in a larger capillary radius. The maximum capillary pressure difference is described by the following equation:

$$\Delta P cm = \frac{2\sigma}{r_{cc}} - \frac{2\sigma}{r_{ce}}$$
(1)

This difference in the capillary radius provides the pumping mechanism which returns the condensed liquid from the condenser to the evaporator.

Operating Limits

During the steady-state operation of the heat pipe, the heat transfer capacity can be limited by the occurrence of several different phenomena, four of which are the boiling limit, capillary limit, sonic limit, and the entrainment limit. Fig. 3 shows a plot of these limits as a function of temperature. At low rates of heat transfer, only the liquid at the liquid-vapor interface is vaporized. When heat transfer rate to the evaporator increases,











Fig. 3 Steady State Heat Transport Limits as a Function of Temperature



Temperature

boiling in the wick structure and in the liquid channel can occur. The bubbles in the wick and liquid channel impede the return of the liquid to the evaporator if the pore size of the wick structure is not large enough for the vapor to escape. The heat transfer rate at which boiling occurs in a heat pipe is defined as

$$Qmax = \frac{2 \Pi \text{ Ke Le Te}}{h_{fg} \rho_{V} \ln (r_{i} / r_{V})} \qquad (2)$$
(2)

where r_n is the critical radius at which nucleation occurs and λ is equal to the wire diameter plus the wire spacing in the wick structure. The boiling limit, therefore, depends on the heat transfer rate to the evaporator, the wick structure, and the properties of the working fluid. As boiling continues, and the heat input is further increased, the liquid return may be completely blocked, resulting in the wick dry-out in the evaporator section.

The capillary limit refers to the inability of the wick structure to provide enough capillary pressure difference to pump the liquid from the condenser to the evaporator. During the normal operation of a horizontal heat pipe, the capillary pressure difference is equal to or greater than the total pressure due to friction in the vapor flow and liquid flow and the normal hydrostatic pressure drop. The axial hydrostatic pressure drop is negligible since no body force acts on the fluid. This capillary limit can therefore be expressed by the following equation:

$$\Delta P_{c} = \Delta P_{norm} + \Delta P_{l} + \Delta P_{v}$$
(3)

The capillary limit varies with the physical dimensions of the heat pipe, such as the type of wick and pore size, the fluid properties, heat flux, as well as the operating temperature of the heat pipe.

At the sonic limit, the vapor flow approaches the speed of sound, resulting in the choking of the flow. Chi (1976) compared this phenomenon to the choking of flow in a converging-diverging nozzle. In a converging-diverging nozzle, increases in velocity are achieved by varying the cross-sectional area. The highest vapor velocity is therefore, at the smallest cross-sectional area in a converging-diverging nozzle. In a heat pipe, the variation

in vapor velocity is caused by the rate of evaporation and condensation of the working fluid. The mass flow rate of the vapor and therefore the vapor velocity increases up to the end of the evaporator and remains constant in the adiabatic section. With condensation taking place and a lower temperature in the condenser, the mass flow rate decreases. If the condenser temperature is further decreased, the operating temperature of the heat pipe is suppressed until the vapor velocity reaches the sonic velocity and the flow is choked. Further increase in heat rejection does not eliminate the choked flow condition. The sonic limit is predicted by Busse (1973) to be

$$Q_{\rm S} = 0.474 \, A_{\rm v} \, h_{\rm fg} \, P_{\rm v,e} ^{1/2} \, \rho_{\rm v,e}. \tag{4}$$

The sonic limit is the maximum possible heat transfer that can be achieved at a given operating temperature, wick structure, and working fluid.

The fourth operational limit on heat pipes is the entrainment limit. It is the least understood of the four operating limits and is the topic of this investigation. In a heat pipe, the flow of vapor opposes the flow of liquid. The two opposing flows create a shear force on the liquid at the vapor-liquid interface that is proportional to the difference between the vapor velocity and the liquid velocity. When the difference is large enough, the shear force overcomes the surface forces of the liquid and tears off liquid droplets, and the vapor flow carries the droplets to the condenser. The maximum heat flux at the entrainment limit is a function of the properties of the working fluid and the physical dimensions of the heat pipe. The next section discusses in more detail the phenomenon of entrainment and the analytical equations developed to predict the entrainment in heat pipes.

Entrainment Limit

The entrainment of liquid droplets in the vapor flow in a heat pipe as a heat transfer rate limiting mechanism has not been thoroughly investigated. The entrainment limit studies available to date yield four different equations for the maximum heat transfer rate of a heat pipe at the start of liquid droplet entrainment. The first equation was developed by Cotter (1967) and is derived with the assumption that at the onset of entrainment the Weber number is equal to 1. The Weber number is the ratio of the viscous shear force to the forces resulting from surface tension, or

$$We = \frac{2r_{hw} \rho_v V_v^2}{\sigma}$$
(5)

The vapor velocity can be determined from the heat flux in this equation:

$$V_{\mathbf{V}} = \frac{Q}{A_{\mathbf{V}} \rho_{\mathbf{V}} h_{\mathrm{fg}}}$$
(6)

At entrainment, We = 1, and an expression for the heat flux at entrainment has the form

$$Q_{\text{max}} = A_{v} h_{\text{fg}} \left(\frac{\sigma \rho_{v}}{2 r_{\text{wh}}} \right)^{1/2}$$
(7)

this equation indicates that the entrainment limit in operating heat pipes is related to both the physical size of the heat pipe and properties of the working fluid.

Kemme (1976) modified Cotter's equation to include the effect of buoyancy on entrainment during vertical operation. He approximated the buoyancy effect by the term ρgD . The buoyancy term was derived from the Wallis (1969) criterion for flooding correlation for two-phase vertical flow. The modified equation for entrainment limit then becomes

$$Q_{\text{max}} = A_{v} h_{fg} \left(---- \left(----+ \rho_{1} g D \right) \right)^{1/2}$$

$$A^{*} \qquad \lambda$$
(8)

where λ , the wavelength associated with the liquid surface, is defined as the wire diameter plus the distance between two wires in a mesh, and A^{*} is defined as a dimensionless constant equal to 2.2 for turbulent flow.

The third equation for predicting entrainment limit was developed by Prenger (1984) for textured wall, gravity assisted heat pipes. The textured wall partially protects the liquid flow from the vapor flow, therefore minimizing the effect of the vapor inertia on entrainment. The liquid inertia, therefore, became the dominant inertial effect, and the equation becomes

$$Q_{\text{max}} = 2 \text{ A}_{v} h_{\text{fg}} - (-----)^{1/2}$$

$$D_{\text{pipe}} \pi D_{\text{pipe}}$$
(9)

This equation included the effect of both the working fluid's properties and the physical dimensions of the heat pipe on the maximum heat flux at entrainment limit. Prenger also stated that rather than flooding, the depth of the liquid channel limits the heat flux at entrainment.

With the exception of Prenger's equation, the entrainment limit correlations developed thus far presume that the wick structure was flooded. In theory, though, the wick was not flooded in normal heat pipe operation. The fluid is retained in the wick structure and the shear force at the vapor-liquid interface must then overcome the energy required to tear the liquid droplets from the wick, not the energy needed to tear liquid droplets from the surface waves that formed above the wick structure. Rice and Fulford (1987) developed an equation for the entrainment limit based on these assumptions.

$$Q_{\text{max}} = h_{\text{fg}} A_{\text{v}} \left(\frac{10}{1} \right)^{0.5}$$
(10)

where l, the characteristic length of the wick, is twice the radius of curvature of the largest sustainable drop of working fluid in the pores of the wick. At the time of writing the paper, this equation had not been experimentally verified.

Similar to a study conducted by Bage and Peterson (1989), the theoretical maximum heat fluxes predicted by these four equations have been calculated for the heat pipe apparatus used in this investigation, with dimensions given in Table 1. The calculated entrainment limits were graphed in Fig. 4. This plot shows that a large discrepancy existed among the data obtained from the four equations. This large discrepancy indicated that the entrainment limit is not well understood, and therefore requires further in-depth studies.

 Table 1: Dimensions of heat pipe apparatus for the calculation of theoretical heat flux at entrainment

	=====	=====	===
Wire Diameter	0.023"	(0.0584	cm)
Wire Spacing	0.105"	(0.267	cm)
Vapor Passage Width	1.05"	(2.67	cm)
Vapor Passage Height	0.32"	(0.813	cm)
	====		

THEORETICAL HEAT FLUX AT ENTRAINMENT LIMIT



Fig. 4: Theoretical Maximum Heat Flux at the Entrainment Limit as a Function of the Operating Temperature

Q, Heat Flux, (W)

EXPERIMENTAL APPARATUS

The experimental apparatus used in this investigation is shown in Fig. 5 and consists of a rectangular heat pipe with a copper wire screen for the wick and water as the working fluid. The physical dimensions of the heat pipe were previously calculated (Lorenz, 1989) to ensure that the entrainment limit would be reached before reaching the other steady-state operational limits. One of the physical dimensions that influences the onset of entrainment is the vapor passage cross-sectional area. For a given power input, a smaller vapor passage area gives a higher vapor velocity (or higher mass flow rate), therefore creating a higher shear stress at the vapor-liquid interface. A 40-mesh copper wire screen, with a wire diameter of 0.023" and a hydraulic radius of 0.053", was selected for this investigation because of its large pore size or hydraulic radius. The coarseness of the 40-mesh translates into a weaker surface tension retaining the fluid in the wick, therefore decreasing the force required to tear liquid droplets from the wire screen.

The heat pipe had a clear Lexan top to facilitate a window through which entrainment could be monitored visually. Two different plate thicknesses--0.125", and 0.375"--were used to provide different vapor passage cross-sectional areas, therefore different vapor velocities. A greased O-ring provided the vacuum seal between the lid and the heat pipe. The assembly was then bolted together with an aluminum guide on top of the Lexan to help distribute the pressure more evenly. At the condenser end of the heat pipe, a hole was tapped in the wall and plugged with clear Lexan to provide an entrance window for a laser beam. The laser beam traveled through the window in the wall and down the length of the heat pipe. In a pure vapor flow, the laser beam was invisible. When entrainment occurred, the beam was reflected by entrained liquid droplets in the vapor stream. The laser used for this investigation was an Aerotech LSR12MR.



Fig. 5: Experimental Apparatus

Thermal power was provided by a variable transformer and was measured by a digital wattmeter. Electrical resistance heating using nichrome wires embedded in silicon rubber provided heat to the evaporator. The condenser section of the heat pipe was cooled by circulating a 50-50 mixture of freon and water through a Forma-Scientific constant temperature cooling bath and then through a separate channel along the bottom of the condenser section. By adjusting the evaporator power input and/or the temperature of the condenser, different operating temperatures, or different adiabatic vapor temperatures of the heat pipe were obtained.

The experimental facility also included a vacuum pump for evacuating the heat pipe prior to charging the pipe with pure distilled water. The vacuum pump used for this investigation was a Centorr model 14-3x8T-16.5.

Data acquisition for this investigation was automated because of the fast response of the heat pipe. To obtain a representative instantaneous temperature and pressure profile, the readings was recorded in succession and as quickly as possible. A BASIC program interfaces a Hewlett Packard Vectra 37A personal computer with a Hewlett Packard model 3497A data acquisition system. Twenty-six holes were tapped along the length of the heat pipe, with thirteen holes in the liquid channel and thirteen holes in the vapor channel. Twenty-six sheathed copper constantan thermocouples were then inserted, thirteen in the vapor channel and thirteen in the liquid channel. To map the axial pressure distribution, thirteen Omega Series PX236 pressure transducers were mounted on the Lexan cover. The BASIC program was designed to convert the voltages from the pressure transducers and the thermocouples into pressure and temperature values, respectively.

EXPERIMENTAL PROCEDURE

The procedure used in this investigation was designed to imitate the commercial procedure for manufacturing heat pipes.Before charging the heat pipe with distilled water, a vacuum was drawn inside the heat pipe. The vacuum achieved inside the heat pipe was 0.0 torr, with initial leakage rate of 0.5 torr/ 6 min. Next, the heat pipe was back-filled with the proper amount of water to flood the wick structure, accounting for the condensation on the Lexan cover.

After charging the heat pipe, the constant temperature bath was started, and cooling fluid was circulated through the channel below the condenser. When the condenser reached a constant temperature, heat was supplied to the evaporator at a constant power setting. From the temperature-time profile for a test run plotted in Fig. 6, the heat pipe temperature stabilized within 30 minutes after the heat source had been applied to the evaporator section. Therefore, in the experimental procedure, the heat pipe was allowed to run at each power level for 30 minutes. Because of the known vacuum leaks, the heat pipe was vented of any noncondensable gas at the end of the stabilization time. To allow the heat pipe to respond to the changes in the conditions, temperature and pressure profiles were obtained 30 seconds later. When entrainment was observed or when the power input exceeded the theoretical maximum heat flux predicted from Eq. (7) - (10), the power setting and/or the condenser temperature were adjusted to obtain a higher adiabatic temperature. When entrainment was not observed, the power input and the cooling bath temperature were adjusted to maintain the same adiabatic temperature. The procedure was repeated at several temperatures.



Fig. 6: Transient Temperature Response of a Selected Thermocouple

OBSERVATIONS

While operating the heat pipe, the behavior of the working fluid was carefully monitored to help understand the operating of this device. During Trial 1, when the power level was raised to 177 watts, boiling in the liquid channel and in the wick was observed. The non-uniform wall thickness where thermocouples had been tapped into the wall provided a favorable nucleation site for boiling to initiate. Both Trials 1 and 2 exhibited some boiling in the liquid channel and the wick.

Boiling in the wick resulted in the retardation of liquid flow back to the evaporator. At higher heat input, this process led to the dry-out of the wick structure in the evaporator and decreases the heat transfer efficiency of the heat pipe.

Another behavior observed during the operation of the heat pipe in both Trials 1 and 2 is the "slugging" of water at higher power inputs--177 watts for Trial 1 and 103 watts for Trial 2. Figure 8 shows graphically the development of liquid slugging. Vapor bubbles and flooding in the wick created and sustained pools of water above the wick structure. Due to the small vapor passage area, 0.32" in height, the condensation on the lid was periodically in contact with the water in the liquid channel and formed a slug or a wall of water which blocked the vapor passage. Eventually, the pressure difference between the evaporator side of the slug of water and the condenser side pushed the liquid to the condenser end, sweeping away the condensed liquid on the Lexan lid. This slugging effect was most prominent immediately after venting the heat pipe when the axial pressure gradient was increased as a result of venting. Up to 10 slugs for the first minute were recorded for the first trial and 5 slugs during the first minute for the second trial.

Finally, in Trial 2, the working fluid inside a major portion of the condenser solidified, for a -30° C cooling bath temperature and 156 watt power input. The only section which remained in the liquid state was the first four inches of the condenser, adjacent to the adiabatic region. The freezing of the working fluid in the condenser

SLUGGING



Heat Input

Fig. 7: Development of Liquid Slugging

probably resulted from the accumulation of noncondensable gas over the condenser region which prevented the transfer of heat by vapor flow to maintain some adiabatic or operating temperature despite the increased power input to the evaporator. However, the accumulation of noncondensable gas also kept the evaporator from drying out, after the working fluid in the condenser solidified. If the noncondensable gas had not been present, the liquid which condensed out of the vapor in the condenser region would also have frozen into ice. Eventually, the return of liquid could completely stop, resulting in a dryout of the evaporator.

ANALYSIS OF RESULT

Noncondensable gases

Because of the known leakage, an effort was made during the operation of the heat pipe to track the increase in noncondensable gas inside the heat pipe. The presence of noncondensable gas in the condenser section of the heat pipe was manifested by the lack of condensation on the Lexan lid. During one trial, the noncondensable gas front was observed to move six inches in five minutes, increasing in the volume it occupied in the vapor space over the condenser region. Cracks in the Lexan lid contributed to the rapid deterioration of the vacuum inside the heat pipe. For this reason, the temperature profile at 30 seconds after venting the heat pipe was chosen to be more representative of the operating condition for the given power input, fluid, and condenser temperature.

Comparison with Analytical Results

Two sets of trials were conducted using the same vapor passage area and the same mesh size wire screen for the wick structure. The first set of trials, as plotted in Fig. 8 were completed by keeping the condenser temperature at 0° C. By increasing power input the operating temperature was thus increased. Higher heat fluxes could not be obtained, however, at the same operating temperature with the available power supply. Therefore, for the second set of trial runs, the condenser temperature was also adjusted to increase the heat flux without increasing the operating temperature of the heat pipe.

The thermocouples used for obtaining the temperature profile were designed for automated data acquisition systems. First, a reference voltage was recorded by the Hewlett Packard 3497 A data acquisition unit. The voltages corresponding to each thermocouple reading were then converted into temperatures in degrees Celsius using the reference voltage and other constants previously determined by the manufacturer. Manufacturer specifications indicated that the total accuracy for temperature readings with the Hewlett Packard 3497A was $\pm 0.009^{\circ}$ C/°C.





Temperature Compared to Analytical Heat Flux at Trial I Experimental Heat Flux vs. Operating Fig. 8:

Temperature (°K)

<u>Trial 1</u>. Entrainment was not observed at any of the power levels tested. Therefore, the entrainment limit must be greater than the heat fluxes tested, or possibly at least as high as the entrainment limits predicted by Cotter and Prenger. It was discovered after the trials that the heat pipe was tilted slightly with the condenser above the evaporator, which provided gravity assistance in returning the condensed liquid back to the evaporator. The tilting of the heat pipe could have resulted in a higher heat flux at a given operating temperature because of the enhanced liquid return to the evaporator.

<u>Trial 2</u>. Fig. 9 plots experimental heat flux versus operating temperature for the various combinations of power input and condenser temperatures selected to maintain a relatively constant adiabatic temperature. For all combinations, entrainment was not observed. When superimposed on the analytical results, the plot indicates that the experimental heat flux at the selected adiabatic temperatures fell below the lowest values of heat flux at the entrainment limit as predicted by Cotter's and Prenger's equations. This graph thus indicates that the heat transfer at the entrainment limit must be higher than the experimental heat flux obtained at the adiabatic temperatures tested.

Trial 1 vs. Trial 2. The data obtained from the two sets of trials both indicate that the heat flux at the entrainment limit must be greater than the heat flux at the temperatures tested and may be as high as the heat fluxes predicted by the analytical equations. For the same condenser temperature, the operating temperature obtained for Trial 1 is lower than that of Trial 2. Three possible reasons could account for the discrepancies observed. The first possibility is the amount of noncondensable gas present in the heat pipe. The accumulation of noncondensable gas was more than likely different between the first trial and the second trial. In addition, liquid return to the evaporator was gravity assisted in the first trial, whereas for the second trial the liquid return was only facilitated by the capillary pumping force. Therefore, a higher heat transfer rate was achieved in the first trial than in the second trial at the same operating temperature. In other words, the operating temperature is lower in the first trial than in the second trial for a constant power input. The





Fig. 9: Trial 2 Experimental Heat Flux vs. Operating Temperature Compared to Analytical Heat Flux at Entrainment Limit

Temperature (°K)

Q, Heat Flux, (W)

main reason for the difference in results is the difference in the amount of compression of the O-ring. The O-ring provides the vacuum seal between the lid and the heat pipe. The more the O-ring is compressed, the smaller the vapor passage. One of the factors in predicting entrainment is the vapor passage area. The smaller the vapor passage area, the higher the velocity, and the faster entrainment will occur.

In Trial 1, the conditions of the experiment were not controlled well, therefore the data obtained may not be as representative of the operation of the heat pipe as the second trial.

Static vapor pressure profiles. An analysis of the vapor pressure profile was not possible due to the large uncertainty involved in the instrument and in the pressure measurement. The pressure transducers used to obtain the static vapor pressure profile were designed to measure gage pressure with a remote atmospheric pressure reference. These transducers were assumed to have a linear pressure-voltage relationship for the range of pressures concerned--from -14.7 psig to 30 psig. Each pressure transducer was calibrated at these two pressures. The slope and intercept values for each pressure were used to determine an equation for converting voltage readings to pressures. Manufacturer's specification indicated that the deviation for a 25 degree increase or decrease from room temperature (25°C) was ± 3 mV. With a 10V dc voltage excitation, the typical voltage reading at -14.7 psig was 40 mV. At 0 psig, the voltage reading was typically 0 mV. The voltage/psi ratio was therefore approximately 2.72. The ± 3 mV deviation translated into a pressure uncertainty of ± 0.1 psi. A meaningful pressure profile, therefore, could not be obtained. A sample pressure profile has been included in the appendix.

CONCLUSION/RECOMMENDATIONS

An examination of the experimental data indicated the maximum heat flux may be as high as the minimum values predicted by the analytical equations, for heat pipes in which the wick structure has been flooded. Due to time constraints and troubleshooting, not enough data points were available to determine the actual entrainment.

Several of the steady-state operational behavior characteristics which were observed in this heat pipe included boiling, temporary wick dry-out, and slugging. The non-uniform thickness of the bottom wall provided nucleation sites for boiling to occur. A temporary wick dry-out was observed at high evaporator power input and probably was a result of boiling. Slugging was present also at power inputs as low as 103 watts for one set of trials. All three operational behaviors affected the efficiency of the heat pipe.

It is recommended that more tests be run to determine the actual entrainment limit. In conducting further experiments, more care should be taken in controlling the operating conditions. The compression of the O-ring must be carefully controlled, since a slight variation will change the vapor passage area of the heat pipe. Also, the procedure may need to be modified to vent the heat pipe more than once to withdraw the noncondensable gases more effectively.

The observed slugging affected the operation of the heat pipe. It was a behavior previously suspected in other heat pipes. Further analytical studies should be conducted to quantify the effect of slugging on heat pipe operation.

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REFERENCES

- Bage, B. K. and Peterson, G. P., 1989, "A Review of Entrainment Limits in Thermosyphons and Heat Pipes," ASME HTD Vol. 117, ASME Winter Annual Meeting, San Francisco, CA, December 10-15, pp. 1-8.
- Busse, C. A., 1973, "Theory of the Ultimate Heat Transfer Limit of Cylindrical Heat Pipes," Int. J. Heat Mass Transfer, Vol 16, pp. 169-186.
- Chi, S. W., 1976, Heat Pipe Theory and Practice, McGraw-hill Publishing Co., New York, N. Y.
- Cotter, T. P., 1967, "Heat Pipe Startup Dynamics," Proc. of the SAE Thermionic Conversion Specialist Conf., Palo Alto, CA.1967
- Faghri, A., Reynolds, D. B., and Faghri, P., 1989, "Heat Pipes for Hands," Mechanical Engineering, Vol. 102, No. 6, pp. 70-74.
- Gaugler, R. S., U. S. Patent Application (Dec 21, 1942). Published U. S. Patend No. 2350348. (June 6, 1944)
- Grover, G. M., Cotter, T. P., and Erickson, G. F., 1964, "Structures of Very High Thermal Conductance," J. Appl. Phys., 218, 1990-1991.
- Kemme, J. E., 1976, "Vapor Flow Consideration in Conventional and Gravity-Assist Heat Pipes," Proc. of the 2nd Int'l. Heat Pipe Conf., pp. 11-21.
- Lorenz, S. T., 1989, "Development of a Facility for Observing the Entrainment Limit in an Operating Heat Pipe," Texas A&M University, College Station, TX.
- Prenger, F. C., 1984, "Performance Limits of Gravity-Assist Heat Pipes," Proc. of the 5th Int'l. Heat Pipe Conf., pp 1-5.
- Rice, G. and Fulford, D., 1987, "Influence of a Fine Mesh Screen on Entrainment in Heat Pipes," *Proc. of the 6th Int'l. Heat Pipe Conf.*, pp. 168-172.
- Wallis G. B., 1969, One-dimensional Two-phase Flow, McGraw-Hill Book Company, New York, N. Y.

APPENDIX

- A. BASIC program for reading temperatures and pressures
- B. Data sheets for Trial 1
- C. Data sheets for Trial 2
- D. Microsoft Excel spreadsheet for the calculation of theoretical heat flux at entrainment
- E. Ideal plot of experimental heat flux versus operating temperature
- F. Vapor temperature profile for varying power input and condenser temperature

A. BASIC program for reading temperatures and pressures

```
10 DATA 120,159,4 ! THESE ARE TC LOCATIONS & RANGE
20 READ First_tc,Last_tc,N_readings
30 CLEAR 709
40 PRINTER IS 1
50 OUTPUT 709;"SA"
                      ! BEEP?
60 OUTPUT 709; "SDO" ! TURN OFF DISPLAY FOR FASTER READING
70 DIM Temp(120:160,4), U(120:160,4), Tave(120:160), T loc(120:160), Tsub(120:160)
80 DIM No sec(1:10), B(1:14), Zer(1:14), Volt(1:14,4), Pave(1:13), Slope(1:14)
90 !DATA -40.079, -41.097, -39.595, -40.777, -39.11, -38.667, -38.577, -38.33, -38.371, -
38.772, -38.173, -38.446, -38.359
91 LINPUT "CALIBRATE PRESSURE TRANSDUCERS? (Y/N)", Cal$
92 IF Cal$="Y" OR Cal$="y" THEN GOTO 1680
93 ASSIGN @Path1 TO "IVAC"
100 FOR I=1 TO 14
110 ENTER @Path1;Zer(I)
130 NEXT I
131 LINPUT "READ VOLTAGE AT VACUUM? (Y/N)", Call$
132 IF Call$="Y" OR Call$="y" THEN GOTO 1680
140 !DATA .271, -.454, .376, -.374, -.329, .076, .099, .623, .457, .258, .631, .437, .475
141 ASSIGN @Path1 TO "IVOUT"
150 FOR I=1 TO 14
160 ENTER @Path1;B(I)
165 \text{ Slope}(I) = (Zer(I) - B(I)) / (-14.7)
170 NEXT I
180 R0=5.25792984E-7 ! SET UP CONSTANT VARIABLES FOR THERMOCOUPLE
190 R1=3.860071243E-5 ! TEMPERATURE CALCULATIONS
200 R2=4.186486602E-8
210 PO=.1238117795
220. P1=26861.17637
230 P2=-896494.288
240 P3=-46489260.88
250 P4=12441142450
260 P5=2.275304922E+12
270 P6=-6.399496867E+14
280 P7=5.435757807+16
290 P8=-2.02361537E+18
300 P9=2.830121167E+19
310 OUTPUT 709; "AC39"
                                 ! READ IN REFERENCE VOLTAGE FOR THERMOCOUPLES
320 ENTER 709;R
330 !SETTING TIMED LOOP
340 PRINT "SETTING UP FOR TIMED DATA ACQUISITION..."
350 INPUT "ENTER THE COOLING BATH TEMPERATURE (IN DEGREES C) ", Bath
360 INPUT "ENTER THE MESH NUMBER ", Mesh
370 INPUT "ENTER THE POWER INPUT (IN WATTS) ",Watts
380 PRINT
390 INPUT "ENTER THE NUMBER OF TIMES TO TAKE DATA (<10)", No run
400 FOR I=1 TO No run
410 PRINT "NUMBER OF SECONDS DELAY BEFORE RECORDING TRIAL"; I
420 PRINT "(number of seconds between successive trials)"
430 INPUT No sec(I)
440 NEXT I
441 !
442 LINPUT "THE NAME OF THE DATA FILES WILL BEGIN WITH ... ?", Data$
450 ! Data$=CHR$(68)&CHR$(65)&CHR$(84)&CHR$(70)
                                                           !SETTING UP DATA FILES
460 Tc wall=First tc+9
470
       T loc(Tc wall+2)=5.6875
                                                  !CALCULATING THERMOCOUPLE
480
       T loc(Tc wall+1)=6.1875
                                                  !LOCATIONS IN INCHES FROM
490
       Sub2=T_loc(Tc_wall+2)
                                                  !THE EVAPORATOR END
       Subl=T loc(Tc wall+1)
500
510 REM CALCULATING LOCATIONS
```

```
520 FOR L=Tc wall+3 TO Last tc-2 STEP 2
530 T loc(L)=6+Sub1
        T_loc(L+1)=6+Sub2
540
       Subl=T_loc(L)
550
     Sub2=T_loc(L+1)
560
570 NEXT L
     T_loc(Last_tc-4)=77.1875
T_loc(Last_tc-5)=77.6875
580
590
600
             T$=TIME$(TIMEDATE)
610
             Start=TIMEDATE
620
              D$=DATE$(TIMEDATE)
630 FOR M=1 TO No run
631 CLEAR SCREEN
640 ON TIME (Start+No sec(M)) MOD 86400 GOTO 690
650 PRINT TABXY(27,11), "PLEASE WAIT..."
660 PRINT TABXY(27,13), TIME$(TIMEDATE)
670 PRINT TABXY(27,15), TIME$(Start+No_sec(M))
680 GOTO 650
690 File name$=Data$&CHR$(64+M)
710 FOR J=1 TO N readings
                                          ! READ IN THE VOLTAGES FROM ALL OF THE
                                         ! THERMOCOUPLES
720
    FOR I=First tc TO Last tc
740
          OUTPUT 709; "AC"; I
          ENTER 709; U(I,J)
750 NEXT I
762 FOR I=0 TO 14
                                          ! READ IN VOLTAGES FROM PRESSURE
                                         ! TRANSDUCERS
763 IF I=9 THEN GOTO 768
764 OUTPUT 709; "AC"; I
765 ENTER 709;V

        766
        IF I<9</th>
        THEN Volt(I+1,J)=V*1000

        767
        IF I>9
        THEN Volt(I,J)=V*1000

768 NEXT I
769 NEXT J
771 FOR J=1 TO N_readings! CONVERT THERMOCOUPLE VOLTAGES IN780 FOR I=First_tc TO Last_tc! TEMPERATURE IN DEGREES FARENHEIT
                                          ! CONVERT THERMOCOUPLE VOLTAGES INTO
790
          V0=R*10
800
          V1=R0+V0*(R1+V0*R2)
810
          V2=V1+U(I,J)
        T1=P5+V2*(P6+V2*(P7+V2*(P8+V2*P9)))
T1=P0+V2*(P1+V2*(P2+V2*(P3+V2*(P4+V
T2=32+1_8*T1
820
830
          T1=P0+V2*(P1+V2*(P2+V2*(P3+V2*(P4+V2*T1))))
840
          T2=32+1.8*T1
         Temp(I,J)=INT(T1*1000+.5)/1000
850
     NEXT I
860
870 NEXT J
880 FOR L=First_tc TO Last_tc
890 Tsum=0
900 FOR K=1 TO N readings
                                                       !CALCULATING AVG TEMP
                                                       !TO 3 DECIMAL PLACES
910 Tsum=Tsum+Temp(L,K)
920
      NEXT K
930
      Tsub(L)=INT(Tsum/N readings*1000+.5)/1000
940 NEXT L
941 FOR L=1 TO 13
                                                       !CALCULATING AVG PRESSURES
942 Vsum=0
943 FOR K=1 TO N_readings
944 Vsum=Vsum+Volt(L,K)!+Volt(14,K)
945 NEXT K
946 Vave=Vsum/N readings
947 Pave(L)=INT((Vave-B(L))/Slope(L)*10+.5)/10
948 NEXT L
                                                      !MAPPING THERMOCOUPLES TO
950
     FOR L=First tc TO First tc+9
                                                      !CHANNELS--1st 10 CHANNELS
960
      Tave(L) = Tsub(L)
```

```
970 NEXT L
                                                 ! UNCHANGED
980 FOR L=First_tc+10 TO First_tc+17
                                               MAPPING THERMOCOUPLES TO
990 Tave(L+2)=Tsub(L)
                                               !CHANNELS--10th-18th CHANNELS
1000 NEXT L
                                                !SHIFTED BY 2 POSITIONS
1010 F=First_tc
1020 FOR L=First_tc+20 TO First_tc+29 !MAPPING THERMOCOUPLES TO
1030 Tave(L+4)=Tsub(L)
                                              !CHANNELS -- 21st - 30th CHANNELS
1040 NEXT L
                                              !SHIFTED BY 4 POSITIONS
1050 \text{ Tave}(F+10) = \text{Tsub}(F+30)
                                         MAPPING THERMOUCOUPLES TO CHANNELS
1060 Tave(F+11)=Tsub(F+31)
                                          !ADDITIONAL ADJUSTMENTS
1070 Tave(F+20)=Tsub(F+34)
1080 Tave(F+21)=Tsub(F+35)
1090 Tave(F+22)=Tsub(F+32)
1100 Tave(F+23)=Tsub(F+33)
1110 Tave(F+34)=Tsub(F+36)
1120 Tave(F+35)=Tsub(F+37)
1130 OUTPUT 709;"SD1"
                                 ! TURN DISPLAY BACK ON
1140 REM Print headings
1150 CLEAR SCREEN
1160 PRINTER IS 26
1170 PRINT TABXY(20,1), "DATASHEET FOR ENTRAINMENT IN HEAT PIPE"
1180 PRINT
1190 Dend$=DATE$(TIMEDATE)
1200 Tend$=TIME$(TIMEDATE)
1210 PRINT "RUN DATE: ";Dend$
1220 PRINT " TIME @ END OF RUN:
                                   ";Tend$
1230 PRINT Bath; " DEGREES C CONSTANT TEMPERATURE BATH"
1240 PRINT "THE MESH NUMBER IS ";Mesh
1250 PRINT
1260 PRINT "THE POWER INPUT IS "; Watts; " WATTS"
1270 PRINT
1280 PRINT "TEMPERATURE READINGS ALONG THE HEAT PIPE WALL"
1290 PRINT "Location from the evaporator end"
1300 PRINT "TC #","in. ",TAB(29);
1310 PRINT "Avg Temp"
1320 REM AVERAGING TEMPERATURE READINGS
1330 Tc wall=First tc+9
1340 FOR L=First tc TO Tc wall
1350 Tc loc=(L-First tc+1)*4
1360 PRINT L, Tc loc, TAB(29);
1370 PRINT Tave(L)
1380 NEXT L
1390 PRINT
1400 PRINT "LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE"
1410 PRINT "** locations are measured from the evaporator end **"
1420 PRINT
1430 PRINT "
              VAPOR TEMPERATURE", TAB(46); "LIQUID TEMPERATURE"
1440 PRINT
1450 PRINT " TC #"," in. "," TEMP. ";TAB(42);" TC #"," in. "," TEMP."
1460 FOR L=Tc wall+1 TO Last tc-4 STEP 2
1470 PRINT L,T loc(L),Tave(L),TAB(42);L+1,T loc(L+1),Tave(L+1)
1480 PRINT
1490 NEXT L
1491 GOTO 1500
1493 PRINT "TC LOC.", "PSIG."
1494 FOR L=First tc+10 TO Last tc-4 STEP 2
1495 PRINT (T loc(L)+T loc(L+1))/2, Pave((L-128)/2)
1496 NEXT L
1497 PRINTER IS 26
1500 CREATE File name$,600
```

```
1510 ASSIGN @Path TO File name$; FORMAT ON
        FOR L=First tc+10 TO Last tc-4
1520
                                                    !OUTPUT TEMPERATURES
            OUTPUT @Path; VAL$(T loc(L)), VAL$(Tave(L))
1530
       NEXT L
1540
1541 GOTO 1590
1550 FOR L=First_tc+10 TO Last_tc-4 STEP 2 !OUTPUT PRESSURE DATA
            Toc = (T \ loc(L) + T \ loc(L+1))/2
1560
            OUTPUT @Path; VAL$(Toc), VAL$(Pave((L-128)/2))
1570
     NEXT L
1580
      OUTPUT @Path;VAL(Tend$)
1590
1600
         OUTPUT @Path; VAL$ (Mesh);
       OUTPUT @Path;VAL$(Watts);
OUTPUT @Path;VAL$(Bath);
1610
1620
       OUTPUT @Path;VAL(Dend$)
ASSIGN @Path TO *
1630
1640
1650 PRINT "END OF PROGRAM"
1660 PRINTER IS 1
1665 OUTPUT 709; "SA" ! BEEP?
1670 NEXT M
1680 END
```

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Power input = 34 Watts

TT 10 30.078 30.424 TT 910 VAROR TEMPERATURES ALONG THE HEAT PIPE

** locations are measured from the evaporator end **

SOPOR TEMPERATURE

LIQUID VEMPERATURI

	100 m 이 것 같은 11월 12일	TEMP. 37.393			144. 1.1874	
an tha An tha	.12.1825	37.16.		n na saya Sa saya	11.4875	
	18,1875	36.684		1 - 12 - 12 1 - 12 - 12	·7.6875	
1 2 2	24.1875	36.166		4 	23.6875	34. MAG
1000	10.1875	35.451		·	27.6875	
$\frac{1}{2} \mathcal{A} \mathcal{C}_{\mathcal{F}}$	Co.1075	32.629		5 4 1	35.6675	
142	42.1875	23.31Y	Advaluation	7-43	41.6875	<u>1</u> 1. (1.2)
and the second second second	48.1875	24.446	Tenterature	* 2 1 200	47.6875	25. oraș
144	54.1875	20.329	cr	147	53.6875	atstr
143	60.1875	17.937	cherming	149	59.4873	18.355
150	66.1875	17.13	CENTER CLARK	151	65.6875	17.437
$ \frac{1}{2} = \frac{2 \int_{-\infty}^{+\infty} e^{i t t} dt}{2 \int_{-\infty}^{+\infty} e^{i t} dt} + \frac{1}{2} \int_{-\infty}^{+\infty} e^{i t t} dt + \frac{1}{2} \int_{-\infty}^{+\infty} e^{i t} dt + \frac{1}{2} \int_{-\infty}^$	72.1875	15.561		123	71,6875	
	27.4875	15.610		e unitar to on unitar	77.1275	1 <u>2</u> <u></u>

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12. 	. i.,	-57			· ;	3.5
		2.5	73			15 a 4
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· · · · ·		•	73		··	3.5
. 121.			72.		··· ;	nar i dar Nar i Nar
	с. 1 г.	9.0	73		· · · •	
						216
5.		5 5	N:S	•	;	
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RUN DATE: TIME O GEER THE PESH THE POWER THE POWER TEMPERATI Location TC # 120 121 122	: 11 Ma @ END OF R EES C CONST. NUMBER TO R INPUT IS JRE READING from the e in. 4 8 12	r 1990 UN: 20:37:21 ANT TERPERATIONS BATH 40 47 WATTS 5 ALONG THE HEAT PIP vaporator end Avg Temp 41.578 42.101	EWALL	41 m			
124 125 126 127 128 128	20 24 28 32 36 40	41.22 41.179 43.63 31.061 37.486 33.488					
LIQUID AN ** locati VAPOP	LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **						
TC # 130	in. 6.1375	TEMP. 41.309	TC # 1∃1	in. 5.6873	TEMP. 41.923		
132	12.1875	41.19	133	11.6875	41,337		
134	18.1875	40.773	135	17.6875	41.447		
:36	24.1875	40.223	137	23.6875	41.33		
138	30.1875	39.559	139	29.6875	40.207		
140	36.1875	35.975	141	35 .6 878	37.385		
142							
	42.1875	30.893	143	41.6875	31.004		
144	42.1875 48.1875	30.390 25.815	143 145	41.6875 47.6875	31.974 27.336		
144 146	42.1875 48.1875 54.1875	30.393 25.815 21.087	143 145 147	41.6875 47.6875 53.6875	31.994 27.338 22.103		
144 146 148	42.1875 48.1875 54.1875 60.1875	30.590 25.815 21.087 18.471	143 145 147 149	41.6875 47.6875 53.6875 59.6875	31.994 27.336 22.103 19.16		
144 146 148 150	42.1875 48.1875 54.1875 60.1875 66.1875	30.590 25.815 21.087 18.471 16.68	143 145 147 149 151	41.6875 47.6875 53.6875 59.6875 65.6875	31.994 27.336 22.105 19.16 17.178		
144 146 148 150 152	42.1875 48.1875 54.1875 60.1875 66.1875 72.1875	30.593 25.815 21.087 18.471 16.68 15.442	143 145 147 149 151 153	41.6875 47.6875 53.6875 59.6875 65.6875 71.6875	31.994 27.336 22.105 19.16 17.176 15.687		

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i ta ha ha ha n	H 16116 a
$ \frac{G_{\rm eff}}{G_{\rm eff}} = \frac{G_{\rm eff}}{G_{\rm eff}} \sum_{i=1}^{N-1} \frac{G_{\rm eff}}{G_{\rm eff}} \label{eff} $	-13.4
11.03755	-13,4
17.7373	-13.5
$\sum_{i=1}^{n-1}\sum_{j=1}^{n-1}\left(\frac{1}{2}\sum_{i=1}^{n-1}\left(\frac{1}{2}\sum_{i=1}^{n-1}\right)\right)$	-13.3
29 . 연건가영	-1.3.4
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700 - #0	in.	in the second	
1.12.5].	4.6 7713	
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	• 2.5 5.44	47.142	
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123	24	A60576	
1.2.6		49.637	
127			
1.72(6)		42.623	
4. j2 Q	4.0	37.784	

SIGUID AND WHEER TEMPERATURES ALONG THE HEAT PIPE Kallections are measured from the evaporator end **

VAPER TEMPERATURE -

LIQUID TEMPERATURE .

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TC # 130	in. 5.1875	TEMP. 46.41	TC # 131	in. 5.6875	TEMP. 47.226
132	12.1075	4/ _{5 a} 447	133	11.6875	4 7.1234
134	18.1875	46.037	135	17.6875	46,287
135	24.1875	45.537	137	23.6875	47. (S)
1.363	30.1875	44.943	t 39	29.6875	46.1St
1.3.0	36.1975	40.734	141	35.6875	al Que (a Solar
142	42.1875	34.344	1.4.3	41.6875	36.087
144	48.1875	28:736	1.45	47.6875	30.58°
12.45	54.1875	2275 a. 1	147	53.6875	24.384
148	60.1875	19.864	149	59.6875	21.09
150	66.1875	19.356	151	65.6875	18.799
152	72.1875	16.417	153	71.6875	17.033
134	77.6875	16.301	155	77.1875	16.30

C L.	ΩС.	PSIG.
8.9	375	-13.1
1.2.7	9076	
$1 \ge \pi$	9279	-1.3 1
23.	9278	-12.9
29e	9375	4 <u></u>
25.	92.72	-12.2
$\{\cdot,\cdot\}_n$	$Q [[S]] \neq [s]$	$= - \frac{1}{2} \left(\frac{T}{T} \right)_{\rm s} = 0$
7 a	$\Psi M/M$	
с. с [.] т	9 Y - 5	-1 I.J.
	5. 1 · · ·	
,		

RUN DATE: 11 Mar 1990 TIME & END OF RUN: 23:39:15 O DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 67 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp

1.02 11		marg resulp
120	~] .	46.238
$1 \ge 1$	8	44.974
1.22	12	45.962
1.23	16	47.508
124	20	45.133
1.255	() A.	45.192
126	28	49.273
126 127	28 32	49.273 29.617
126 127 128	28 32 34	49.273 29.617 42.8
126 127 128 129	28 32 36 40	49.273 29.617 42.8 40.354

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAFOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 41.118	TC # 131	in. 5.6875	TEMP. 41.743
132	12.1875	41.022	133	11.6875	41.624
134	18.1875	41.177	135	17.6875	41.90.5
136	24,1875	41.04	137	23.6875	40.709
138	30.1875	40.915	139	29.6875	41.939
140	36.1875	40.998	141	35.0075	41.463
142	42.1875	41.005	143	41.6875	39.737
144	48.1875	40.945	145	47.6875	37.46
146	54.1875	31.527	147	53,6875	29.592
148	60.1875	22.042	149	59.6875	22.162
150	66.1875	19.489	151	65.6875	19.65
152	72.1875	18.086	1.555	71.6875	18.22
1 (5) 4].	77.6875	17.503	155	77.1975	17.606

fe Loc.	PSIG.
5.9375	<u>1</u> × <u>1</u>
11.9378	
17.9325	
25.9.778	- 13 . O
29.0375	-13.0
1.5.9175	····] 2]
라이 이 건물	-13.9
47,9775	-14
5 mar (1916) 19	- L - L
$\sigma = 10^{-2}$ $\sigma = 10^{-2}$ $\sigma = -10^{-2}$ $\sigma = -10^{-2}$	

EUN DATE: 11 Mar 1990 TIME @ END OF RUN: 23:47:22 O DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 67 MATTE

TEMPERATURE READINGS ALONG THE HEAT PIFE WALL Location from the evaporator end

TC #	i. n.,	Avg Temp
120	<i>4</i>],	42.972
121	8	43,517
t22	1.2	43.652
123	1.6	45.168
124	20	43.528
125	(24)	44.373
126	28	49,145
127	13 M2	29.623
128	36	43.859
129	47. (_)	41.986

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130 .	in. 6.1875	TEMP. 42.806	TC # 131	in. 5.6875	TEMP. 42.901
132	12.1875	42.675	1.335	11.6875	42.2
134	18.1875	42.723	135	17.6875	42.365
136	24.1875	42.658	137	23.6875	43.496
138	30.1875	42.6	139	29.6875	43.706
140	36.1875	42.77	141	35.6875	43.404
142	42.1875	42.806	1.4.2	41.6875	42,462
144	48.1875	36.609	145	47.6875	37.207
146	54.1875	25.645	147	53,4875	26.095
148	60 . 1875	20.484	149	59.6875	20.374
150	66.1875	19.245	151	65.6875	18.316
14722	72.1875	18.503	153	71.6875	17.253
154	77.6875	13.616	1.55	77.1875	17.945

TC LCC.	PSI6.
8.9375	-13.9
11.9375	
17.9375	-13.9
23.0375	-1.25 , 7
20.9375	-13.7
1969 - 963 7463 A	
41.9375	-13.8
47.0379	<u>1</u> 2.
63.9575	- 1 4
$\sup_{n \in \mathbb{N}} \{ \sum_{n \in \mathbb{N}} \{ e_n \} \in \mathbb{N} \} \in \mathbb{N} \}$	-17.5
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신출면 다시 관정 병습니다. 법단

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i a tr	i sa ta Tt		1.1.7	51.6675	
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130	ad.1875	11.423	151	55.6875	C.S.
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ne Lee.	PSIG.
5.9075	777.1
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125		34.919
1.26	10	JA. 387
1.27		33.567
128	36	32.455
129	41()	27.297

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the preparator and f*

VAPOR TEMPERATURE

LIQUID TEMPERATURE

10 # 130	69. 6.1875	18MP. 35.095	7752 - 1 1-5-1	16. 5.5825	17.195. 755. 195
1 52	13.1875	24.155		11.0075	
1 34	18.1875	34.802	 March State 	17.5875	·
135	04.18.15	[14]」第443			1. J.
L 38	10.U8/11	201 - 19 7 -	1 - 2	17. oli (* 1	
1.40	16. Tere			 All Annual Annual	
	12.1375	25.200		41.0313	
	43. 1875	10.277	1. 212	4 10	
145	54.ta75	13.040	1 47	53.a877	li tu
148	60.1875	11.013	(1-2	39.38TT	
l ÉO	36.107E	10.107	2 11.27 A	55.687E	3 . 2 - 1
152	72.1875	11.143		71.6875	41.14
154	77.6875	11.022	152	77.1975	a. 26. 5

TC LCC. PSIG. 5.9375 785.4 11.9375 601.4 17.9375 487 475.7 23.9375 29.9375 457.7 35.9375 ~ 4 년 5 역 484.8 41.9375 47.9375 195.8 53.9375 274.3 271.9 59.9375 65.9375 289

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LIQUID AND VAPOR (EMPERATURES ALONG THE HEAT PIPE Af instituous are measured from the evaporator and are

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CLARINE ESTREMAN

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1 4 4	-8.1872	18. 775		+7.687€	
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142	er, 1915	10.964	1 x4 💬	59.6875	(4., - ¹
150	66.1875	10,131	151	5 5. 6678	- - -
152	72.1875	11.29	183	71.6875	9. ₀₅ 9
154	77.6875	11.479	155	77.1875	10. HE

END OF PROGRAM

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RUN DATE: 14 Mar 1990 TIME @ END OF RUN: 02:42:-3 O CEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 50 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end

11. 44	1. FT	Avg lemp
120	4	41.363
121	8	41.743
t 22	12	41.827
123	16	42.884
1/24	20	41.523
1.255	24	41.963
126	28	46.589
127	32	40.605
128	36	40.175
129	40	37.822

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

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TC # 130	in. 6.1875	TEMP. 39.529	TC # 131	in. 5.6875	TEMP. 40.802
132	12.1875	39.451	133	11.6875	40.133
134	18.1875	39.481	1.365	17,6875	40,337
136	24.1875	39.571	137	23.6875	41.361
138	30.1875	39.547	139	29.6875	39.87
140	36.1875	39.403	141	35.6875	38.448
142	42.1875	39.379	143	41.6875	34.96
344	48.1875	24.574	145	47.6875	24.863
146	54.1875	15.093	147	53.6875	15.566
148	60 . 1875	11.892	149	59.6875	12.243
150	66. 18 75	9.993	$1 \le 1$	65.6875	10.535
172	72.1875	11.703	153	71.6875	10.221
$\frac{1}{2} = \frac{1}{2} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{i=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{i=1}^{n$	77.6875	11.611	1,6315	77.1875	10. 문방식

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CUICO AMO VAPOR TEMPERATURES ALONG THE HEAT PIPE >> locations are measured from the evaporator and **

VAPOR TEMPERATURE

LIGUID TEMPERATURE

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TC # 130	ln. 6.1875	TEMP. 38.429	TC # 131	in. 5.6875	1829 40.15
1.322	12.1875	38.241	133	11.6875	40.234
$\frac{\partial_{i}}{\partial t} = \frac{\partial_{i} \mathcal{L}}{\partial t} + \frac{\partial_{i} \mathcal{L}}{\partial t} + \frac{\partial_{i} \mathcal{L}}{\partial t}$	18.1873	38.507	135	17.6875	39.377
186	24.1875	38.387	137	23.6875	41.135
158	30,1875	38.339	139	29.4875	39 E15
140	36.1875	38.453	141	35.6875	39.537
142	42.1875	38.567	143	41.6875	38.662
144	48.1875	38.411	145	47.6875	33.943
146	54.1875	38.152	147	53.6875	30.603
148	60.1875	20.139	149	59.6875	20,474
150	66.1875	13.725	131	65.6875	13.357
1.1972	72.1875	12.683	1,533	71.6875	1). se2
154	77.3875	12.436	1. (1) (1)	77.1875	11.040

END OF PROGRAM

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NUN DATA: 14 Mag 1990

- TIME @ END OF RUN: POSTERIES O DEBREES C CUMSIANT IN SERVICE RATE
- THE MEET NURGER 10 1

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TEMPERATURE PRADIMER ALDER AND HER HER PIPE AND Locations from the exeponation end

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1 (20)	х4.	44.535
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122		林府,亦墨海
1.223	14	명수, 2:34
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1.1.1.1.1	3.4	
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NUMBER VAROR TEMPERATURES ALLOS THE HEAT RIPE

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er lonsiders are measured here this evaporator end **

LIQUID TEMPERATURE

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	1999. 1997 -	515.07 37.1229	TC # 131	in. 5.6875	TENG. 39.337
	10.1875	37.187	133	11.6875	38.884
134	18.1875	37.44	135	17.6875	38.8
136	24.1875	37.447	137	23.6875	39.253
t38	30.1875	37.513	139	29.6875 •	39.379
140	36.1875	37.38	141	35.6875	38.572
142	42.1875	37.29	143	41.6875	37.863
144	48.1875	37.079	145	47, 6875	36.733
146	54.1875	36.945	147	53.6875	35.225
148	60.1875	36.607	149	59.6875	32.792
150	66.1875	37.313	151	65.6875	30.418
152	72.1875	34.323	153	71.6875	25.484
184	77.6375	17.736	155	77.1875	17.031

END OF PROSRAM

RUN DATE: 14 Mar 1990 TINE @ END OF RUN: 05:14:07 O DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 475 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end

IL W	1n.	Avg Temp
1.20	4	44.313
1 22 1	8	45.197
1.2.2	1.2	45,715
123	16	50.136
1.22.4	20	44.03
1 (255	24	45.121
126	28	53.467
127	125.72	43.599
128	36	43.644
129	40	45.71

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAFER TEMPERATURE

LIQUID TEMPERATURE

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TC # 130	in. 6.1875	TEMP. 37.93	TC # 131	in. 5.6875	TEMP. 39.504
132	12.1875	37.791	133	11.6875	39.04
134	18.1875	38.117	135	17.6875	40,204
136	24.1875	37.894	137	23.6875	38.787
138	30.1875	37.918	139	29.6875	40.449
140	36.1875	37.966	141	35.6875	39.186
142	42.1875	37.912	143	41.6875	38.02
144	48.1875	38.014	145	47.6875	37.539
146	54.1875	37.707	147	53.6875	36.463
148	60.1875	37.707	149	59.6875	36.396
$\frac{1}{2} \frac{1}{2} \frac{1}$	66.1875	38.146	151	65.6375	36.021
182	72.1875	37.719		71.6875	
1,654	77.6875	32.312	1.65	77.1875	

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SUN DATE: 14 Mar 1990 TIME @ END OF RUN: 07:12:21 O CESREES C CONSTANT TENPERATURE BATH THE MESH NUMBER IS 40

THE FOWER INPUT IS 177 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end

TC #	i. m	Avg Temp
120	.д.	53.524
121	8	54.016
122	1/2	54.743
123	16	59,429
124	20	52.595
1.255	24	53.208
126	12 (B	62.461
127	32	33.997
128	36	52.065
129	4 ()	46.971

LIGUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

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TC # 130	in. 6.1875	TEMP. 45.903	TC # 131	in. 5.6875	TEMP, 47.501
132	12.1875	46.055	133	11.6875	47.731
134	t8.1875	46.014	135	17.6875	47.OÉV
136	24.1875	45.873	137	23.6875	150 J S 4 7
138	30.1875	45.996	139	29.6875	47. <u>5</u> 77
140	36.1875	45.777	141	35.6875	46.588
142	42.1875	45.92	143	41.6875	46.126
144	48.1875	45.914	145	47.6875	45.456
146	54.1875	45.738	147	53.6875	44.714
148	60.1875	45.556	149	59.6875	44.e7
150	66.1375	46.333	151	65.6875	42.419
1.6522	72.1875	44.513	1.52	71.6875	38.442
154	77.6875	30.036		77.1875	29. °S

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 5.9375
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RUN DATE: 14 Mar 1990 TIME @ END OF RUN: 08:47:22 O DEGREES C CONSTANT TEMPERATURE SATH THE MESH NUMBER IS 40

THE POWER INPUT IS 177 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end

TC #	i. m "	Avg Temp
120	4.	54.677
1.2.1	8	56.164
1.2.2	1.22	56.346
123	16	60.907
124	20	54.163
1.255	24	55.761
126	28	64.905
127	32	- 34.938
128	36	54.718
1.2.9	40	55.619

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 49.839	TC # 131	in. 5.6875	TEMP. 50.8
1.352	12.1875	49.572	1.22	11.6875	49.931
134	18.1875	49.572	1.1555	17.6875	30.239
t36	24,1875	49.543	137	23.6875	45.666
138	30.1875	49.688	139	29.6875	51.031
140	36.1875	49.792	141	35.6875	50.534
142	42.1875	49.943	143	41.6875	50.29
44	48.1875	49.624	145	47.6875	49.247
146	54.1875	49.438	147	53.6875	48.636
148	60.1875	49.287	149	59.6875	47.J¢
150	66.1875	49.1946	151	63.6875	47.000
4 60 62 1	72.1875	49.09	153	71.4875	45 (AN
	77.6875	40.769	1.000	77.1875	

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RUN VATE: 14 Mar 1990 TIME @ END OF RUN: 09:42:01 O DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 250 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALLLocation from the evaporator endTC #in.Avg Temp120460.375121862.4321221261.785

di ada di	1.6	68.83
124	20	60,913
1 235	24	62.671
126	28	72.631
 127	32	59.027
128	36	60.544
129	40	62.716

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 56.301	TC # 131	in. 5.6875	TEMP. 57.271
132	12.1875	56.37	1.000	11.6875	57.086
134	18.1375	56.432	135	17.6875	57.056
136	24.1875	56.37	137	23.6875	37.455
138	30.1875	56.574	139	29.6875	57.498
140	36.1875	56.25	141	35.6875	56.817
142	42.1875	56.381	143	41.6875	56.505
144	48.1875	56.284	145	47.6875	55.396
146	54.1875	56.143	147	53.6875	55.026
148	60.1875	œui "QA.≐	149	59.6875	20. C40
150	66.1875	55.909		65.687E	
152	72.1875	53.001		71.6875	titu Proj
164	77.6875	44.60%	1 4.57 3.52 1	77.1875	(1.64) 1.1 -

RND OF PROPEST

C. Data sheets for Trial 2

1541 GOTO 1590 1550 FOR L=First tc+10 TO Last tc-4 STEP 2 !OUTPUT PRESSURE DATA 1560 $Toc=(T_loc(L)+T_loc(L+1))/2$ OUTPUT @Path; VAL\$ (Toc), VAL\$ (Pave((L-128)/2)) 1570 OUTPUT @Path;VAL(Tend\$) OUTPUT @Path;VAL\$(Mesh); OUTPUT @Path;VAL\$(Watts); OUTPUT @Path;VAL\$(Bath); OUTPUT @Path;VAL\$(Bath); 1580 NEXT L 1590 1600 1610 1620 1630 1640 ASSIGN @Path TO * 1650 PRINT "END OF PROGRAM" 1660 PRINTER IS 1 1665 OUTPUT 709; "SA" ! BEEP? NEXT M 1670 1680 END DATASHEET FOR ENTRAINMENT IN HEAT PIPE 24 Mar 1990 RUN DATE: TIME @ END OF RUN: 16:34:53 -1 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40 THE POWER INPUT IS 25 WATTS TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp 4 120 30.977 121 8 30.995 12 16 122 30.786 123 30.78
 124
 20

 125
 24

 126
 28

 127
 32
 29.639 28.208 28.704 25.623 36 128 24.156 40 129 22.253 LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE			LIQU	LIQUID TEMPERATURE			
TC # 130	in. 6.1875	TEMP . 30.774		TC # 131	in. 5.6875	TEMP. 30.952	
132	12.1875	30.256		133	11.6875	30.521	
134	18.1875	29.515		135	17.6875	29.947	
136	24.1875	28.14		137	23.6875	29.004	
138	30.1875	25.954		139	29.6875	27.444	
140	36.1875	23.565		141	35.6875	24.89	
142	42.1875	20.429		143	41.6875	21.867	
144	48.1875	16.397		145	47.6875	19.008	

END OF PROGRAM DATASHEET FOR ENTRAINMENT IN HEAT PIPE 24 Mar 1990 RUN DATE: TIME @ END OF RUN: 16:46:35 -1 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40 THE POWER INPUT IS 25 WATTS TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp 120 4 31.541 121 8 31.535 12 31.319 122 123 16 32.121 20 30.309 124 24 29.952 125 126 28 31.006 127 32 28.281 36 26.521 128

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

24.882

VAPOR TEMPERATURE

40

129

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 31.406	TC # 131	in. 5.6875	TEMP. 31.602
132	12.1875	30.938	133	11.6875	31.154
134	18.1875	30.39	135	17.6875	30.618
136	24.1875	29.52	137	23.6875	30.67
138	30.1875	28.046	139	29.6875	28.387
140	36.1875	25.202	141	35.6875	25.559
142	42.1875	20.884	143	41.6875	21.107
144	48.1875	14.778	145	47.6875	14.681
146	54.1875	8.275	147	53.6875	7.221
148	60.1875	7.406	149	59.6875	6.171
150	66.1875	9.914	151	65.6875	6.151
152	72.1875	7.847	153	71.6875	6.832
154	77.6875	8.866	155	77.1875	7.761

TPRESSURF	PROC	2	256	2772	$2/1 = 0.00 \pm -8.9$	10.47
IINDUT	ACCTT	2	250	2772	24 - 000 - 00	22.06
	ASCII	Z	256	2774	2-Mar-90	23:00
NPRESSURE	PROG	4	256	2776	6-Nov-89	11:06
DATAM	HP-UX	768	1	2780	14-Feb-90	10:23
NT T VAX	PROG	20	256	2784	24-Jan-90	18:05
NT T TIMED	PROG	29	256	2804	14-Mar-90	12:29
LPRESSURE	PROG	4	256	2840	6-Nov-89	18.22
	PROC	18	256	2844	5 - Mar - 90	12.21
	PROG	10	250	2044	16 Nor 90	15.10
WUN	PROG	10	256	2003	10-NOV-09	15.10
WUNI	PROG	12	256	2873	11-Dec-89	10:00
WUNS	PROG	11	256	2885	30-Nov-89	10:01
AJT1	PROG	7	256	2896	12-Mar-90	23:58
TRIAL	PROG	23	256	2903	30-Jan-90	10:24
DATAN	HP-UX	768	1	2926	14-Feb-90	10:28
DATAO	HP-UX	768	1	2929	14-Feb-90	10:39
AJT5	PROG	17	256	2935	22-Mar-90	2:15
VOLT	PROG	2	256	3039	16-Jan-90	14:21
DATAD	HP-UX	768	1	3047	14-Feb-90	11.58
EVTDA CODV	PPOC	11	256	3203	26 - 100 - 90	17.42
LAIRA_COFI		769	200	2205	20-Jan-90	10.04
DATAL	HP-UX	/68	1	3214	14-Feb-90	12:04
DATDA	HP-UX	/68	1	3217	1/-Feb-90	10:13
DATBA	PROG	12	256	3220	11-Mar-90	12:29
TRANSIENT	HP-UX	2560	1	3232	10-Mar-90	19:50
DATBC	HP-UX	768	1	3245	14-Feb-90	15:31
DATBD	HP-UX	768	1	3248	14-Feb-90	15:37
DATBE	HP-UX	768	1	3251	14-Feb-90	15:42
DATCA	HP-UX	768	1	3254	14-Feb-90	15:58
DATCB	HP-IIX	768	1	3257	14-Feb-90	16.02
DATCC		768	1	3260	$14 - F_{0} = 90$	16.06
DATCE		760	1	3260	14-Feb-90	16.10
DATCE	HP-UX	700	1	5265	14-Feb-90	10:15
DATCF	HP-UX	/68	1	3266	14-Feb-90	16:18
DATCG	HP-UX	768	1	3269	14-Feb-90	18:02
DATCH	HP-UX	768	1	3272	14-Feb-90	18:06
DATDB	HP-UX	768	1	3275	17-Feb-90	10:18
DATDC	HP-UX	768	1	3278	17-Feb-90	10:23
DATDD	HP-UX	768	1	3281	17-Feb-90	10:27
DATDE	HP-UX	768	1	3284	17-Feb-90	10:33
DATEA	HP-IIX	768	1	3287	17 - Feb - 90	15.42
DATER		760	1	3207	17 Feb 00	15.42
DATEC		700	1	3290	17-Feb-90	15.50
DAIFC	HP-UX	/68	1	3293	17-Feb-90	15:50
DATFD	HP-UX	/68	1	3296	1/-Feb-90	15:54
NT_TRANS	DATASHEET F	OR ENTRAI	NMENT IN H	HEAT PIPE		
RUN DATE:	24 Mar 1	990				
TIME @	END OF RUN:	14:06:	33			
-5 DEGREES	C CONSTANT	TEMPERAT	URE BATH			
THE MESH NU	IMBER IS 40					
THE DOLLED T		2 174 7770				
THE POWER I	NPUI 15 10	5 WAIIS				
TEMPERATURE	E READINGS A	LONG THE	HEAT PIPE	WALL		
Location fr	om the evap	orator en	d			
TC # i	.n.	Avg	Temp			
120	4	56	.672			
121	8	56	.168			
122	12	56	622			
123	16	50	534			
12/	20	50	216			
105	20	57	.210			
125	24	55	. 335			
126	28	61	.492			

127	32	53.5 03
128	36	53.147
129	40	53.64

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 52.952	TC # 131	in. 5.6875	TEMP. 53.675
132	12.1875	52.958	133	11.6875	54.212
134	18.1875	53.01	135	17.6875	53.52
136	24.1875	52.969	137	23.6875	55.531
138	30.1875	52.935	139	29.6875	53.119
140	36.1875	52.849	141	35.6875	51.49
142	42.1875	52.929	143	41.6875	48.413
144	48.1875	39.239	145	47.6875	35.909
146	54.1875	16.604	147	53.6875	13.483
148	60.1875	9.101	149	59.6875	7.971
150	66.1875	9.531	151	65.6875	6.031
152	72.1875	6.824	153	71.6875	5.978
154	77.6875	7.944	155	77.1875	6.586

END OF PROGRAM

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 14:09:23 -5 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp 4 120 54.441 121 8 55.702 12 16 20 122 55.611 123 57.178 124 56.293 125 24 56.088 28 126 61.811 127 32 54.156 128 36 53.261 129 40 54.081

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 17:42:20 -5 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 52 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp 4 120 41.02 120 4 121 8 122 12 123 16 124 20 125 24 126 28 127 32 41.491 41.182 42.086 40.967 40.375 41.646 32 36.148 127 33.814 128 36 40 129 31.5

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 41.044	TC # 131	in. 5.6875	TEMP. 41.313
132	12.1875	40.752	133	11.6875	41.283
134	18.1875	40.376	135	17.6875	40.919
136	24.1875	39.4	137	23.6875	40.789
138	30.1875	37.64	139	29.6875	38.434
140	36.1875	34.15	141	35.6875	34.308
142	42.1875	29.194	143	41.6875	28.613
144	48.1875	21.128	145	47.6875	22.159
146	54.1875	12.227	147	53.6875	14.314
148	60.1875	9.557	149	59.6875	10.656
150	66.1875	10.335	151	65.6875	8.973
152	72.1875	6.782	153	71.6875	6.9
154	77.6875	6.901	155	77.1875	6.505

END OF PROGRAM

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 17:45:23

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RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 18:41:27 -10 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

TEMPERATU	JRE REA	DINGS	ALONG	THE	HEAT	PIPE	WALL
Location	from t	the eva	aporato	or er	nd		
TC #	in.			Avg	g Temp	0	
120	4			51	L.7		
121	8			51	L.937		
122	12			52	2.932		
123	16			55	5.499		
124	20			52	2.748		
125	24			53	3.052		
126	28			58	3.536		
127	32			50).32		
128	36			50).36		
129	40			52	2.138		

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 46.077	TC # 131	in. 5.6875	TEMP. 46.815
132	12.1875	45.989	133	11.6875	47.084
134	18.1875	46.035	135	17.6875	47.002
136	24.1875	46.112	137	23.6875	47.488
138	30.1875	46.235	139	29.6875	46.599
140	36.1875	45.859	141	35.6875	46.106
142	42.1875	45.748	143	41.6875	45.707
144	48.1875	45.56	145	47.6875	41.41
146	54.1875	45.29	147	53.6875	33.227
148	60.1875	26.512	149	59.6875	19.811
150	66.1875	12.42	151	65.6875	9.281
152	72.1875	4.304	153	71.6875	4.211
154	77.6875	4.643	155	77.1875	3.68

END OF PROGRAM

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 18:44:15 -15 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp

120	4	61.497
121	8	62.846
122	12	62.479
123	16	63.589
124	20	62.846
125	24	61.335
126	28	65.168
127	32	54.612
128	36	50.681
129	40	46.865

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR	TEMPERATURE LIQUID TEMPER			ID TEMPERAT	ATURE	
TC # 130	in. 6.1875	TEMP. 62.128	TC # 131	in. 5.6875	TEMP. 62.74	
132	12.1875	62.128	133	11.6875	62.539	
134	18.1875	61.86	135	17.6875	62.752	
136	24.1875	60.445	137	23.6875	59.961	
138	30.1875	56.695	139	29.6875	57.596	
140	36.1875	49.133	141	35.6875	49.749	
142	42.1875	36.863	143	41.6875	38.628	
144	48.1875	20.971	145	47.6875	24.077	
146	54.1875	4.831	147	53.6875	9.098	
148	60.1875	. 508	149	59.6875	1.967	
150	66.1875	-2.479	151	65.6875	-1.848	
152	72.1875	702	153	71.6875	-2.807	
154	77.6875	.38	155	77.1875	-2.097	

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 16:43:25 -15 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

END OF PROGRAM DATASHEET FOR ENTRAINMENT IN HEAT PIPE 24 Mar 1990 RUN DATE: - S TIME @ END OF RUN: 20:40:22 -20 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40 THE POWER INPUT IS 103 WATTS TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp 120 4 62.032 62.912 8 121 62.572 122 12 123 16 64.476 124 20 63.334

125	24	63.056
126	28	67.514
127	32	57.555
128	36	53.507
129	40	49.214

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE LIQUID TEMPERATURE TC # TEMP. TC # TEMP. in. in. 62.305 5.6875 63.14 130 6.1875 131 64.008 12.1875 62.478 11.6875 132 133 62.578 17.6875 63.462 134 18.1875 135 24.1875 61.876 23.6875 63.686 136 137 30.1875 58.904 139 29.6875 60.17 138 140 36.1875 51.2 141 35.6875 51.471 142 42.1875 38.338 41.6875 38.741 143 23.492 144 48.1875 21.755 145 47.6875 146 54.1875 4.996 147 53.6875 5.413 148 60.1875 1.933 59.6875 -.535 149 65.6875 -1.416 150 66.1875 .514 151 152 72.1875 1.311 153 71.6875 -.824 154 77.6875 2.547 .42 155 77.1875

END OF PROGRAM DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 23:06:59 -20 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end in. TC # Avg Temp 4 61.323 120 61.809 61.284 62.332 59.568 57.295 60.377 49.875 47.144 40 129 44.955

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 61.557	TC # 131	in. 5.6875	TEMP. 62.154
132	12.1875	60.831	133	11.6875	64.02
134	18.1875	59.169	135	17.6875	59.546
136	24.1875	56.359	137	23.6875	58.419
138	30.1875	52.387	139	29.6875	52.427
140	36.1875	45.955	141	35.6875	46.183
142	42.1875	36.868	143	41.6875	38.122
144	48.1875	24.616	145	47.6875	26.524
146	54.1875	8.797	147	53.6875	10.322
148	60.1875	3.853	149	59.6875	5.121
150	66.1875	4.206	151	65.6875	2.46
152	72.1875	. 372	153	71.6875	.526
154	77.6875	1.384	155	77.1875	.164

NEW20 103RA

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 23:17:48

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 22:46:57 -20 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 103 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL
Location from the evaporator endTC #in.Avg Temp120461.231121862.1741221261.2481231662.2411242060.2851252458.5131262862.1411273251.78128364044.909

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 61.533	TC # 131	in. 5.6875	TEMP. 62.102
132	12.1875	60.8	133	11.6875	62.454
134	18.1875	59.915	135	17.6875	60.689
136	24.1875	58.237	137	23.6875	59.504
138	30.1875	55.612	139	29.6875	56.039
140	36.1875	49.6	141	35.6875	49.403
142	42.1875	38.799	143	41.6875	40.236
144	48.1875	26.396	145	47.6875	28.709
146	54.1875	11.189	147	53.6875	15.658
148	60.1875	5.225	149	59.6875	9.178
150	66.1875	4.682	151	65.6875	6.534
152	72.1875	1.348	153	71.6875	3.031
154	77.6875	2.251	155	77.1875	.765

END OF PROGRAM

DATASHEET FOR ENTRAINMENT IN HEAT PIPE

RUN DATE: 24 Mar 1990 TIME @ END OF RUN: 22:49:45

RUN DATE: 25 Mar 1990 TIME @ END OF RUN: 00:54:14 -20 DEGREES C CONSTANT TEMPERATURE BATH THE MESH NUMBER IS 40

THE POWER INPUT IS 156 WATTS

TEMPERATURE READINGS ALONG THE HEAT PIPE WALL Location from the evaporator end TC # in. Avg Temp TC # in. 120 4 121 8 122 12 123 16 124 20 125 24 126 28 127 32 128 36 129 40 74.286 74.073 74.751 77.98 75.423 75.395 82.868 70.316 69.961

LIQUID AND VAPOR TEMPERATURES ALONG THE HEAT PIPE ** locations are measured from the evaporator end **

71.763

VAPOR TEMPERATURE

LIQUID TEMPERATURE

TC # 130	in. 6.1875	TEMP. 71.039	TC # 131	in. 5.6875	TEMP. 71.142
132	12.1875	71.05	133	11.6875	72.915
134	18.1875	71.056	135	17.6875	71.239
136	24.1875	71.083	137	23.6875	71.596
138	30.1875	71.158	139	29.6875	71.121
140	36.1875	70.996	141	35.6875	67.517
142	42.1875	71.001	143	41.6875	61.103
144	48.1875	63.72	145	47.6875	40.048
146	54.1875	23.085	147	53.6875	12.793
148	60.1875	5.145	149	59.6875	-1.169
150	66.1875	.391	151	65.6875	-1.58
152	72.1875	1.255	153	71.6875	-1.776
154	77.6875	927	155	77.1875	-2.532

:CS80, 1500, 2 VOLUME LABEL: HPW C

FILE	NAME	PRO	TYPE	REC/FILE	BYTE/REC	ADDRESS	DATE	TIME

	A	В	С	D	E	F
1	Properties of	of water				
2						
3	Temperature	Density, liq	Density, Vap	h, fg	Surface Tension	
4	°K	kg/m^3	kg/m^3	J/kg	N/m^2	
5	285	1000	0.010	2473000	0.0743	
6	295	1002	0.014	2461000	0.0737	
7	310	1007	0.044	2414000	0.0700	
8	325	1013	0.090	2378000	0.0675	
9	360	1034	0.378	2291000	0.0614	
10						
11						
12						
13	Height(m)	0.008128				
14	Width(m)	0.0267				
15	A, vapor(m)	0.0002				
16	r, wh	0.0013				
17	g	9.8100				
18	A *	2.2000				
19	l, char. leng	0.0054				
20	D, wire	0.0001				
21	D, pipe	0.0062				
22						
23	Calculated h	neat flux at	entrainment	limit		
24						
25	Temperature	Cotter's	Kemme	Prenger	Rice, Fulford	
26						
27	285	283.805	440.549	619.531	802.723	
28	295	335.911	522.527	614.644	950.100	
29	310	559.860	881.361	589.041	1583.523	
30	325	779.797	1238.951	571.496	2205.598	
31	360	1465.182	2389.695	530.535	4144.161	

D. Microsoft Excel spreadsheet for the calculation of theoretical heat flux at entrainment



EXPERIMENTAL HEAT FLUX AT ENTRAINMENT LIMIT



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operating temperature (Expected results)

Q, Heat Flux, (W)





Vapor temperature profile for varying power input and condenser

temperature

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temperature after (C°) .əəs 30