THE ROLE OF DOCOSAHEXAENOIC ACID IN MEDIATING MITOCHONDRIAL MEMBRANE LIPID OXIDATION AND APOPTOSIS IN COLONOCYTES

A Thesis

by

YEE VOON NG

Submitted to the Office of Graduate Studies of Texas A&M University in partial fulfillment of the requirements for the degree of

MASTER OF SCIENCE

August 2004

Major Subject: Nutrition

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ABSTRACT

The Role of Docosahexaenoic Acid in Mediating Mitochondrial Membrane Lipid Oxidation and Apoptosis in Colonocytes. (August 2004)

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Colon cancer is the second leading cause of cancer death in the United States. Epidemiological data indicate that the consumption of dietary fiber and fish/marine products favorably modulate colon tumorigenesis. Docosahexaenoic acid (DHA, 22:6n-3) from fish oil, and butyrate, a fiber fermentation product generated in colon, protect against colon tumorigenesis in part by inducing apoptosis. We have shown that DHA is incorporated into mitochondrial membrane phospholipids, which enhances oxidative stress and mitochondrial membrane potential (MP) dissipation. To elucidate the subcellular origin of oxidation induced by DHA and butyrate exposure, young adult mouse colonocytes (YAMC) were treated with 0-200 µM DHA, linoleic acid (LA, 18:2n-6) or no fatty acid (control) for 72 h with or without 5 mM butyrate for the final 6-24 h. Real time analysis of cellular membrane lipid oxidation, as indicated by oxidation of a lipophilic vital dye, mitochondrial permeability transition (MPT), as characterized by MP dissipation, and cytosolic ROS production, as depicted by hydrophilic ROS reactive fluorophore accumulation, were measured by living cell fluorescence microscopy. After 24 h of butyrate treatment, DHA primed cells showed a 29% increase in lipid oxidation (p<0.01), compared to no butyrate treatment, which could be blocked by a mitochondria targeted antioxidant, MitoQ (p <0.05), whereas LA treatment did not show an effect. In the absence of butyrate, DHA treatment, compared to LA, increased resting MP by 14% (p <0.01). In addition,

butyrate-induced MP dissipation was greater (20%) in DHA primed cells as compared to LA (10%). This effect was blocked by pre-incubation with MPT inhibitors, cyclosporin A or bongkrekic acid at 1 μ M. These data suggest an increase in mitochondrial lipid oxidation and the resultant change in MP may contribute to the induction of apoptosis by DHA with butyrate as shown previously.

This thesis is dedicated ~

to my dearly missed godmother, Mdm. Quay, Siew-Ching,, whose charisma has touched my soul;

to my incredible parents, Mr. Ng, Ah-Hai and Mdm. Ong, Soo-Jit, whose unconditional love has nurtured and brought me up to be a decent person;

to my two brothers, Kok-Ming and Kok-Leong, whose brotherly love has protected and saved me from many hurdles in life;

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to all of my friends, who have taught me many good lessons in life; and lastly,

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CHAPTER I

INTRODUCTION

Colon cancer is the second most prevalent type of cancer and is a leading cause of cancer death in the United States and in essentially all economically developed countries (1). Over 940,000 cases of colorectal cancer occur annually worldwide, among which North America is one of the continents reporting the highest incidence rate. It is noteworthy that 50% of these cases result in death (1, 2), *i.e.* colorectal cancer is responsible for some 470,000 deaths each year, among which 57,100 are Americans (1). Nevertheless, it is estimated that 70% of colorectal cancers are preventable by moderate changes in diet and lifestyle. For instance, the European Prospective Investigation into Cancer and Nutrition (EPIC) suggests that a daily consumption of 500 grams (1.1. lbs.) of fruits and vegetables can decrease incidence of cancers of the digestive tract by up to 25% (2).

Diet, tumorigenesis and apoptosis—an overview

Epidemiological studies indicate that populations ingesting higher amounts of fish are at a lower risk for colon cancer and a lower mortality rate of colorectal cancer (3-5), compared to those ingesting diets high in saturated fat or other polyunsaturated fatty acids (PUFA). These studies indicate a chemoprotective role for omega-three (n-3) PUFA. Dietary fibers also play a favorable role in the prevention of colon carcinogenesis (5-7). Different combinations of long chain PUFA and fibers in the diet modulate colon tumorigenesis. Most recently, dietary fish oil, and fiber, such as those from fruits and vegetables, in combination has been demonstrated to be beneficial in the prevention of colon cancer (3, 8). In a rat model of colon carcinogenesis, diets containing fermentable fiber and fish oil have been shown to protect against colon tumorigenesis, compared to

This thesis follows the style and format of Cancer Research.

diets rich in cellulose and corn oil (7, 9).

With regard to dietary intervention, apart from cell proliferation and differentiation, multistep programmed cell death, *i.e.*, apoptosis, has been extensively investigated as a biological marker regulating colon cancer risk. Cancer studies at the cellular level provide evidence that an abnormality in the molecular expression, configuration, functionality and/or interaction of proapoptotic signaling molecules contribute independently or synergistically to the inhibition of cancer cell growth (see review (10)). For instance, reactive oxygen species (ROS) and their counterparts, intracellular antioxidants, are involved in a multitude of biological effects including mutagenesis, tumor promotion, regulation of proliferation, activation or inactivation of biomolecules, regulation of transcriptional activities as well as the activation and execution of apoptosis.

Effect of dietary long chain fatty acids on cellular and mitochondrial membrane structures

Together with eicosapentaenoic acid (EPA, 20:5n-3), docosahaexenoic acid (DHA, 22: 6n-3) is a member of n-3 PUFA family found in high concentration in fish oil. Dietary fatty acids are incorporated to cell membrane phospholipids and have effects on, but not limited to, cell membrane composition and function (9, 11), membrane lipid oxidation (12, 13), transcriptional or translational regulation (14, 15), eicosanoid biosynthesis (15) and molecular signal transduction (10). These pleiotropic effects are believed to contribute to the observation that dietary fish oil, specifically n-3 PUFA, suppresses the initiation and promotion phases of colon cancer development.

Cumulative evidence shows that n-3 PUFA, in contrast to n-6 PUFA, uniquely affect cell cytokinetics by altering the cellular microenvironment through changes to membrane composition, thereby promoting apoptosis (7, 16-18). In animal studies, supplementing diets with fish oil, a rich source of EPA and DHA, significantly alters tissue levels of these fatty acids. n-3 PUFA from fish oil or purified ethyl esters were incorporated into colonocyte mitochondrial membrane

phospholipids (16, 18), which coincided with the enhancement of apoptosis in the colon (7, 19). A similar trend was observed utilizing colonocyte cell lines, in which the incorporation of DHA into cardiolipin, a mitochondrial membran: phospholipid, was associated with the induction of apoptotic signaling (11, 17). To date, the significance of fatty acid incorporation into membrane lipids in relation to the induction of apoptosis has not been determined. We hypothesize that the incorporation of n-3 PUFA increases cell susceptibility to oxidative damage, and subsequently perturbs the integrity of mitochondrial membranes.

n-3 PUFA and oxidative stress

Dietary n-3 PUFA may increase apoptosis in colonocytes via a mechanism involving ROS production (20). We have shown that an increase in ROS production following fish oil supplementation in the diet is correlated with the induction of apoptosis in colonic crypts (21, 22). In addition, data from cell culture studies have demonstrated that DHA treatment increases ROS production and induces apoptosis (11, 13, 16). Furthermore, antioxidants, which specifically target membrane lipid oxidation, partially reverse the effect of DHA-induced apoptosis in human colon carcinoma cells (17). This evidence in its entirety suggests that oxidative stress, in particular lipid oxidation may favorably modulate apoptosis in cells enriched with n-3 PUFA. It is noteworthy that even though EPA and DHA are both n-3 PUFA, they may exert protective effects against colon cancer through independent mechanism (23, 24). In the present study, DHA was used to determine the mechanism by which dietary fish oil relative to corn oil, reduce colon cancer risk.

DHA is a long chain n-3 PUFA, and compared to linoleic acid (LA, 18:2n-6), the major PUFA constituent of corn oil, is more susceptible to oxidation (25). The pro-oxidative properties of DHA are attributable to both the position and the number of its double bonds. Specifically, the oxidation of PUFA is linearly dependent on the number of double bonds present (13, 25). Ex vivo data have also shown that colonocytes are capable of incorporating substantial quantities of EPA and DHA, each of which contains five and 6 double bonds, respectively, are highly susceptible to

lipid oxidation (18, 19, 23). Interestingly, the number of fatty acid double bonds, ROS levels, and the proportion of cells undergoing apoptosis in cancer cell lines has been correlated (9, 13). In addition, with respect to fatty acid class, n-3 PUFA, *e.g.*, DHA is more susceptible to oxidation compared to n-6 PUFA, *e.g.*, LA (20). Collectively, these data support our hypothesis that n-3 PUFA, found primarily in fish oil, can induce apoptosis through a mechanism involving the oxidation of membrane phospholipids.

The oxidation of PUFA during metabolic stress gives rise to a series of toxic α,β-unsaturated aldehydes, *i.e.*, enals, including the electrophile, 4-hydroxy-2-nonenal (4-HNE). With respect to additional/alternative mechanisms of action, these class of lipid oxidation products could be cytotoxic and may be pro-apoptotic (26-28). In vivo, animal studies examining the effect of dietary fish oil or DHA supplementation on DNA lesion formation showed that a fish oil/DHA fortified diet significantly increased oxidative damage (20) but decreased the net DNA adduct (8-oxodeoxguanine; O⁶-methylguanine) levels in colonocytes (29). These observations suggest that dietary fish oil protects intestinal cells against oxidative DNA damage in part via deletion mechanisms, since the increase of DNA lesion was coupled to an incremental activation of apoptosis (27, 29, 30).

In addition to increasing oxidative stress by giving rise to pro-oxidants, DHA has the ability to modulate gene transcription and the homeostasis between ROS and antioxidants, thereby rendering cells more susceptible to apoptosis. Specifically, DHA has been found to induce lipid oxidation indirectly by influencing gene transcription. PUFA, in general, are capable of enhancing the activation of PPAR- α , which subsequently has been linked to the induction of several genes encoding proteins responsible for lipid transportation and oxidation (14). Collectively, these data suggest that DHA may be pro-oxidative and pro-apoptotic via several overlapping mechanisms.

To further illustrate the importance of oxidative stress and antioxidants balance with regard to apoptosis induction, several studies have examined the activities and expression levels of intracellular antioxidants (31, 32). Endogenous antioxidants may attenuate ROS-induced cell

damage and reduce cell death, initiated by pro-oxidants, such as n-3 PUFA. In vivo, colon epithelial cells expressed a greater endogenous antioxidant capacity in part as a compensatory mechanism towards chronic oxidative stress exposure, compared to the small intestine, sharing similar morphology (33). It is now readily apparent that, besides being pro-oxidative, n-3 PUFA are capable of suppressing the activity and expression of intracellular antioxidant enzymes, thereby enhancing tissue susceptibility towards oxidative stress and subsequent apoptosis induction (20, 32, 34). For example, the activities of endogenous antioxidants, such as glutathione transferase, glutathione reductase, glutathione peroxidase, and catalase appear to be depleted in animals fed a diet enriched with n-3 PUFA compared with diets supplemented with n-9 monounsaturated fatty acid or with n-6 PUFA (35). In a preliminary study, we have also demonstrated that in the colonic crypts of animals fed a fish oil/pectin supplemented diet, the antioxidant activities of catalase and superoxide dismutase were lowered (33) resulting in an increase in ROS and oxidative stress. Thus far, from a disease prevention standpoint, accumulating evidence suggests that n-3 PUFA or DHA promote cancer cell death. These data suggest that the perturbation of redox balance within a cell may be beneficial to the host under certain biological conditions, especially in malignant host cells.

Interaction between dietary fatty acids and butyrate

In conjunction with n-3 PUFA, which have anti-carcinogenic and anti-tumorigenic effects, butyrate in the lumen may synergistically modulate colon carcinogenesis by enhancing or inducing ROS production in vivo (36-38). Our preliminary data show that the feeding of pectin, a fermentable fiber that increases luminal butyrate levels, upregulates ROS production in the proximal colon, primarily in rats fed a fish oil supplemented diet (22, 30, 33). Ex-vivo incubation of rat colonic crypts with physiological concentrations of butyrate also exhibited a similar ROS upregulation property (9). In contrast, n-6 PUFA incorporation into colonic epithelial cell mitochondrial membranes in corn oil fed rats resulted in a decrease in ROS levels following butyrate treatment (19).

Butyrate, along with other short chain fatty acids (SCFAs), mainly acetate and propionate, is produced in the colonic lumen by microorganisms via the break down of complex carbohydrates and fermentable fibers (*e.g.*, pectin from fruits and vegetables). Among the SCFAs produced, butyrate is the most efficient energy source and the most potent inhibitor of colon cancer invasion (6, 39). It is an essential energy source capable of activating apoptotic pathways in colonocytes. Mice with a homozygous deletion of the gene for short chain acyl dehydrogenase, which encodes an enzyme necessary for the mitochondrial β -oxidation of SCFAs, showed a reduced ability to metabolize butyrate, resulting in a greater than 90% decrease in apoptosis in colonocytes (6).

Localized to the gut, fiber fermentation permits colon epithelial cells to have direct access to an energy source, and creates a channel for butyrate to modulate colon cancer development through epigenetic or genetic means. It is noteworthy that the efficacy of butyrate uptake is dependent on factors such as the co-existing substrate in the lumen (40) and the pathogenesis phase of the mucosa (41). In addition, butyrate could modulate colon tumorigenesis via a gene transcription regulated pathway (42). Hence, the effects of butyrate may take time to appear and could be subtle in nature.

Butyrate can induce cellular oxidative stress and can modulate cellular gene expression leading to changes in cell cycle arrest, maturation, differentiation, as well as apoptosis (36-38, 42). These paradoxical effects of butyrate may be determined by the state of activation of the cells, such as the timing and the amount of butyrate administered, the source of butyrate, and the interaction with dietary fat (43, 44).

With respect to alternative mechanisms to induce apoptosis, early studies showed that butyrate is a potent histone deacetylase (HDAC) inhibitor, which causes nonspecific hyperacetylation of histones and successive alteration in gene expression (42, 45) in favor of colon cancer cell death. In addition, butyrate treatment can increase mitochondrial protein expression, which appears to be related to the enhancement of mitochondrial function and the induction of apoptosis in colonic epithelial cells (46). Our preliminary data, in particular, have shown that

butyrate may activate apoptosis through the activation of Fas/Fas-L and/or a mitochondrial regulated pathway (9, 47). In summary, data from these studies suggest that various mechanisms may account for the protective effects of butyrate against colon cancer. For the purpose of this study, oxidative stress induction and the mitochondrial-targeted apoptosis activation properties of butyrate were investigated.

Due to the complex interaction among dietary components, the ability and the effectiveness of butyrate to upregulate apoptosis and to protect against colon tumorigenesis appears to be dependent on the dietary lipid source, with fish oil being more protective than corn oil (9, 16). In immortalized mouse colon cell lines, the ability of PUFA, including DHA and LA to promote apoptosis was enhanced by a 24 h incubation with ≥ 1mM butyrate (16). Herein, we have hypothesized that butyrate creates a permissive environment for apoptosis in part by enhancing n-3 PUFA-induced oxidative stress.

Reactive oxygen species regulates cellular events

Preexisting oxidative stress levels within a cell differentially modulate cell sensitivity towards successive ROS assault (33, 37, 48). Under basal conditions or when transient oxidative stress is encountered, some cell populations have adapted to the existence of reactive intermediates by evolving defense mechanisms that either scavenge these intermediates or repair the damage they cause. On the other hand, a transient oxidative stress surge may also lead to cell death through apoptosis or necrosis and inhibit proliferation in some cell lines, which most probably does not favor cancer progression (28). It is possible therefore that ROS may represent key elements for selective and efficient apoptosis induction by natural antitumor systems (49).

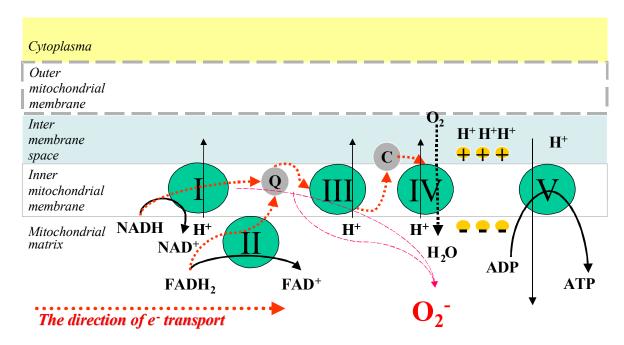
On the other hand, following prolonged exposure to oxidative stress, cells may become hyper-resistant to ROS-induced apoptosis, an event contributing to aging and to carcinogenesis (50). Interestingly, fish oil and butyrate are potent inducers of apoptosis, especially in malignant transformed cells that may be chronically overexposed to oxidative stress. Dietary fish oil or DHA

capable of altering the oxidative balance in colonocytes to rectify an increase in oxidative stress, could promotes apoptosis (3, 5, 17, 29, 34).

Intracellular formation and propagation of ROS

During aerobic metabolism, oxidative damage can accumulate in the mitochondria relative to the rest of the cell because electrons continually leak from the respiratory chain to form damaging ROS (49, 51). This oxidative damage impairs mitochondrial ATP synthesis and Calcium (Ca⁺⁺) homeostasis and may induce the opening of the mitochondrial permeability transition (MPT) pores, thereby triggering necrotic or apoptotic cell death (52, 53).

Within the inner mitochondrial membrane, NADH is oxidized to NAD⁺ by NADH-Ubiquinone reductase, while FADH₂ is oxidized to FAD by succinate-Ubiquinone reductase. The electrons from NADH and FADH₂ are both transferred to Ubiquinone, generating Ubiquinol. Located near complexes I, II and III is Ubiquinone (Q), which receives 2 electrons (to form Ubiquinol) from complex I and II and further passes electrons to complex III (to form Ubiquinone). In addition, Ubiquinone can donate electrons to oxygen to form superoxide (O₂⁻) (and Ubiquinol) thereby transferring ROS into the aqueous mitochondrial matrix. Ubiquinone is recycled to Ubiquinol and continues serving as an antioxidant within the inner mitochondrial membrane (51). Electrons flow from Ubiquinol through a series of cytochromes to oxygen, the ultimate acceptor of electrons, which is reduced to water at complex IV. The transfer of electrons along mitochondrial complexes I, III and IV are coupled to the pumping of protons from the mitochondrial matrix to the inter membranes space (Fig. 1). The extrusion of protons from the mitochondrial matrix generates a pH gradient and an electrical potential difference across the membrane, which together constitute the proton motive force that drives ATP synthesis (54). Protons are translocated down a gradient through complex V to produce ATP.



I: NADH-ubiquinone reductase; II: Succinate-ubiquinone reductase; III: Ubiquinol-cytochrome C reductase; IV: Cytochrome C oxidase; V: ATP synthase; Q: Ubiquinone/ubiquinol; C: Cytochromes

Fig. 1. Electron transportation and mitochondrial complexes. Electrons (e) are transported down mitochondrial complexes I, III and IV, a process coupled to the pumping of protons (H^+) from inside of the inner membrane to the inter membrane compartment. This electron transport process results in the leakage of superoxide (O_2^-) and the initiation of ROS propagation.

Incomplete reduction of oxygen during aerobic metabolism generates O_2^- . It is estimated that up to 10% of the reducing equivalents from NADH can leak out of the respiratory chain in the form of O_2^- (54). Mitochondria have the ability to counter oxidative stress hyper-production, *i.e.* electron transport and leakage, and damage by shunting H⁺ back to the mitochondrial matrix. Via the effect of a group of Uncoupling Proteins or through allosteric inhibition of the electron transport chain by ATP/ADP ratio through a negative feedback mechanism (54), hyper-production of oxidative stress could be regulated.

In many cases, the initially generated reactive intermediates in the mitochondrial respiratory chain convert cellular constituents into second-generation reactive intermediates capable of inducing further damage. Specifically, the majority of O_2 generated is dismutated to form H_2O_2 while some serves as a substrate in Fenton reactions or can react with nitric oxide to generate

other reactive nitrogen species and electrophiles. Among all, α,β -unsaturated aldehydes are the most studied membrane phospholipid oxidation products capable of denaturing protein or nucleic acids (20, 28) and causing apoptosis in some models (55).

Mitochondrial structure and membrane potential changes during apoptosis

The permeability transition (MPT) pore is a multiprotein complex formed at the contact site between the mitochondrial inner and outer membranes (Fig. 2). This multiprotein complex is constituted primarily of a voltage dependent anion channel (VDAC), anion nucleotide transferase (ANT) and cyclophylin-D (56-58). The MTP pore is directly involved in the regulation of mitochondrial matrix Ca⁺⁺, transmembrane potential and cellular activation of apoptosis. The binding of cyclosporin A (CsA) to cyclophylin-D and bongkrekic acid (BKA) to ANT, inhibits MPT activation, subsequently preventing the dissipation of mitochondrial membrane potential (MP) (57, 59, 60).

MPT is characterized by a sustained opening of MPT pores, following a Ca⁺⁺ overload in the mitochondria inner compartment (53, 57, 61), leading to apoptosis. The full-blown activation of MPT pores permits a non-specific release of soluble mitochondrial proteins from the intermembrane space during MPT (53, 57). Some well-studied mitochondrial proteins being released can serve as apoptosis inducing intermediates, *i.e.* cytochrome C and SMAC/DIABLO, by amplifying or transmitting apoptotic signals. On the other hand, transient or full-blown MPT activation has always been coupled by dissipation of the MP as a result of the disruption of the inner mitochondrial membrane and the inability of membranes to maintain a proton gradient. These changes occur in most, if not all, models of cell death and can be taken advantage of to detect apoptosis at an early stage. In addition, changes in the inner mitochondrial membrane and the permeability transition of the mitochondrial membrane are coupled in most cases to a series of catabolic reactions, which result in cell death (62). Interestingly, during the activation of the

energy requiring apoptosis, only a few polarized mitochondria can be detected, because a subpopulation of respiring mitochondria within an individual cell generate ATP needed for the completion of the apoptotic cascade (63).

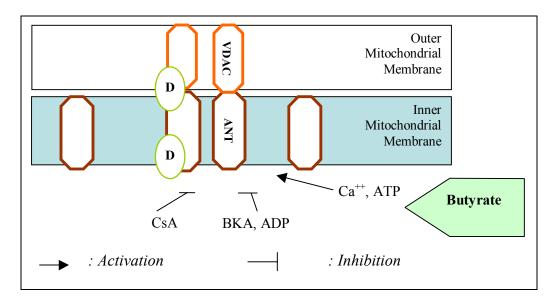


Fig. 2. Mitochondrial permeability transition pores. The mitochondrial permeability transition pore contains the voltage dependent anion channel (VDAC) and the anion nucleotide transferase (ANT). Cyclosporin A (CsA) and Bonkrekic acid (BKA) inhibit membrane permeability transition through binding to Cyclophilin D (D) and ANT, respectively.

Mitochondria related pro- and anti-apoptotic Bcl-2 family proteins

The activation of the MPT and subsequent dissipation of MP appears to be an early apoptotic event regulated by members of the Bcl-2/Bax family, and is irreversible in many cell lines, such as colon epithelial cells (9, 48, 62). Bcl-2 family proteins, which reside within mitochondrial membranes or the inter-membrane space, demonstrate unique binding domains for one another that affect their activation and function (64). These proteins dimerize or oligomerize with one another to induce or inhibit apoptosis, depending on their pro-apoptotic or anti-apoptotic roles (57). For example, the pro-apoptotic Bid protein and its truncated form (tbid) have a BH₃ only domain, which is required for the dimerization or oligomerization with other pro-apoptotic mitochondrial proteins to induce mitochondrial-involved apoptosis (64). In addition, tBid

specifically targets to the mitochondria via cardiolipin binding (65), causing the release of holocytochrome C, which is an effector in a cascade of events leading to apoptosis (65, 66).

We have found that fish oil feeding increases apoptosis while decreasing Bcl-2 expression in colonic crypts of carcinogen injected rats (67). In addition, butyrate via an unknown mechanism, also enhanced cytochrome-C translocation from the mitochondrial to the cytosolic compartment in rats fed a fish oil or n-3 fatty acid ethyl ester rich diet (9). Hence, we have hypothesized that DHA and butyrate induce apoptosis through a mitochondria mediated pathway.

Oxidative stress and mitochondria function

The precise role of mitochondrial ROS and membrane lipid oxidation in colon cell death remains elusive. A plethora of publications have shown that a oxidative assault triggers MPT thereafter activating apoptosis in many cell lines (49, 52, 53, 68). In addition, it is noteworthy that a full-blown MPT and ROS production may amplify each other to warrant an induction of apoptosis (53).

To date, limited evidence links n-3 PUFA and MPT directly to apoptosis in colonic epithelial cells. Herein, we investigated whether DHA and lipid oxidation of mitochondrial membrane phospholipids, followed by oxidative activation of MPT, is capable of triggering colonocyte apoptotic pathways.

Diet modulates mitochondrial membrane permeability transition and apoptosis

Dietary factors may alter the metabolism and redox balance within mitochondria, thereby priming cells for the induction of apoptosis. The acyl groups of mitochondrial phospholipids, *i.e.* cardiolipin, glycerophosphocholine and glycophosphoethanolamine, are altered in rats fed diets differing only with respect to fatty acid content (9, 11). Interestingly, we have shown that the incorporation of n-3 PUFA into glycophosphoethanolamine is associated with the dissipation of MP (19). In addition, butyrate treatment and butyrate/n-3 PUFA co-treatment have been shown to

enhance cytochrome-C relocation (9), MP dissipation and apoptosis in colonocytes (17, 46, 47).

We were able to demonstrate that the pronounced effect of n-3 PUFA or dietary fish oil on apoptosis can be reproduced in a cell culture model. Previously, using young adult mouse colonic (YAMC) epithelial cell lines, we have demonstrated that incubation with DHA increases caspase-3 activity, *i.e.* a marker for apoptosis (16). The conditionally immortalized mouse colon cell line, YAMC, bears a SV-40 heat sensitive large T antigen gene and can escape cell cycle arrest regulated by p53 and retinoblastoma when incubated in a permissive environment (33°C with 5% CO₂) with γ-interferon (16, 47, 69, 70). p53 signaling regulates cell cycle delay and apoptosis in colon carcinoma cells (71, 72). Studies have suggested that dietary factors may differently affect p53-dependent and p53-independent pathways to colon cancer (72). This is relevant because butyrate is capable of inducing apoptosis in several human colon carcinoma and adenoma cell lines in a p53-independent manner (47, 69, 73). Furthermore, mutations in the p53 tumor suppressor gene have been found in more than 50% of human colon tumor cell types (73-75). These data in its entirety suggests that YAMC cells culture is an appropriate model to further investigate the mechanism by which DHA and butyrate modulate mitochondria function.

CHAPTER II

HYPOTHESIS AND OBJECTIVES OF STUDY

Hypothesis

The incorporation of n-3 PUFA increases cell susceptibility to oxidative damage, and subsequently perturbs the integrity of mitochondrial membranes, a process potentiated by butyrate.

Objectives

1. To quantify cellular and mitochondrial membrane lipid oxidation in colonocytes following n-3 and n-6 PUFA incubation.

Lipid oxidation was detected using diphenyl-1-pyrenylphosphine (DPPP). DPPP is a novel non-toxic lipophilic fluorochrome that is retained within lipid membrane compartments, wherein it reacts specifically with lipid hydroperoxides to yield fluorescent DPPP-oxide (76). The fluorescence of this oxidized vital dye allows for quantification of membrane oxidant production in living cells.

Our previous data has shown that n-3 PUFA increases cytosolic, *i.e.* water soluble, ROS production in rat colon primary cultures (22, 30). To extend these observations and to assess membrane lipid oxidation, we investigated whether n-3 PUFA incorporation enhances membrane lipid oxidation in colon cell cultures. YAMC cells pretreated with BSA complexed-DHA (n-3 PUFA) or -LA (n-6 PUFA) at 50 µM with or without the addition of 5 mM butyrate were examined with respect to lipid oxidation.

2. To determine the association between membrane lipid oxidation and mitochondrial function.

MPT pore opening is an early and a pivotal event in the mitochondria-dependent apoptotic pathway, resulting in the dissipation of MP ($\Delta\Psi_{mt}$) and the induction of apoptosis (56). The relative levels of MP following butyrate/fatty acid co-incubation and under basal conditions were measured using a cell permeant, cationic lipophilic fluorescent dye, *i.e.*, Rhodamine 123 (Rhd123)

as described previously (9, 77). Briefly, this fluorescence probe is accumulated in mitochondrial matrix by the MP and lipid solubility. In general, a stable real time MP level indicates a dynamic equilibrium of the probes entering and leaving the mitochondrial matrix, *i.e.*, the accumulation of Rhd 123 may fluctuate as the environmental factors (*e.g.* temperature) change, hence a time matched control is used for normalization for each treatment.

To determine if lipid oxidation is localized within mitochondrial membrane phospholipids, we evaluated the effects of MitoQ. MitoQ refers to a mixture of mitoquinone (oxidized) and mitoquinol (reduced) moieties, which are the two redox forms of mitochondria-targeted quinones. The antioxidant activity of MitoQ is associated with mitoquinol's ability to take up electrons from the vicinity of complexes I and II and to donate them to complex III of the respiratory chain or to oxygen to form O_2^- at the aqueous phase (78, 79). This reduced moiety is also recyclable within the respiratory chain through redox reactions. Based on a study by Kelso et al. (78), the covalently attached triphenylphosphine cation and lipophilic alkyl chain of MitoQ contribute to its uptake and accumulation within the matrix face of the inner mitochondrial membrane, where most of the mitochondrial ROS production occurs. The driving force of MitoQ uptake is the membrane potential. While mitochondrial MP is the greatest MP among all organelles, it is assumed the preferential and predominant accumulation of MitoQ is localized to the mitochondrial compartment (61, 79).

3. To determine if butyrate potentiates lipid oxidation and the alteration of mitochondrial function in colonocyte membranes enriched with n-3 and n-6 PUFA.

Select YAMC cultures were incubated with butyrate (0-10 mM) for 6, 12 or 24 h with BSA-complexed DHA or LA (50 µM) or with no fatty acid for 72 h. Butyrate or vehicle (RPMI 1640 complete media) was incubated as indicated in protocol #A-5. Lipid oxidation and MP dissipation were measured using fluorescence microscopy imaging techniques.

CHAPTER III

MATERIALS AND METHODS

Materials

RPMI 1640 and Hanks' balanced salt solution (HBSS) were from Mediatech (Herndon, VA). Fetal bovine serum was from Hyclone (Logan, UT). Insulin, transferrin, selenium without linoleic acid were obtained from Collaborative Biomedical Products (Bedford, MA). *Leibovitz* buffer, GlutamaxTM and recombinant mouse interferon-γ (γ-IFN) were from GIBCO BRL (Grand Island, NY). Fatty acids were purchased from Nuchek Prep (Elysian, MN). The fatty acid free bovine serum albumin (BSA) for fatty acid preparation, Cell Death Detection ELISA, Plus and Cytotoxicity-Detection (LDH) kits were from Roche Applied Science (Indianapolis, IN). Fluorescence probes, diphenyl-1-pyrenylphosphine (DPPP), Rhodamine 123 (Rhd 123), 5-(and-6)-chloromethyl-2',7'-dichlorodihydrofluorescein diacetate acetyl ester (CMH₂-DCFDA) were from Molecular Probes (Eugene, OR). Cyclosporin A (CsA) was from Calbiochem (San Diego, CA). MitoQ was a gift from Dr. Michael Murphy (Medical Research Council Dunn Human Nutrition Unit, Cambridge, UK). All other reagents were obtained from Sigma (St Louis, MO).

Cell culture

YAMC cells were obtained from R.H. Whitehead, Ludwig Cancer Institute, Melbourne, Australia (70). For all fluorescence assays, prior to fatty acid pretreatment, $4\text{-}5\text{x}10^3$ YAMC cells (passages 17-23) per well were seeded into two wells sterile #1 German borosilicate chambered cover glasses (Nalge Nunc, # 155380) for 24 h to achieve 50-70% confluence. Under a permissive temperature, 33°C, mycoplasma free cultures were grown in RPMI 1640 complete media supplemented with 5000 u/L γ -IFN, 5% fetal calf serum, 1% insulin/transferrin/selenium without LA. For the Cell Death Detection assay, cells were incubated in 35 mm cell culture dishes at 35,000 cells per dish. See Fig. 3 for experimental design and timeline.

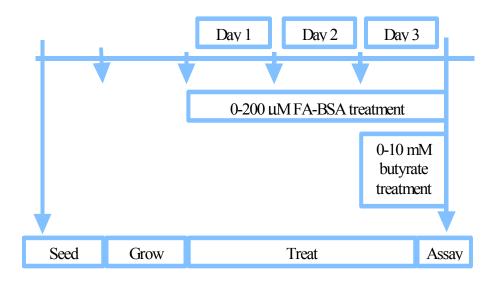


Fig 3. Experimental Design and Timeline

Fatty acid and butyrate preparation and incubation

DHA and LA were delivered to cells complexed to bovine serum albumin (BSA). To make the FA stock solutions, sodium salts of DHA and LA were complexed to fatty acid-free BSA in a 4:1 molar ratio, filter sterilized and stored in smaller aliquots in a freezer at -20° C (47). 2.5 mM BSA complexed fatty acid stock solutions were prepared and used within one-month period. LA is considered chemopromotive and served as the control in all experiments (16, 19).

Upon reaching 50-70% confluence, YAMC cells were pre-treated with or without DHA- or LA-BSA complex for 72 h at 50 μ M. Co-incubation with 0, 1 or 5 mM sodium butyrate or isobutyrate (negative control) dissolved in complete RPMI 1640 media was initiated during the final 6, 12, or 24 h of fatty acid pre-treatment. Isobutyrate is a branched-chain isoform of butyrate previously shown to induce neither differentiation nor apoptosis in colonic cell lines (80). The working solutions were filter sterilized before use.

Lipid oxidation and lipid antioxidants

Following fatty acid treatment, cells were washed twice with phosphate buffer saline

solution (PBS) and loaded with 5 μ M DPPP (final concentration) for 10 min in the dark at room temperature. The relative levels of DPPP-oxide fluorescence intensities were monitored using 351 nm (excitation) and 380 nm (emission).

Each analysis was performed under basal or oxidative stress conditions, *i.e.* with or without 25-100 μ M cumene hydroperoxide (CumOOH) or 10 μ M hydrogen peroxide (H₂O₂), to serve as a control. Due to the solubility of these compounds in the lipid bilayers, they differentially react with DPPP. CumOOH initiates lipid oxidation in the membrane, while H₂O₂ generates peroxides primarily and initially in the cytosol without affecting membrane oxidation (76).

To determine the association between PUFA treatment and lipid oxidation in the mitochondrial membrane, select cultures were treated with the mitochondrial targeted antioxidant, 10-(6'-ubiquinoyl) decyltriphenylphosphonium bromide (MitoQ) at 1 - 50 μM (78). 2.5 or 5 μM MitoQ was loaded one or 24 h prior to dye loading, *i.e.* during the last one hour or during the 24 h butyrate loading period. This antioxidant is expected to selectively decrease lipid oxidation within the mitochondrial inner membrane. A lower DPPP-oxide fluorescence intensity following MitoQ treatment would indicate the involvement of mitochondrial membrane lipid oxidation.

In comparison, a general lipid soluble antioxidant, vitamin E (α -tocopherol succinate; VEs), at both 10 μ M and 25 μ M, was used to estimate the involvement of total cellular lipid oxidation. A 10 mM VEs stock solution prepared in DMSO was filtered sterilized using DMSO safe tuffryn membrane (Acrodisc, #4433) and stored in smaller aliquots under an inert gas, N_2 atmosphere, at -80° C. The stock solution was subsequently diluted in RPMI 1640 media and added to culture.

Mitochondrial membrane potential

Following fatty acid pretreatment, cells were rinsed with PBS, incubated with 656 nM Rhd123 for 15 min at 33°C in the dark, and washed twice with PBS. Cells were then analyzed for

rhodamine fluorescence by confocal microscopy. The relative levels of Rhd123 fluorescence intensities were monitored at 488 nm (emission) and 530 nm (excitation) using confocal microscopy (Meridian Ultima). A reduction of fluorescence intensity reflects the decrease of MP.

Cytosolic ROS production

To measure the change in cytosolic ROS production, *e.g.* H₂O₂, after n-3 PUFA and n-6 PUFA pretreatment, 5 μM CMH₂-DCFDA (5-(and-6)-chloromethyl-2', 7'-dichlorodihydrofluorescein diacetate acetyl ester) dissolved in DMSO was loaded into selected cultures for 10 minutes in the dark at 33°C before imaging (19). Cleavage of acetate moieties by esterases inside the cell causes the 2,7-dichlorofluorescein to be trapped inside the cell, rendering it available for oxidation by peroxides or hydroperoxides (81, 82). ROS production was quantified based on the intensity of the oxidized fluorescent probe trapped within the cell. H₂O₂ (10 μM) served as a positive control to induce cytosolic ROS production. DCFDA was laser excited at 488 nm and monitored with a 530 nm barrier filter.

Fluorescence microscopy and quantification

Laser excitation confocal microscopy (Meridian Ultima, Okemos, MI) was used to capture all images requiring quantification of fluorescence intensity. Cells loaded with fluorophores and washed with PBS were replenished with phenol-free *Leibovitz* media (Gibco, #21083-027) for image analysis. *Leibovitz* media contains pyruvate for cells to be sustained in a CO₂ free environment temporarily. Excitation energy and emission filters were adjusted based on the property of each fluorescent probe. In addition, cells were observed for contamination and other artifacts using phase contrast microscope (Nikon) after each treatment. Groups of cells (>5 cells) from at least fourteen fields for each sample were captured using laser excitation confocal microscopy (Meridian Ultima). Average fluorescence intensities of all the pixels in the image of individual cells were captured to measure the mean brightness of each treatment.

Cell death detection assay

Following fatty acid treatment, floating cells were harvested, washed, lysed and centrifuged at 13,600 x g to sediment nuclei and intact cells. Supernatants containing histone-complexed DNA fragments were collected for the quantification of late stage apoptosis using an ELISA (47). Absorbance and reference wavelengths were set at 405 nm and 490 nm, respectively. Values were normalized by the numbers of adherent cells from each dish. As a control for the mouse cell lines, cultures of cells were incubated for 24 h at a non-permissive (39°C) temperature, in RPMI 1640 media without γ -IFN supplementation. We have previously demonstrated that both non-permissive (39°C) temperature and 5 mM butyrate incubation are capable of enhancing YAMC cell apoptosis (16, 69).

Cytotoxicity study

For maximum sensitivity, $1.4x10^4$ YAMC cells per well were seeded into 96-well tissue culture dishes and incubated for 24 h with RPMI 1640 medium supplemented with γ -IFN. Cells were incubated with culture medium containing the compound of interest, *i.e.* MitoQ, at 0-50 μ M for 0-72 h. Supernatants were harvested, and the amount of LDH released was assayed by ELISA. LDH release was compared to untreated wells, which were lysed with 1% Triton-X 100, *i.e.* total releasable LDH present in untreated cells, according to the manufacturer's instructions (Roche, cat#1644793).

Statistical analysis

For all lipid oxidation and cytosolic ROS production studies, the statistical tests were performed using contrast testing under linear mixed models (83). Fixed effects were constructed to indicate the treatments under which the data points were generated. Well level random effects were used to account for the dependency of wells within the same dish. When one set of

experiments were conducted over a prolonged time frame, a factor accounting for the day to day variation was also added into the model.

For all other studies, the effects of independent variables (main treatment effects) were assessed using SuperAnova. A difference between means was tested using the least square means test. A 95% level of probability was accepted as being statistically significant.

CHAPTER IV

RESULTS

DHA and butyrate synergistically enhance lipid oxidation

Dose dependently, butyrate treatment over a 24 h period increased lipid oxidation in DHA (50 μM for 72 h) primed YAMC cultures (Fig. 4A, data are shown at the end of Chapter IV and in Appendix B). DHA pretreated YAMC cells demonstrated an increase (p<0.05) in lipid oxidation after co-incubation with physiological levels of butyrate (1-10 mM). However, in LA pretreated cultures, no trend was observed. DHA or LA treatment, in the absence of butyrate, did not increase lipid oxidation, compared to no fatty acid treatment.

In select experiments, in which 5 mM butyrate treatment was carried out for 24 h, DHA primed cells showed a 29% increase in lipid oxidation (p=0.0013), when compared to no butyrate treatment (Fig. 4C). The increase in oxidation was detected as early as 12 h following butyrate treatment in DHA primed cell (26%; p=0.0095) (Fig. 4B). Specifically, lipid oxidation in response to butyrate (12 h) treatment was not different from the control, *i.e.*, the no fatty acid pretreated cultures. At 6 h, butyrate treatment was insufficient to induce a change in lipid oxidation in any of the fatty acid treatment groups. Figures 4D and E are representative photomicrographs of YAMC cells illuminated following the oxidation of the DPPP fluorescence probe in cell membrane lipid bilayers. Even though the difference in the induction of lipid oxidation with the addition of 24 h butyrate co-treatment was not significantly different between the two fatty acids (p< $\frac{1}{2}$ | t | = 0.0579), DHA pre-treated samples showed a two-fold greater induction of oxidation relative to the LA group (Fig. 4C).

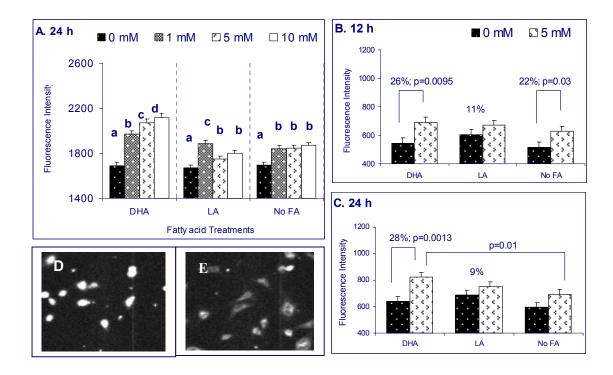


Fig. 4 . Effect of fatty acid and butyrate co-treatment on membrane lipid oxidation. Select YAMC cultures were incubated with butyrate (0-10 mM) for 12 or 24 h with BSA-complexed DHA or LA (50 μ M) for 72 h. A) Butyrate dose dependently increased lipid oxidation in DHA primed cells. In comparison, after co-treatment with 5 mM sodium butyrate for the final B) 12 h or C) 24 h of fatty acid treatment, DHA primed cells consistently exhibited a greater increase in lipid oxidation (p<0.01), when compared to no butyrate treatment, whereas LA treatment did not show an effect. Lipid oxidation was measured by quantifying the fluorescence intensity of oxidized DPPP incorporated into cellular membranes. Shown are the means (\pm SE) of changes in average pixel intensities of all cells in each treatment group. The number of cells analyzed (n) ranged from 181 - 280, collected from 5 independent experiments. Values not sharing the same letters are significantly different (p<0.05). Panels D and E show representative photomicrographs (200 x) of DPPP loaded YAMC cells treated with DHA and LA, respectively, for 72 h and with 5 mM butyrate for the final 24 h.

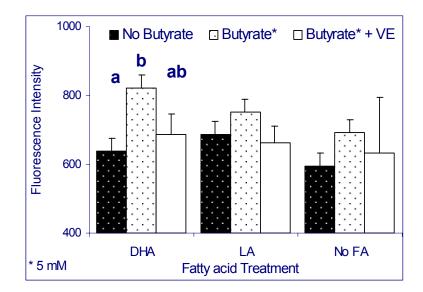


Fig. 5A. Vitamin E succinate reduces butyrate-induced membrane lipid oxidation in YAMC cells. Cultures were treated with 50 μ M BSA-complexed DHA, LA or with no fatty acid \pm VEs (10 μ M) for 72 h with or without butyrate (5 mM) for the final 24 h. Lipid oxidation induced by butyrate in DHA primed YAMC cultures was partially reversed by 10 μ M VEs incubation. The number of cells analyzed (n) ranged from 46-118, collected from 2 independent experiments. Values not sharing the same letters are significantly different (P<0.05).

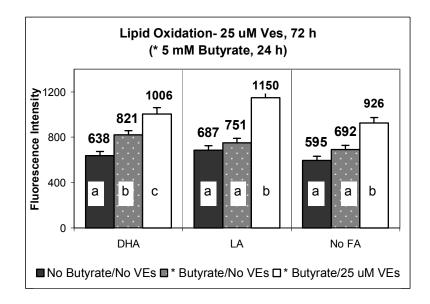


Fig. 5B. VEs (25 μ M) appears to promote lipid oxidation in YAMC cells. The number of cells analyzed (n) ranged from 46 -118, collected from 2 independent experiments. Values not sharing the same letters are significantly different (P<0.05).

Effects of vitamin E succinate and MitoO

The data in Fig. 5A demonstrate that the accumulation of lipid oxidation induced by DHA and butyrate co-treatment for 24 h, were partially reversed by a lipid antioxidant, Vitamin E succinate (VEs) at 10 μM (p<0.05). A similar anti-lipid oxidation effect of VEs was also detected in cells co-treated with butyrate for 12 h (Appendix B: Table B1-3). VEs did not affect the basal level of membrane oxidation in both LA-treated and the control groups. The effect of a higher dose of VEs (25 μM) was also evaluated, and appeared to promote lipid oxidation (Fig. 5B). The results depicted in Fig. 6 suggest that the subcellular origin of oxidation induced by DHA and butyrate was localized to the mitochondrial lipid membrane bilayers. Lipid oxidation induced by butyrate in DHA primed cells was blocked by a mitochondria targeted antioxidant, MitoQ (79). MitoQ (p<0.05) at as low as 2.5-5.0 μM (Figures 6A-B) reduced DPPP oxidation following a 12 or 24 h co-incubation period with butyrate (5 mM). In comparison, in LA treated cultures, MitoQ did not affect lipid oxidation. MitoQ toxicity was evaluated by determining the percentage of lactate dehydrogenase (LDH) release in YAMC cultures over a 12 - 24 h incubation period using 0 - 50 μM MitoQ. MitoQ concentrations up to 10 μM did not perturb cell viability (Fig. 6C).

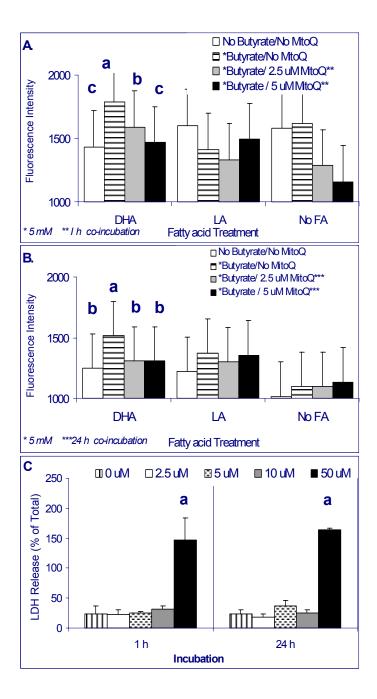


Fig. 6. MitoQ prevents lipid oxidation induced by DHA and butyrate co-treatment. Select YAMC cultures treated with DHA (50 μ M) showed an increase in lipid oxidation following 24 h butyrate (5 mM) co-treatment. A) Lipid oxidation was reversed by MitoQ incubation in a dose dependent manner when MitoQ was loaded for the final 1 h of butyrate co-treatment. B) Following 24 h of co-incubation with butyrate, MitoQ (2.5 or 5 μ M) reduced oxidation to basal levels. In LA (50 μ M) treated cultures, no increase in lipid oxidation was observed initially. C) The level of LDH release induced by MitoQ incubation indicates a lack of cytotoxicity. YAMC cells were incubated with complete medium containing MitoQ at 0-50 μ M, for one or 24 h. Supernatants were harvested, and the levels of LDH release were assayed, compared to an untreated well and expressed as percentages LDH present in cultures lysed with 1% Triton-X 100 (total releasable LDH). Data represent n=3 wells. For panels A and B, the number of cells analyzed (n) ranged from 283 -655, collected from 5 independent experiments. Values not sharing the same letters are significantly different (P<0.05).

Effects of fatty acids on mitochondrial MP and MPT

DHA incubation increased the basal level of MP in a dose dependent manner, whereas no effect was observed in LA treated cells (Fig. 7A). Compared to LA, 72 h DHA treatment increased resting MP by 14% (p<0.01) (Fig. 7B). This finding is consistent with previous observations (11) showing that DHA is capable of increasing resting MP.

A reduction of fluorescence intensity reflects a decrease in MP (Fig. 8). After 24 h butyrate co-incubation, the dissipation of MP was two fold greater in DHA (20%), compared to the LA (10%) treated cells (Fig. 8A). Fig. 8B shows that CsA at 1 μM, loaded 30 min prior to butyrate incubation and replenished 12 h after the initial dose, blocked the dissipation of MP following 50 μM DHA or LA treatment for 72 h with 5 mM butyrate co-incubation for the final 24 h. Similarly, BkA at 1 μM inhibited MP dissipation induced by the same reagents and treatment (Fig. 8C). Comparable observations were made upon co-incubation with CsA or BkA with butyrate for the final 6 and 12 h. The ability of CsA and BKA to prevent the dissipation of MP, a phenotypic event depicting the opening of MPT pores, indicates the activation of either a transient or a prolonged MPT (59).

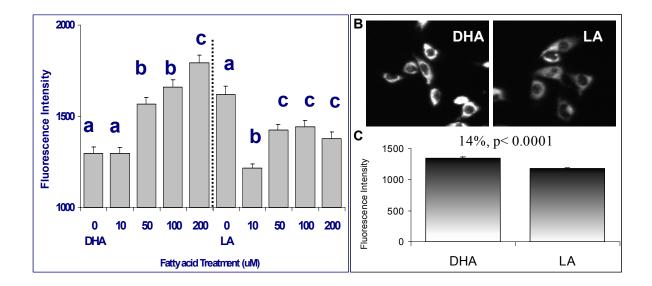


Fig. 7. Mitochondrial membrane potential is enhanced by DHA in a dose dependent fashion in the absence of butyrate. A) YAMC cultures were treated with DHA or LA at 0, 10, 50, 100 and 200 μ M for 72 h. To measure the changes in the basal level of mitochondrial membrane potential following fatty acid treatment, cells (post-treatment) were loaded with Rhodamine 123 (656 nM) for 15 min. Each treatment group contained 107 - 221 cells from 2 independent experiments. B) Representative photomicrographs (400X) of cells incubated with 50 μ M DHA or LA for 72 h. DHA enriched cells demonstrated a higher fluorescence intensity as compared to LA. C) In another set of experiments (n=4), DHA incubation, compared to LA, at 50 μ M increased resting membrane potential by 14% (P<0.0001). Mean (\pm SE) fluorescence values were from 4 wells from 2 independent experiments containing 1058 (DHA group) and 847 (LA group) cells. Values not sharing the same letters are significantly different (P<0.05).

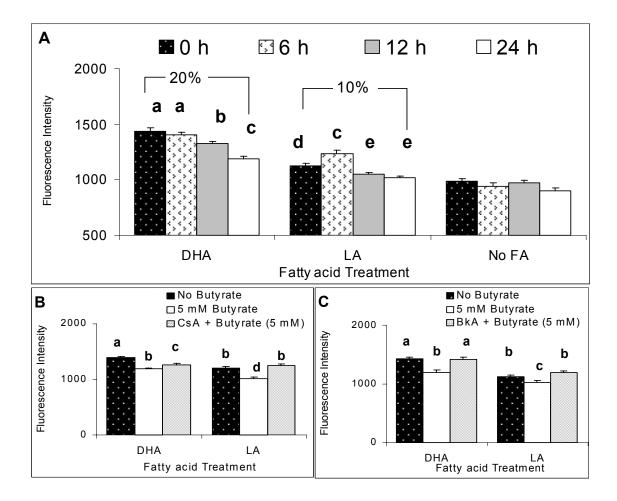


Fig. 8. Effects of fatty acid treatment on mitochondrial membrane potential. A) MP was dissipated by butyrate (5 mM) co-incubation, for the final 6, 12 and 24 h, following 50 μ M DHA or LA treatment for 72 h, in a time dependant manner. After 24 h butyrate co-incubation, the dissipation of MP was two-fold greater in the DHA (20%), compared to the LA (10%) treated cells. Mean (\pm SE) fluorescence values were collected from 4 wells from 2 independent experiments. The number of cells analyzed (n) ranged from 106 - 362. B) Cyclosporin A (CsA) at 1 μ M, loaded 30 min prior to butyrate incubation and replenished 12 h after the initial dose, blocked the dissipation of MP following 50 μ M DHA or LA treatment (72 h) with butyrate (5 mM) co-incubation for the final 24 h. C) Bongkrekic acid (BkA) at 1 μ M inhibited MP dissipation induced by butyrate (5 mM) co-incubation for 24 h. Similar observations were made upon co-incubation with CsA or BkA with butyrate for the final 6 and 12 h (data not shown). Values not sharing the same letters are significantly different (P<0.05).

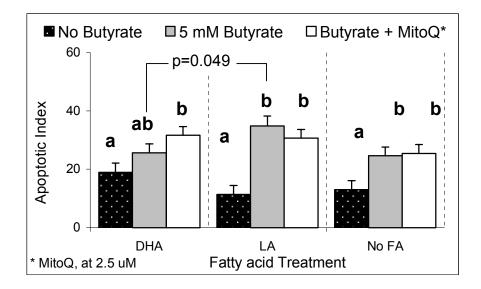


Fig. 9. Selective targeting of a redox active antioxidant (MitoQ) to mitochondria does not block apoptosis induced by fatty acid and butyrate co-treatment as indicated by DNA fragmentation ELISA. Cultures were incubated with 5 mM butyrate and 50 μ M DHA or LA or no fatty acid as described in Fig. 3. MitoQ at 2.5 μ M was loaded 30 min prior to butyrate incubation. The apoptotic index represents optical density, obtained by ELISA described in Material and Methods, per the number of adherent cells in culture. The number of cultures analyzed (n) ranged from 9-12, collected from 3 independent experiments. Values not sharing the same letters are significantly different (P<0.05).

Apoptosis phenotype

DHA and LA treatments (72 h) differentially primed cells for apoptosis induced by butyrate or butyrate plus MitoQ co-treatment. Using a Cell death ELISA (Roche), which measures DNA fragmentation (described in Materials and Methods), the effect of 24 h butyrate incubation to induce apoptosis in LA primed cells was significantly greater than in DHA primed cells (p=0.049). The result is in contrast to our other findings, which shows DHA and butyrate synergistically activate pro-apoptotic events. However, the control treatment butyrate at 5 mM incubated for 24 h induced apoptosis [Fig. 9], which is consistent with a previously published study (47). In addition, contrary to our hypothesis, this pro-apoptotic effect of butyrate was not blocked by MitoQ coincubation. MitoQ co-incubation, which prevented lipid oxidation, did not block apoptosis induced by butyrate, in all treatments.

Cytosolic reactive oxygen species

Using YAMC cell culture model, 5 mM butyrate treatment did not increase cytosolic ROS production, compared to no butyrate treated samples (Fig. 10). This observation is distinct from experiments using animal models (22, 30). The sub-confluent YAMC cell cultures were treated with 50 μM BSA-complexed DHA, LA or with no fatty acid for 72 h with or without 6 - 24 h butyrate (5 mM) co-treatment. The differences in the pair-wise comparison of ROS production between no butyrate and 5 mM butyrate treatments at the same incubation time and with the same fatty acid were obtained by comparing the average fluorescence intensity of YAMC cells loaded with CMH₂-DCFDA (5 μM) for 15 min prior to imaging.

Data collected to generate all of the observations above are summarized in tables 1 to 7 in the following pages.

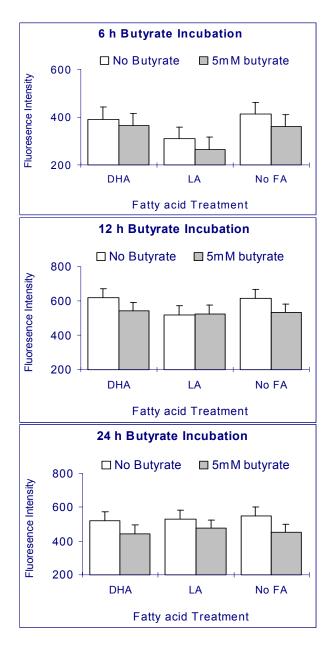


Fig. 10. Pair-wise comparison of reactive oxygen species production between no butyrate and 5 mM butyrate treatment at the same incubation time, treated with the same fatty acid. Cultures were treated with 50 μ M BSA-complexed DHA, LA or with no fatty acid for 72 h with or without a) 6, b) 12 or c) 24 h butyrate (5 mM) co-treatment. YAMC cells (post-treatment) were loaded with CMH₂-DCFDA (5 uM) for 15 min prior to imaging. Mean (\pm SE) fluorescence values were collected from 4 wells from 2 independent experiments. The number of cells analyzed (n) ranged from 434-735. Values not sharing the same letters are significantly different (P<0.05).

Table 1 Effect of fatty acid and butyrate co-treatment on membrane lipid oxidation

					•							
Fig. 4A		Mean Avg Intensity	Intensity			SE	E			Cell analyzed (n)	yzed (n)	
	0 mM	1 mM	5 mM	10 mM	0 mM	1 mM	5 mM	$10 \mathrm{mM}$	0 mM	1 mM	5 mM	10 mM
DHA	1690	1972	2071	2113	30.1	29.2	32.7	39.7	306	325	260	176
LA	1669	1887	1749	1802	27.8	27.2	24.4	22.9	360	375	466	531
No FA	1697	1842	1843	1868	21.6	26.0	24.6	28.6	594	409	459	340
Fig. 4B: 12 h	ų į											
	Mean A	Mean Avg Intensity	1	SE			Cell analyzed (n)	(u) p;				
	0 mM	5 mM	M	0 mM	5 mM	0 mM	ıM	5 mM				•
DHA	546	589		38	38	22	4,	213				
LA	603	029		37	37	22	8.	230				
No FA	516	625	_	37	37	280	0;	260				
Fig. 4C: 24	t h											
DHA		821		37	37	22	5.	229				
LA	289	751		37	37	202	2	181				
No FA	595	692		37	37	23	0;	245				

* Refer to appendix table B-1 to B-5 for more details

Table 2 Vitamin E succinate (VEs) reduces butyrate-induced membrane lipid oxidation in YAMC cells

Fig. 5		Mean Avg Into	ensity		SE	
	No Butyrate	${f Butyrate}^*$	Butyrate* + VEs (10 μ M)	No Butyrate	Butyrate*	Butyrate* + VEs (10 µM)
DHA	638.38	821.07	66'989	37.1567	37.1064	86.221
$\mathbf{L}\mathbf{A}$	687.16	751.17	662.24	37.1761	37.4652	86.221
No FA	595.17	692.11	632.57	36.6451	36.6513	81.3985

* 5 mM, 24 h

Table 3 MitoQ prevents lipid oxidation induced by DHA and butyrate co-treatment

Fig. 6A	Mean	an Avg Intensity	sity	S	Standard Error	÷	Ŭ	Cell analyzed (n)	(u
•	DHA	$\mathbf{L}\mathbf{A}$	No FA	DHA	$\mathbf{L}\mathbf{A}$	No FA	DHA	ΓA	No FA
No Butyrate/No MitoQ	1433	1601	1579	283.6	283.38	283.9	473	572	227
*Butyrate/No MitoQ	1785	1415	1619	283.65	283.16	283.3	405	548	552
*Butyrate/ 2.5 uM MitoQ	1590	1334	1285	283.43	283.2	283.22	296	628	550
*Butyrate / 5 uM MitoQ	1469	1492	1158	283.3	283.2	283.21	463	458	445
* 5 mM, 24 h									
*Butyrate/No MitoQ	1515	1371	1095	283.19	283.26	283.24	548	552	505
*Butyrate/ 2.5 uM MitoQ	1305	1301	1096	283.22	283.24	283.38	604	869	421
*Butyrate / 5 uM MitoQ	1304	1355	1132	283.25	283.36	283.52	558	482	381
* 5 mM, 24 h Data represent means for a minimum of 4 separate wells and 2 separate experiments.	inimum of 4 s	eparate wells	s and 2 separa	ıte experimen	ts.				

Fig. 6C: MitoQ-LDH release

) %	THUT	ed/Total Re	keleased/Total Releasable LDH	н)
Mito Q Concentration (µM)	0	2.5	S	10	50
Duration					
1 hr	23%	22%	79%	31%	147%
24 hr	24%	19%	37%	25%	163%

Number of wells analyzed, n= 3 per treatment.

Table 4 Mitochondrial membrane potential is enhanced by DHA in a dose dependent fashion in the absence of butyrate

Fig. 7A		Mear	Mean Avg Intensity	ensity				SE				Cell	Cell analyzed (n)	(u)	
	Mn 0	10 uM	80 uM	100 uM		Mu 0	10 uM	80 nM	100 uM	100 uM 200 uM 0 uM		10 nM	80 nM	100 nM	200 uM
DHA	1296.4	1296.0	1566.7	1296.0 1566.7 1659.7	l	36.0	30.9	36.6	39.3	42.4	107	160	145	115	68
LA	1619.2	1213.5	1423.5	1440.4		46.6	24.3	30.1	37.6	38.3	124	221	200	123	139

Table 5 Effect of fatty acid treatment on mitochondrial MP

Fig. 8A		Mean Avg Intensit	; Intensity			S	SE			Cell analyzed (n)	yzed (n)	
	0 h	4 9	12 h	24 h	0 h	4 9	12 h	24 h	0 h	4 9	12 h	24 h
DHA	1432.83	1406.01	1324.69	1188.94	30.07	24.90	20.94	18.57	207	322	362	308
$\mathbf{L}\mathbf{A}$	1124.85	1124.85 1235.15 10	1052.21	1017.32	28.17	27.31	15.85	17.83	162	199	343	243
No FA	983.73	939.26	973.52	903.57	27.00	28.53	22.40	22.10	124	106	121	117

Fig. 8B (7/30/2002)

((
	M	Mean Avg Intensity	nsity		SE)	Cell analyzed (n)	(u)
	No Butyrate	No Butyrate Butyrate* CsA	CsA + Butyrate*	No Butyrate	Butyrate*	+ Butyrate* No Butyrate Butyrate* CsA + Butyrate* No Butyrate Butyrate* CsA + Butyrate*	No Butyrate	Butyrate*	CsA + Butyrate*
DHA	1392.91	1183.87	1271.15	17.18	26.33	28.10	444	189	166
LA	1212.57	1007.72	1252.61	16.81	35.84	26.98	464	102	108

* 5 mM, 24 h

Fig. 8C (9/27/2003)

	M	Iean Avg Intensit	nsity		SE		_	Cell analyzed (n)	J (n)
ı	No Butyrate	Butyrate* BKA	BKA + Butyrate*	No Butyrate	Butyrate*	BKA + Butyrate*	No Butyrate	Butyrate*	Butyrate* BKA + Butyrate*
DHA	1432.83	1196.98	1420.77	25.95	34.22	36.96	207	118	102
LA	1124.85	1024.31	1189.58	29.33	31.44	32.74	162	141	130

* 5 mM, 24 h

Table 6 Pair-wise comparison of ROS production between no butyrate and 5 mM butyrate treatments of the same incubation time, treated with the same fatty acid

				•			
Fig. 10	Duration (h)	Mean Av	Mean Avg Intensity		SE	Cell an	Cell analyzed (n)
		No Butyrate	5 mM Butyrate	No Butyrate	5 mM Butyrate	No Butyrate	5 mM Butyrate
DHA	9	391.29	365.09	49.93	49.98	627	637
	12	619.87	542.77	50.07	49.99	602	649
	24	521.56	443.12	50.03	50.08	505	534
LA	9	308.57	264.63	49.87	50.01	735	578
	12	520.52	524.64	50.03	50.18	645	553
	24	531.23	474.83	50.06	50.34	999	434
No FA	9	412.28	360.39	50.08	50.06	559	571
	12	615.02	532.53	50.01	50.02	581	509
	24	549.26	451.55	50.04	50.27	486	454

Table 7 Selective targeting of a redox active antioxidant (MitoQ) to mitochondria does not block apoptotic induction by fatty acid and butyrate co-treatment

Fig. 9	Apoptotic Inc	poptotic Index (OD/ # cell	population)		SE		n = cultured plate (35mm dish) per sample	late (35mm dis	sh) per sample
			Butyrate* +			Butyrate* +			Butyrate* +
	No Butyrate Butyrate*	Butyrate*	MitoQ**	No Butyrate	Butyrate*	MitoQ**	No Butyrate	Butyrate*	MitoQ**
DHA	18.99	25.60	31.65	3.11	3.11	2.98	11	11	12
LA	11.33	34.83	30.64	3.11	3.43	2.98	11	6	12
No FA	13.03	24.65	25.42	3.11	2.98	3.11	11	12	11

*Butyrate 5 mM ** MitoQ, 2.5 uM

CHAPTER V

DISCUSSION AND SUMMARY

The preferential incorporation of n-3 PUFA, including DHA (11, 17), into mitochondrial membrane phospholipids, *i.e.* cardiolipin, increases cell (membrane) susceptibility to damage by oxidative stress (18). Data from our laboratory have suggested that mitochondria and ROS production mediate the apoptotic enhancing effect of butyrate and dietary fish oil in colonic crypts (7, 9, 19). The current experiments support this hypothesis, since DHA, a major fatty acid found in fish oil, enhanced YAMC mitochondrial lipid oxidation induced by co-incubation with physiological concentrations (0 - 10 mM) of butyrate (Fig. 4). It seems likely that the mitochondrial pool of phospholipid is the major site for lipid oxidation induced by butyrate and DHA, since the mitochondrial targeted lipid antioxidant, MitoQ, completely blocked the effect of butyrate (Fig. 6). In comparison, VEs, a lipid soluble antioxidant (Fig. 5) attenuated but did not fully prevent lipid oxidation.

The electrical potential across the inner membrane of mitochondria ranges between 150 and 180 mV, negative inside and is the highest electrical potential difference across any membrane in a typical cell, making up 70 to 80 % of the mitochondrial electrochemical proton gradient (61). In the present study, we show that DHA pretreated YAMC cells exhibit a higher resting MP (Fig. 7C), compared to LA. In addition, DHA up to 200 µM dose dependently increased the resting level of MP (Fig. 7). These findings suggest that incorporation of DHA into mitochondrial membranes induces changes in its bioenergetic properties: electron transport, ATP production, H⁺ permeability and MP, congruent with the report by Stillwell et al. (see (84) for review). It appears likely that DHA incorporation (and LA to a lesser extent) increased the function of the proton pump within the respiratory chain. The respiratory chain is coupled to the extrusion of protons from the mitochondrial matrix generates a proton gradient on the both sides of the inner

mitochondrial membrane. This proton gradient makes up part of ΔP (proton motive force), which is the driving force for ATP synthesis through ATP synthase. It is also the driving force for the uptake of various cationic molecules such as Ca^{++} (61). Therefore, the elevated MP associated with increasing DHA concentration is consistent with a fundamental shift in the activity of the respiratory chain.

An increase in mitochondrial bioenergetics is the result of increased ATP production and oxygen consumption (54, 57, 61). Under normal conditions, i.e., mild oxidative stress environments, a high ATP/ADP ratio inhibits the rate-limiting step, cytochrome-C oxidase activity through a feedback loop and causes a regulated mitochondrial respiratory rate. This is an established mechanism within the mitochondria to prevent ROS hyper-production (54). In comparison, the mild uncoupling effect of uncoupling proteins, e.g. UCP, to dissipate ΔP by permitting a controlled leakage of protons across the mitochondria membrane, allows electron transport to occur without the constraint of ATP utilization in some cells (54). Studies have shown that free fatty acids are capable of uncoupling energy production in isolated mitochondria. Unsaturated fatty acid incorporation could increase membrane permeability for H⁺ while increasing resting state respiration (84, 85). Nevertheless, the potency of this long chain PUFA to collapse MP decreased with increasing carbon chain length (85). DHA incorporation into living cells may increase respiratory chain action through its mild protonophoric action, to sustain a prolonged respiratory chain activation while maintaining an elevated resting level of MP as shown in Fig. 7. Consistent findings have been reported in other studies demonstrating that MP in DHA-enriched cells was higher relative to control groups treated with EPA, LA, arachidonic acid and oleic acid (11).

A population of cells exhibiting an increased MP should produce more cellular oxidants (85). Using the YAMC cell model, our data show that lipid oxidation in cells treated with 50 μM DHA or LA was not different from the control, in the absence of butyrate (Fig. 4A). Moreover, cytosolic ROS was not increased in all treatments with or without butyrate co-incubation (Fig. 9) as

compared to the no fatty acid treatment (control) group. Nevertheless, upon butyrate co-incubation for 24 h, DHA primed cells demonstrated a greater level of lipid oxidation compared to LA and no fatty acid group (Fig. 4C). These data suggest that DHA sensitizes or primes cells to subsequent oxidative stress events.

Since a more negative membrane surface potential, *i.e.*, a greater MP, has been correlated to an increase in the respiration rate (57) and DHA incorporation, which may favor PT pore opening (59), it is likely that DHA treatment primes cells for apoptosis via a mitochondrial PTP dependant mechanism.

Malis et al. (18) showed that the incorporation of PUFA into mitochondrial membrane phospholipids sensitizes cells towards the activation of apoptosis by Ca⁺⁺. Ca⁺⁺ is a well-known signaling messenger implied in MPT activation. An increase in intracellular free Ca⁺⁺ concentration is in most cases followed by an increase in the mitochondrial free Ca⁺⁺ concentration, of which the Ca⁺⁺ uniporter is believed to be one of the channels for Ca⁺⁺ uptake (52, 61). Ca⁺⁺ overloading has been shown to activate PT pore opening and has been correlated to ROS stress induction (52, 53, 61, 86). Furthermore, Ca⁺⁺ uptake into the mitochondrial matrix may be driven by MP (61). A cytosolic Ca⁺⁺ surge and subsequent mitochondrial Ca⁺⁺ overload would eventually reach a threshold level of Ca⁺⁺ to trigger MPT, resulting in MP dissipation (86). This is consistent with the effect of DHA in the absence of butyrate, where DHA dose-dependently increased resting MP levels, while LA treatment at identical concentrations did not (Fig. 4).

On the other hand, studies have shown that butyrate treatment, which is capable of increasing intracellular ROS stress (36, 37), also triggers the accumulation of endogenous Ca⁺⁺ within the intracellular compartment (87). It is known that supraphysiological concentration of Ca⁺⁺ suffice to induce PT, while lower doses facilitate the induction of PT by other stimuli (57). The increase in the resting level of MP by DHA and release of intracellular Ca⁺⁺ into the intracellular compartment by butyrate may facilitate a more effective uptake of Ca⁺⁺ into mitochondrial matrix. This is noteworthy, because as a ligand, Ca⁺⁺ binds to VDAC-ANT-

Cyclophilin D complex of the mitochondrial PTP, whereby it activates MPT (57).

Ca⁺⁺ accumulation within the mitochondrial compartment has been shown to trigger the release of PUFA enriched at the sn-2 position of phospholipids in mitochondrial membranes (18). Therefore, when cells are treated with butyrate and DHA together, these short and long chain fatty acids may induce Ca⁺⁺ overloading, which has been correlated to the uncoupling effect and the ability to induce MPT of free fatty acid. It is likely that these sequential alterations result in the perturbation of mitochondrial membrane and induce a mitochondrial-dependant apoptosis.

Oxidative damage of mitochondrial membrane proteins and phospholipids (49) can directly trigger MPT (53). We have demonstrated that DHA and butyrate dissipate MP that is preventable by CsA and BKA. It is possible that MPT is involved in the mechanism by which DHA and butyrate protect against colon cancer development. Literature review suggested oxidative stress accumulation within mitochondria activates MPT and is a self-amplifying process, meaning that several consequences of an oxidative reaction may feed forward to amplify MPT activation and ROS hyper-production within cells (57). Focusing on a single mitochondrion level, cardiolipin, a unique mitochondrial phospholipid localized primarily within the inner membrane, is needed for the activity of enzyme complexes involved in electron transport such as cytochrome-C oxidase and NADH reductase (11). The incorporation of DHA into of cardiolipin following dietary fish oil supplementation favors oxidative stress accumulation and apoptosis induction. Specifically, ROS production is correlated to the unsaturation index of cardiolipin acyl chains (9, 11). Moreover, the oxidation of the cardiolipin acyl chains is associated to the release of cytochrome-C from mitochondria triggering caspase activation (49, 68) an event inhibited by α -tocopherol, resveratrol, and Bcl-xl protein expression (49, 88). Furthermore, cardiolipin oxidation and subsequent cytochrome-C release are correlated to MP depolarization (52, 66), ROS generation (49) and MPTregulated apoptosis (89). On the other hand, redox metabolism, caspases, amphipathic peptides and Bcl-2 family proteins have been shown to act on mitochondria to induce MPT. Further research is required in order to elucidate whether these pathways are involved in the DHA and butyrate induced MPT activation.

The oxidation of PUFA yields α,β-unsaturated enals, *e.g.*, malodealdehyde (MDA), acrolein (Acr), crotonaldehyde (Cro), t-4-hydroxy-2-nonenal (HNE), and 4-oxo-2-nonenal (26, 28). These enals differentially react with deoxyguanine to form cyclic DNA adducts (an indicator of DNA lesion) or with protein to covalently modify the function within cells (28, 90). Interestingly, oxidized n-3 PUFA as opposed to the native unoxidized n-3 PUFA, are capable of inducing a PPAR-α mediated anti-inflammatory effect (90). With regards to the efficacy of a specific type of PUFA, DHA has been found to be the most efficient precursor for DNA adducts formation in some cell lines, when compared to other PUFA with lower numbers of double bonds (26). Specifically, the formation of DNA adduct is proportional to the number of double bonds of its PUFA precursors (26). It is likely that n-3 PUFA incorporation into mitochondrial membranes provides substrate for lipid oxidation (91) and increases the generation of aldehydes (enals), which are proapoptotic.

Concomitant with a role of ROS in the activation of apoptosis, the inhibition of ROS production by Bcl-2 and Bcl-xL is associated with their anti-apoptotic roles (92, 93). Interestingly, butyrate and n-3 PUFA have been found to decrease Bcl-2 expression in colon cancer cell lines (94) and in rat colonic crypts (67), respectively. In comparison, Bax protein (proapoptotic) increases the membrane unsaturation index, especially in cardiolipin, which promotes membrane lipid oxidation (88). Furthermore, this pro-apoptotic effect of Bax is enhanced in cells primed with polyunsaturated fatty acids. This evidence in its entirety suggests that DHA or n-3 PUFA exposure may increase oxidative stress and damage, hence priming cells for apoptosis induction by butyrate. This may explain why DHA and butyrate synergistically enhance apoptosis in the colon (7, 22, 30).

Other than Ca⁺⁺ overload and oxidative stress accumulation, a variety of different proapoptotic signal-transduction pathways may be involved in the activation of apoptosis induced by DHA and butyrate. For instance, butyrate incubation may activate apoptosis via a Fas receptor/Fas-L dependant pathway (47), which is an extrinsic pathway that may not involve the function of mitochondria. This may explain why MitoQ, which blocked lipid oxidation induced by butyrate in DHA primed cells (Fig. 3), did not block apoptosis (Fig. 6), compared to the control treated with fatty acid only. Alleva et al. has shown that coenzymeQ could block biochemical but not the apoptotic pathway that is Fas receptor-mediated (95).

It has also been noted that the endoplasmic reticulum (ER) can induce apoptosis independent (96) or dependent of mitochondria (97). Relative to mitochondria, the ER exhibits similar membrane potential maintenance properties and the ability to regulate Ca⁺⁺ homeostasis, and may therefore play a pivotal role in ROS production and subsequent apoptotic events. The involvement of ER in the regulation/activation of apoptosis in colonocytes needs further investigation.

Figure 11 illustrates the putative mechanisms by which the incorporation of DHA into mitochondrial membrane phospholipids results in an increase in MP. Intracellular events related to mitochondrial function are activated subsequent to butyrate treatment.

Significance

Collectively, these data further elucidate the apoptotic cellular targets that are modulated by DHA and butyrate in colonocytes, i.e. by eating a diet high in dietary fish oil and fermentable fiber. We propose that n-3 PUFA alter colonocyte mitochondrial membrane composition and function, thereby creating a permissive environment for apoptosis induced by subsequent pro-apoptotic agents, such as butyrate. Using non-malignant transformed immortalized YAMC, we tested our overall hypothesis that metabolically-induced (butyrate) stress, the result of membrane lipid oxidation and disrupts mitochondria function. We demonstrate for the first time that mitochondrial lipid oxidation was induced by DHA and butyrate co-incubation in living cells. In conclusion, dietary fish oil and pectin may prevent colon cancer tumorigenesis by modulating mitochondrial membranes lipid oxidation and function.

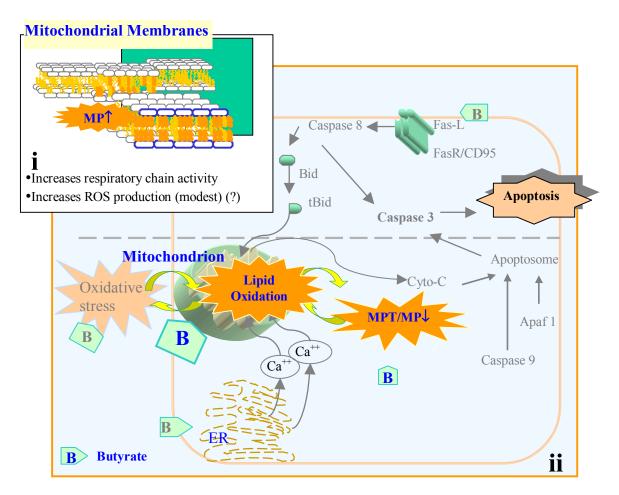


Fig. 11. i) Putative mechanisms by which the incorporation of DHA into mitochondrial membrane phospholipids results in an increase in membrane potential. ii) Intracellular events related to mitochondrial function are activated subsequent to butyrate treatment. The findings from this study show that mitochondrial membrane lipid oxidation and MPT activation are two events likely to be involved in the activation of cell death induced by DHA and butyrate co-treatment.

Abbreviations: B (butyrate), Ca⁺⁺ (calcium), Cyto-C (cytochrome-C), ER (Endoplasmic Reticulum), MP (mitochondrial membrane potential), MPT (mitochondrial permeability transition).

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APPENDICES

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- B. Experimental Results
- C. Results from Preliminary Experiments

Appendix A: EXPERIMENTAL PROTOCOLS

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APPENDIX A

EXPERIMENTAL PROTOCOLS

Protocol # A-1 Preparation of a complete RPMI 1640 media for YAMC culture

Pu	rpose: To prepare a complete RPMI 1640 media for YAMC cell culture.
Re	agents (supplier, catalog number), (storage) 500 ml (one bottle) RPMI 1640 without Glutamine (Mediatech, #15-040-CV), (walk in
	freezer)
	□ 26.6 ml Fetal Bovine Serum (Hyclone, # AK 12434), (chest freezer)
	□ 5.3 ml <i>Glutamax</i> TM (Gibco, 35050-061), (chest freezer)
	0.532 ml ITS "-" (Collaborative Biomed. Products, #4351 "-"), (P.H. fridge) ITS "" "reconstitutions add 5 ml sterile distilled water into learning and a gooden 1 ml.
	ITS "-"reconstitution: add 5 ml sterile distilled water into lyophilized powder. 1 ml of reconstituted solution is sufficient to supplement one
	liter of medium (0.1% dilution).
1.	Thaw FBS and $Glutamax^{TM}$ in the fridge several hours in advance or overnight.
	Add FBS, Glutamax TM and ITS "-" into 500 ml RPMI 1640 medium.
3.	Label bottle as "complete" media and store it at 4°C.
	nal volume: 531.9 ml
ГII	nal concentrations: 5% Fetal Bovine Serum, 1% Glutamax™, 0.1% ITS "-": Linoleic acid 0
	Insulin 5 μg/ ml
	Transferin 5 μg/ ml
	Selenious acid 5 ng/ ml
Th	e following reagent is to be added <i>fresh</i> into the media before use:
	5 units γ-IFN (Gibco BRL, #13284-021, stored in -80°C freezer) per 1 ml complete RPMI 1640 media.
1	For each 10 ml complete DDMI 1640 medium, supplement lular IFM
1.	For each 10 ml complete RPMI 1640 medium, supplement 1μl γ-IFN.
	Protocol # A-2 Preparation of YAMC cell culture
Pu	rpose: To start a culture by growing cells in a T-75 flask.
D۰	operation
	eparation: In warm water bath (37 °C), warm tap distilled water in a clean beaker for 30 min. Turn on the UV in the hood ~15 min prior to using the hood.
Pro	ocedure:

1.		Warm prepared complete RPMI 1640 media to RT.					
2.		Remove a vial of YAMC cells from the liquid nitrogen storage system and thaw it in the					
		pre-warmed water immediately.					
3.		Add thawed cells into ~10 ml media in a 15 ml conical tube.					
4.		Spin down YAMC at 200 x g (1096 rpm: tabletop centrifuge in room 307) for 5					
т.	min at 6 Acc./Dcc.						
5.		Aspirate supernatant to remove freezing media.					
6.		Resuspend pellet in 20 ml complete RPMI 1640 media with 2 μl of γ-IFN.					
7.		Add cells/media into a T-75 flask.					
8.		Incubate cell culture at 33°C under 5% CO ₂ atm.					
	o-da	ys after:					
9.		Feed cells every 2 (maximum 3) days by aspiring old media and by replenishing cells with					
		fresh RPMI 1640 media with γ -IFN.					
10.		When culture reaches 70-90% confluence, passage cell culture through trypsinization					
		(refer to Appendix A: Protocol # A-3 YAMC Cell Culture).					
		Protocol # A-3 VAMC Call Culture: Seeding Possing or Freezing					
		Protocol # A-3 YAMC Cell Culture: Seeding, Passing or Freezing					
		Protocol # A-3 YAMC Cell Culture: Seeding, Passing or Freezing					
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Pu	rpos	Protocol # A-3 YAMC Cell Culture: Seeding, Passing or Freezing e: To pass and to seed or freeze YAMC cell culture.					
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□ Transfer 5 μl of the cell suspension into a 500 μl micro-centrifuge tube.

		Add 5 µl trypan blue, 40 µl PBS into the micro-centrifuge tube. Mix well to yield a homogenized 10% suspension.			
		Transfer $\sim 30 \mu l$ of the cell suspension onto a hemacytometer. Count cell: Cell density (cells count per ml) = <u>Living cells count x 10^4 x 10</u> # Squares counted			
		Seed cell according to desired density.			
	В.	To continue growing more cells in a T-75 flask: Divide cell suspension (sterile) at 1:5 ratio. Aliquot each portion into 20 ml media with γ -IFN. Seed cells into a T-75 flask.			
	_ _	Resuspend cells in 4 ml of freezing media (EmbryoMax, # S-002-D). Load each cryo vial with 1.0 - 1.5 ml cell suspension. Allow room for expansion. (Yield ~1 million cells per vial). Keep vials in Mr. Freezer (Nalgene, #5100-0001) at -80°C for a day (at most two days over the weekend) before transfering vials into liquid nitrogen for storage. Record rack, box and coordinate of storage location into log book.			
8.	Label v	rial/flask with name, date, passage number and cell type.			
		Protocol # A-4: Fatty acid-BSA complex			
	•	To prepare 2.5 mM fatty acid and bovine serum albumin (BSA) complex.			
		n: y caps and baked items (a spatula, a 100 ml beaker, two 4 ml glass conical vials). lassware and spatula wrapped in foil and taped with autoclave tape at 180°C for 4 h			
	Na ₂ CO RPMI Sterile Others: Fatty ac Rec	e BSA (BM, #100069), MW: 68000 3 (JTBaker, #3604-01), MW: 105.99 1640 medium (Mediatech, #150400LV) distilled water (Baxter, 2F7115) filters (0.2 μm, tuffryn membrane), syringes, and icebox. cid stocks in ethanol (stored under N ₂ at -80°C): corded density and date of fatty acid stocks: ample: DHA [10.837 μg/μl in EtOH], (2-01-02) LA [11.022 μg/μl in EtOH], (8-13-01)			
	ocedure				
1. 2. 3.		Aliquot ~40 ml of RPMI 1640 and 10 ml of sterile distilled water in a sterile hood. Keep stocks on ice before and after vortex stock FA-EtOH vigorously. Add 10 mg of FA (dissolved in EtOH) into 2 ml conical glass vial, respectively. ush tubes with N ₂ before returning stocks to -80°C for storage. cample: need 922.76 μl DHA [10.837 μg/μl] 907.28 μl LA [11.022 μg/μl]			

4.	Clean needle with organic solvent. Dry down the FA with low stream of N ₂ .				
5.	Make 0.05 M Na ₂ CO ₃ while the FA-EtOH is drying. Example: 53 mg Na ₂ CO ₃ in 10 ml sterile H ₂ O (need ~2 ml per FA-BSA)				
5.	Make 15% BSA solution.				
	Add 20 ml of RPMI 1640 medium into a 100 ml beaker.				
	Gently layer 3 g of BSA onto the medium. Do not stir but let the BSA powder slowly dissolve down to the medium.				
	Do not still but let the BSA powder slowly dissolve down to the medium.				
6.	After dry down the FA,				
	Add 2 ml of 0.05 M Na ₂ CO ₃ to each FA vial.				
	Flush the vial with N_2 before vortex extensively.				
	Sit the vials at RT for 1 h. Vortex the vials several times to aid FA <u>dissolve completely</u> .				
7.	Calculate the materials needed to make 2.5 mM FA-BSA complex at FA/BSA: 3/1 molar ratio.				
	Example: DHA (MW: 328.5): 10 mg in 2 ml 0.05M Na ₂ CO ₃ ; BSA (MW: 68000): 15% Solution				
	Calculate the volume of 15% BSA solution needed for 10 mg DHA to obtain FA/BSA: 3/1 mole ratio.				
	0.01 g DHA x 1 x 68000 (BSA MW) x 100 = 4.6 (ml)				
	$\overline{328.5}$ (DHA MW) $\overline{3}$ 15				
	Calculate the total volume of solution needed for 10 mg DHA to make 2.5 mM DHA-BSA				
	complex. 10 (mg of DHA)/ 328.5 (DHA MW.)/ 2.5 (mM) = 0.012177 (L) = 12.177 ml				
	10 (IIIg 01 DHA)/ 328.3 (DHA WW.)/ 2.3 (IIIW) = 0.012177 (L) = 12.177 IIII				
	Calculate the volume of basal RPMI medium needed for making the 2.5 mM DHA-BSA				
	complex.				
	12.177 (total volume) - 2 (FA in $0.05M \text{ Na}_2\text{CO}_3$) - 4.6 (15% BSA) = 5.577				
	→ DHA 15% BSA RPMI 2.5mM DHA-BSA				
	$\begin{array}{cccccccccccccccccccccccccccccccccccc$				
	2 III 4.0 III 5.577 III 7 12.177 III				
	**It's difficult to completely retrieve the 2 ml FA-Na ₂ CO ₃ from the glass vial. Therefore, take only 95% of everything to make the DHA-BSA complex.				
	1.9 ml DHA + 4.37 ml 15 % BSA + 5.298 ml RPMI.				
	 _ Apply similar calculation to LA.				
8.	 Add the following volume of FA-Na ₂ CO ₃ , 15 % BSA, and RPMI to 15 ml conical tubes.				
	 <u>FA-Na₂CO₃</u> <u>15 % BSA</u> <u>RPMI</u> <u>FA-BSA</u>				
	DHA 1.9 4.37 5.30 11.57				
	LA 1.9 5.12 6.53 13.55				
9.	 Flush tubes with N_2 . Shake them on a belly dancer for $0.5 \sim 1$ h at RT.				

10	Use 0.2 μ m tuffryn membrane to filter sterilize FA-BSA complex in a cell culture sterile hood. Aliquot the complex.
11	Store aliquots protected from light at -20 °C. Retain for only 1 month.
	Protocol # A-5 YAMC Cell Culture: Fatty acids, Butyrate and other Treatments Ref: Fan et al. Am J Physiol (1999)
Pu	arpose: To treat YAMC cells with butyrate and different polyunsaturated fatty acids.
<u> </u>	Sources: User: Passage: Rack: Box#: R&C: Row cells in a T-75 flask in complete RPMI 1640 medium with γ-IFN. Allow >24 h for cells to adhere and to recover from trypsinization.
D a	 ay 1: (for experiment using chambered cover glass) Treat 2 wells chambered cover glasses (Nalge Nunc Intl, # 155380): 1 ml media per well. □ Thaw. Filter sterilizes FBS with 0.2 μm tuffryn membrane (Acrodisc, #4192). □ Prepare 20 % FBS in RPMI 1640 media (always make 2~3 ml extra). Calculation (example): 2 ml FBS + 8 ml Media> 10 ml Media/FBS □ Load 1 ml media/FBS per well. □ Incubate for ≤ 4 h or overnight at 33 °C.
3.	Warm complete RPMI 1640 to RT. Trypsinize and seed cells (refer to Protocol # A-3) Seed YAMC at 7,000 cells/well in a 2-well chambered cover glass. # chambered slides needed.
5.6.	ml cells/RPMI media +µl γ-IFN (at 1µl per 10 ml media) In a conical tube: add together calculated volume of cell/media and fresh media with γ-IFN. Gently invert to mix Add 1 ml cells/ml (pipette to homogenize) into each pretreated well; incubate at 33 °C under 5 % CO ₂ atm.
Da	ay 3, 4 and 5: [Note: cells should have recovered from trypsinization.]
7.	Treat cells with 50 μM fatty acids. Check if vitamin E treatment is required. The treatment starts on Day3 and last for 72 h, refer to Media Preparation for details. Defrost FA-BSA stock [2.5 mM]. *FA stock should be prepared and used within a month and should be stored at -20°C. Prepare 50 μM fatty acids with complete RPMI 1640. Calculation: (0.02 ml stock + 0.98 ml Media> 1 ml FA/media) ml stock + ml Media> ml FA/media
	*Make fresh FA solution everyday and do not store left over FA stock. Add y-IFN (1 ul per 10 ml FA/RPMI 1640 media)

(An additional step for Day 5):

- Butyrate co-treatment
- □ Prepare 1 ml 500 mM Butyrate stock in a tube wrapped in foil to protect butyrate stock from light. Refer to **Media Preparation**.
- Start 5 mM butyrate treatment by gently pipetting in 10 μl of 500 mM butyrate during the designated time.
 Ref: Fan et al. Am J Physiol (1999)
- ☐ Gently pipette with a 1000P pipette to mix evenly.

 \Box Check if any of the following treatments is involved on day 5. Refer to <u>Media Preparation</u> for the following treatments.

☐ MitoQ ☐ Cyclosporin A (CsA) ☐ Bongkrekic acid (BKA)

A sample timetable: estimated butyrate treatment, dye loading and assay time

(** Modify timetable accordingly based on experimental design)

Incubation time (h)/		Butyrate	Dye loading	Assay time	Note
butyrat	e treatment	loading time	time		
6	+ But				
	- But				
12	+ But				
	- But				
24	+ But				
	- But				

^{**} During planning, expect to spend 7~10 min on imaging per dish.

Day 6: Imaging Day (Analysis is done in the Imaging Analysis lab in vet school) **Preparation:**

- □ Warm *Leibovitz* buffer (Gibco, #21083-027) & RPMI 1640 media to RT (<2 h) before start.
 - □ Bring the following to Image Analysis Lab.
 - a) In an ice chest with blue ice: fluorescence probes and 500 mM butyrate (and isobutyrate, when applicable).
 - b) In a case at RT: Fluorescence probe in media, *Leibovitz* buffer, complete RPMI 1640 with γ-IFN, PBS, gloves, timer, pipettes.
 - c) In a warm case: Cell culture covered in foil. In the case, keep a flask of warm water to maintain a warm environment for the cells and a thermometer to gauge the change of temperature during transfer.
- 8. ____Dye and inhibitors/control loading:

Refer to protocol #5: Lipid oxidation analysis using DPPP.

#6: Mitochondrial membrane potential analysis using Rhodamine 123.

#7: Cytosolic oxidation analysis using CMH₂-DCFDA.

Media preparation [Otherwise stated, all "media" refers to complete RPMI 1640] 5 mM (final concentration, i.e. f.c.) Butyrate (Acros, #263190050)

- Stock preparation: **protect from light and use within 4 h.
 - □ 500 mM butyrate stock: weigh out and dissolve 55 mg sodium butyrate in 1 ml media.
 - □ Filter sterilize stock using a 0.2 μm tuffryn membrane (Acrodisc, #4192).

^{**} Start dye loading for the next dish at ~3 min prior to the finishing of the previous one.

Load $10 \mu l$ butyrate [500 m]	M] stock per 1 ml YAM	IC cell/media.
5 mM (f.c.) Isobutyrate (Sigma, Stock preparation: 500 mM isobutyr	,	Ref: Heerdt et al. Cancer Res (1994) isobutyrate in 1 ml media.
□ Filter sterilizes st Load 10 μl isobutyrate [500	e .	fryn membrane (Acrodisc, #4192). AMC cell/media.
10 μM (f.c.) Vitamin E Succinate Stock preparation:	<u>ee (MW= 530.8)</u> (Sigma	a, $\#95255$) (DMSO (v/v%) = 0.1%)
 10 mM (VEs): dissolve 5. Filter sterilize 10 ml stocl Flush vials with N₂ and st Load μ VEs [10 mM] stoce 	k solution (Acrodisc, #4tore aliquots in -80°C.	4433) and make 1 ml aliquots.
0, 2.5, 5 μM (f.c.) Mito Q (MW= Stock in storage: 50 mM Mito Q/ Working stock solution prep	DMSO stored in inert g	Ref: Kelso et al J Biol Chem (2001) gas: Argon environment, in dark
X	mM(x) = (0.5 ml) 0.5 = 0.005 ml $M(10 \mu l) = (1 \text{ ml}) (5 \mu l)$	
b) 0.5 mM	$(100 \mu\text{l}) = (200 \mu\text{l}) (00 \mu\text{l}) = (200 \mu\text{l}) (00 \mu\text{l}) = (1 \text{ml}) (2.5 \text{ml}) =$	0.25 mM)
Mito Q [f.c.= 5 μ M]: Load 5 μ l Mito Q [50 mM	[] stock per 495 µl med	ia to yield 0.5 ml Mito Q [500 μM]. dia to yield 1 ml Mito Q [5 μM].
• • •	- ·	media to yield 200 μl Mito Q [0.25 mM]. media to yield 1 ml Mito Q [2.5 μM].
1 μM Cyclosporin A (CsA) (Cal Stock preparation:	biochem, #239835)	
	Add 0.8315 ml Ethanol	per 100 mg CsA. (aliquot and store stock
☐ 1 mM CsA working solu	Add 900 µl Media per	SO per 100 μl CsA [10 mM: stock] 100 μl CsA [1 mM: working stock] mbrane (Acrodisc, #4433).
	spirate old media, add a	A) CsA per 1 ml cell/media. another 10 μl CsA into each well at 12 h. m et al. Mol.& Cell Biol. (1998)
1 μM Bongkrekic acid (BkA) (M (100x) Stock preparation:	4W=486.6) (Sigma, #P	36179, in solution)
[500 mM] in advance).	(For treatment requiri	ng butyrate co-loading, prepare butyrate
butyrate, for co-loading)	• •	er 1 ml media (or media with 500 mM

	ad 10 μ l of the stock into each well with ~1 ml YAMC cell/media. ette to mix evenly.
Proto	ocol # A-6 YAMC Cell Culture: Diphenyl-1-pyrenylphosphine (DPPP) Loading and Control Treatments Pof: Tokshophi et al. Free Podio Piel Med (2001)
	Ref: Takahashi et al. Free Radic Biol Med (2001)
Purpose	: To study lipid oxidation using DPPP, lipid soluble fluorescents probe.
Cell cult	ture: refer to Protocol # A-4
5 μM (f.	c.) DPPP (Molecular Probes #D-7894) (MeOH: 0.0096%; DMSO: 0.096%)
Stoc	k preparation:
	4.7 mM DPPP: dissolve 5 mg DPPP in solvents: $250 \mu l$ MeOH + $2.5 m l$ DMSO, aliquot at $100 \mu l$ and store in $-20 ^{\circ}$ C for up to a month.
	Or, thaw a 100 μl aliquot.
	Vortex vigorously until all pellets dissolved.
;	5 μ M DPPP: add 53μ l DPPP [4.7 mM] in 50 ml PBS.
1	Aspirate old media from each well.
	Gently, rinse cells once with PBS.
	Load 1 ml DPPP/PBS [5 μM] into each well. Incubate at RT, protected from light, for 10 min.
	Wash twice with 1 ml PBS before adding 1 ml fresh <i>Leibovitz</i> buffer for imaging.
J	wash twice with 1 mil 1 bb before adding 1 mil fresh betoomiz butter for maging.
For con	trol treatments only:
	f.c.) H ₂ O ₂ (Sigma, #H1009)
	pare DPPP [5 μM] working solution in advance.
Sto	ck preparation:
	50 mM H ₂ O ₂ : 5.67 μl H ₂ O ₂ (30%) per 1 ml distilled H ₂ O.
(Calculation (example): $X (50 \text{ mM}) = (5 \text{ ml}) (10 \mu\text{M})$ $X = 1 \mu\text{l}$
	Add 1 μ l H ₂ O ₂ [50 mM] into 5 ml DPPP/PBS [5 μ M] for co-loading.
	Aspirate old media from each well.
	Wash each well with 1 ml PBS.
3.	Load 1 ml solution into each emptied well.
4.	Incubate at RT, protected from light, for 10 min.
5	Wash twice with 1 ml PBS before adding 1 ml fresh <i>Leibovitz</i> buffer for imaging.
	f.c.) Cumene hydroperoxide (Sigma, #C-0524) (EtOH: 0.048%)
	ck preparation:
	100 mM CumOOH: 18.6 μl CumOOH per 1 ml solvent: 500 μl 95%EtOH and 500 μl
	dH_2O .
	Flush the bottle of reagent with Argon gas. Store it in dark at 2-4°C.
(Calculation (example): $X (100 \text{ mM}) = (5 \text{ ml}) (25 \mu\text{M})$
	$X = 1.25 \mu\text{l}$
	Add 1.25 µl CumOOH (100 mM) into 5 ml DPPP/PBS [5 µM] for co-loading.
1	Aspirate old media from each well.

 Wash each well with 1 ml PBS. Load 1 ml solution into each emptied well. Incubate at RT, protected from light, for 10 min. Wash twice with PBS before adding 1 ml fresh <i>Leibovitz</i> buffer for imaging.
Fluorescence Microscopy Imaging Monitor fluorescence intensities at 351 nm (excitation) and 380 nm (emission) using confocal microscopy (Meridian Ultima).
Protocol # A-7 YAMC Cell Culture: Rhodamine 123 Loading and Inhibitor Treatments
Ref: Scaduto Jr, et al. Biophysical J (1999); Fan et al. Am J Physiol (1999)
Purpose: To study mitochondrial membrane potential using Rhodamine 123 (Rhd 123).
 Stock preparation: Prepare 13.12 mM Rhd 123: dissolve 5 mg Rhd 123 in 1 ml MeOH), aliquot at 100 μl and store in -20 °C for up to a month. Or, thaw a 100 μl aliquot. Prepare 656 nM working solution: Calculation (example): (x) (13.12 mM)= (50 ml) (656 nM) X= 2.5 μl Add 2.5 μl Rhd 123 [13.12 mM] per 50 ml of media. Aspirate old media from each well. Gently, rinse cells once with PBS. Load 1 ml (656 nM) Rhd 123 per well. Incubate at 33 °C protected from light for 15 min.
5 Gently, wash cells with PBS before adding 1 ml fresh <i>Leibovitz</i> buffer (Gibco, #21083-027) for imaging.
Inhibitors:
 1 μM Cyclosporin A (CsA) (Calbiochem, #239835) Stock preparation: 10 mM stock solution: add 0.8315 ml Ethanol per 100 mg CsA (aliquot and store stock solution at -80 °C). 1 mM CsA working solution: add 900 μL DMSO per 100 μl CsA [10 mM: stock] 100 μM CsA in Media: add 900 μL Media per 100 μl CsA [1 mM: working stock]
□ Filter sterilize using a 0.2 μm DMSO safe membrane (Acrodisc, # 4433). 30 min prior to butyrate loading, load 10 μl (100 μM) CsA per 1 ml cell/media For CsA txt >12 h: Without aspirate old media, add another 10 μl CsA into each well at 12 h. Pipette to mix. Ref: Dr Tjalkens; Bradham et al. Mol.& Cell Biol. (1998)
1 μM Bongkrekic acid (BKA) (MW=486.6) (Sigma, #B6179, in solution) (100x) Stock preparation:

		[700 M] · 1	(For treatment requiring butyrate co-loading, prepare butyrate
		[500 mM] in advance).	Load 48.66 µl BKA per 1 ml media (or media with 500 mM
		butyrate, for co-loading)	
		oad 10 μl stock per well wi pette to mix evenly.	th ~1 ml YAMC cell/media.
Fh	inre	scence Microscopy Imagi	ng
	_ Mo	onitor fluorescence intensit	ies at 488 nm (emission) and 530 nm (excitation) using confocal
mi	cros	copy (Meridian Ultima).	
	D	-4L# A Q X/A M/C/C-III	Coltonia CMH, DCEDA Loo For and Control Tourstonate
			Culture: CMH ₂ -DCFDA Loading and Control Treatments al J (1999); Barhoumi et al. Fundam Appl Toxicol (1996).
Pu	rpos		tive oxygen species production using CMH ₂ -DCFDA
		[5-(and-6)-chloromethy	'l-2',7'-dichlorodihydrofluorescein diacetate acetyl ester]
<u>5 µ</u>		f.c.) CMH ₂ -DCFDA (Mo	lecular probes, #C-6827)
		ock preparation:	
			FDA: 50 μg into 18 μl DMSO
		Carefully pipette up and c	
		Calculation (example): (2	x) (5 mM) = (50 ml) (5 μ M) X= 50 μ l
		Add 50 µl (5 mM) per 50	·
1.		Aspirate old media from	each well.
		Gently, rinse cells once	
		Load 1 ml (5 mM) CMH	
4.		Incubate at 33 °C protect	•
5.		Gently, wash cells with l	PBS before adding 1 ml fresh <i>Leibovitz</i> buffer for imaging.
		ol treatments only:	
<u>10</u>		(f.c.) H ₂ O ₂ (Sigma, #H100	99)
		ock preparation:) (200/) m on 1 and distilled H ()
		•	O_2 (30%) per 1 ml distilled H_2O O_2 (50 mM) = (5 ml) (10 μ M)
		Culculation (example). A	X = (30 HeV) $X = (3 HeV)$
		Add 1 μl H ₂ O ₂ [50 mM] μ	•
1.		Aspirate old media from	each well
2.		Wash each well with 1 m	
3.		Load 1 ml solution into e	
4.		Incubate for 10 min prior	

 $\underline{\textbf{25}\ \mu M}$ (f.c.) Cumene hydroperoxide (Sigma, #C-0524) (EtOH: 0.048%)

Sto	ock preparation:
	100 mM CumOOH: 18.6 μ l CumOOH per 1 ml solvent: 500 μ l 95%EtOH and 500 μ l dH ₂ O.
	Flush the bottle of reagent with Argon gas. Store it in dark at 4 °C.
	Calculation (example): $X (100 \text{ mM}) = (5 \text{ ml}) (25 \mu\text{M})$
	$\dot{X} = 1.25 \mu l$
	Add 1.25 μl CumOOH (100 mM) into 5 ml PBS.
1	Aspirate old media from each well.
	Wash each well with 1 ml PBS.
	Load 1 ml solution into each empty well.
4	Incubate for 10 min prior to dye loading.
Moi	cence Microscopy Imaging nitor fluorescence intensities at 488 nm (emission) and 530 nm (excitation) using confocal opy (Meridian Ultima).
	Protocol # A-9 Cell Death Detection ELISA, Plus Source: Roche, # 1774425 Ref: Fan et al. Am J Physiol (1999)
	2: To quantify apoptosis/histone-complexed-DNA fragments of floating YAMC cells and alize values to the number of adherent cells from the same culture dish.
Ref	ture: d 25,000 cells per 35 mm cell-culture-dish, 24 h prior to fatty acid treatment. er to Protocol # A-4 for fatty acid, butyrate and other treatments. e: load only 1.5 ml total volume per dish on the last day of FA treatment.
Prepara	ation:
-	1 2 ml epi-tubes (2 sets): floater collection (step 1) & total cell counting.
	1.5 ml epi-tubes: supernatant collection (step 11).
Reco	nstitute working solutions:
	Dissolve substances in bottle 1,2 and 3 in 450 µl ddH ₂ O for 10 min and mix evenly.
	Bottle 1: anti-histone-biotin
	Bottle 2: anti-DNA-peroxidase (store at 2-8°C for upto 2 months)
	Bottle 3: positive control
	ABTS tablets from bottle 7: one tablet/5 ml substrate buffer (bottle 6).
	**Store reconstituted ABTS protected from light for up to a month at 2-8°C.
	provide the free free free free free free free fr
Warn	m complete RPMI 1640 media and trypsin-EDTA to RT.
Assay P	rocedures:
Superna	atant preparation (35 mm dish): (in room 307)
1.	Swirl dish to gather floaters. Collect all (1.5 ml) media with floaters in a 2 ml epi-tube.

2. 3. 4. 5. 6.	 (Replenish each dish of adherent cells with 1 ml PBS and keep plates at 33°C. These cells will be counted within the next 3hs, see Cell Counting). Centrifuge floaters/media at 5,000 rpm (Eppendorf centrifuge) for 5 min at RT. Remove supernatant manually using a pipette without disturbing the pellet. Wash: resuspend pellet in 1 ml media (without γ-IFN). Centrifuge at 5,000 rpm (Eppendorf centrifuge) for 5 min at RT. Resuspend cell pellet in 125 μl lysis buffer (bottle 5).
7.	Lyse: mix thoroughly and incubate samples for 30 min at 4°C.
8.9.10.11.	the same time (in lab 321), Prepare immunoreagent, volume: 1/20 anti-DNA-POD, 1/20 anti-histone-biotin, 18/20 incubation buffer. [need 80 µl per sample] Prepare an ice-chest with ice. After 30 min incubation (step 7), centrifuge lysate at max speed (13,600 x g, Eppendorf centrifuge) for 10 min, at 4°C. Prepare a template and a microtiterplate (MP; with enough wells) to include all samples, controls and a blank. Set up the spectrometer to read at 405 nm (reference wavelength at 490 nm) from designated wells. Keep centrifuged lysate on ice, while transferring 100 µl of the supernatant carefully into
	an epi-tube (1.5 ml) without shaking the pellet. ISA procedures: Transfer 20 µl of all samples (include histone-DNA complex, bottle 3) into the middle of the MP well.
15.16.	Add to each well 80 μl of the immunoreagent using multiple channel pipette. Cover the MP with an adhesive cover and a foil. Incubate on a MP shaker under gently shaking (300 rpm) for 2 h at RT. Tape both sides of the modules to secure them. Remove solution thoroughly by inverting and tapping it on a piece of paper towel. Rinse each well (3 x) with 250 μl Incubation buffer (bottle 4). Remove solution carefully
19.	by tapping. Lights off. Pipette into each well 100 μl ABTS solution. Incubate on a plate shaker at 250 rpm until the color development is sufficient for a photometric analysis. (~10 min, but check after 5 min of incubation). Measure the optical density at wavelength set (step 11) against ABTS solution as a blank.
For 1. 2. 3. 4. 5.	reach sample from a 35 mm dish: Swirl dish to agitate and to remove non-adherent cells remaining. Aspirate off PBS. Load 0.5 ml trypsin-EDTA into each 35 mm dish containing adherent cells. Incubate at 37°C for 3 min. Terminate trypsinization by loading 1 ml complete RPMI 1640 media into each dish. Tap and swirl plate lightly. Collect all lifted cells in 1.5 ml epi-tube. Count cells using a hemacytometer.

Protocol # A-10 LDH Activity Assay for Adherent YAMC Cell Culture

Source: Roche, #1644793 Cytotoxicity-Detection Kit (LDH)

Purpose: To measure the cytotoxicity of a reagent on adherent YAMC cultures following a 24 h incubation period.

Cell culture/Assay preparation:

- □ All samples and controls should have an n>3. Prepare a 96-well template to include all samples and controls.
- □ Wash cells in complete RPMI 1640 medium. Dilute cell density to [~1.4x10⁴ cells/100μl complete RPMI 1640 media] for optimal sensitivity.
- Add 100 μl cell-suspension per well in a sterile 96-well tissue culture plate except for wells designated for background control, substance control I and substance control II.
- ☐ Incubate cells overnight (>24 h) at 33 °C for cells to recover from trypsinization.

Day 1 Preparation:

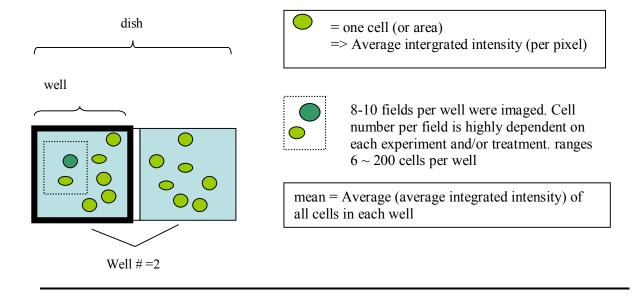
- \Box Prepare complete RPMI medium with γ -IFN.
- Prepare (2 x) solution of several concentrations of **test substance** (e.g. cytotoxic agents) in medium by serial dilution (will add 100 μ l medium to a final vol. of 200 μ l/well)

Procedures:

1.		Samples:
		Aspirate old medium and floating cells from wells. Add 100 μ l fresh media with γ -IFN into each well.
		Transfer $100 \mu l$ of the test substance dilutions into corresponding wells containing adherent cells.
2.		Low control : add 100 μ l medium (without test substance) to each of the triplicate wells containing 100 μ l/well cells.
3.		High control : add 100 μ l Triton-x-100 (Sigma, x-100) solutions (2% in RPMI 1640, filter sterilized) to each of the triplicate wells containing 100 μ l/well cells.
4.		Background control : fill 200 μl medium into each of the triplicate wells without cells.
5.		Substance control I : add 100 μl substance (maximum concentration used in the experiment) to triplicate of wells containing 100 μl/well medium without cell.
6.		Substance control II: load 50 μ l test substance + 50 μ l LDH standard + 100 μ l medium to triplicate of wells without cell.
7.		Incubate cells in a (33 °C, 90% humidity, 5% CO ₂) incubator for 24 h.
Da	y 2 A	Assay:
8.		Use a multi-channel pipette, transfer 100 μ l/well supernatant carefully from the culture plate into corresponding wells of an optically clear 96 well flat bottom MP.
9.		Add 100 μl reaction mixture to each well and incubate for ~30 min at RT, in dark.
10.		Set ELISA reader to measure absorbency at 490 nm (reference wavelength: 650 nm).
11.		Read.

Protocol # A-11 Image Analysis field selection

Purpose: To show the field of cells analyzed to facilitate statistical analysis and data interpretation.



Protocol # A-12 Criteria for cell selection

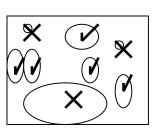
Purpose: To set up the criteria for cell (boundary of a cell) selection for all assays involving image analysis using the Meridian Instrument system.

Include:

a) only a WHOLE CELL. The keys are "whole" (see, **Exclude**, "c") and "cell". Make sure an area selected is a cell and use palette options, e.g., hue-lum to justify cell boundary.

Exclude:

- a) Any cell with an odd shape.
 - It is possible to have contamination of cell type. Select only epithelial cells, which are of research interest.
- b) Any cell with an odd size.
 Within each screen, select areas of similar sizes. Exclude those that are too big or too small. Refer to Figure 1 (below).
- c) Any cell that < 90% of its area is within the screen. Refer to Figure 2.
- d) Actively dividing cell.
 - These cells show different properties compared to the non-dividing cell. They are normally very bright regardless of treatment.
- e) Two neighboring cells that do not show clear edges or boundaries. Splitting the two cells subjectively may alter the average integrated values. Refer to Figure 3.
- f) When in doubt (exclude the cell)!





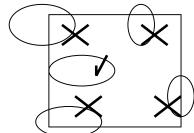


Figure 2

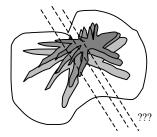


Figure 3

Protocol # A-13 Protein Extraction and Western Blotting

Purpose: To extract cytosolic and mitochondrial protein for protein assay or western blotting.

Mitochondrial fraction:

1 2 3 4 5	Resuspend mitochondria pellet into 100 µl RIPA buffer. Pass it through 29 gauge needle and sit it on ice 20-30 min. Label 500 µl tubes. Centrifuge at 12,000 x g for 10 min to remove unsolubilized material Transfer supernatant (solubilized mitochondrial perparation) to new 1.5 ml epi-tube Mix and make 11 µl aliquots. Store at -20°C.
6	Save aliquots of complete mitochondrial buffer and RIPA buffer for protein assay.

	Western blotting
Doy 1	Preparation:
Day 1-	Thaw samples in a ice chest with ice. Label 0.6 ml epi-tubes. Prepare western template sheet. Transfer membrane (Immobilon-P, Invitrogen, LC2005-2)
2.	Pipette samples, H_2O , 5 x pyronin into 0.6 ml epi-tubes (final volume <35 μ l/well). Mix by gentle inversion. Quick spin in a table-top centrifuge. Heat at 98°C, 5 min (or 10 min, if sample >20 μ l). Quick spin to collect condensation. Remove tape and comb off the gel cassette. Mark lane with a marker for visibility for sample loading. Align 3 rd grid with lower gasket and clamp unit. (For one gel, put a
6	alumine plate to block other side). Add "running buffer" into space between gels and into basin. Load samples (use 10 µl tips, helps to prevent bubble accumulation in the well). Lid on. Bring the unit into 4°C cold room. Connect electrodes to power supply (red to red, black to black). Run at 45 mAmp (1 gel) or 90 mAmp (2 gels) for 1 h. a dye front nears bottom of gel, turn off power supply, set current to 0, and unplug unit.
8	Pour "transfer buffer" into a clean staining tray. Return "running buffer" from unit into original bottle for re-use (*fresh running buffer runs faster).
9 10	Place gel case on bench, large side down. Crack it open with a spatula along the seam. To mark lane, cut the corner above the 1st lane of gel. Cut gel away from case from the thick bottom lip. Carefully transfer gel to staining tray containing transfer buffer. Set up transfer cassette in large staining tray: black side down, thick sponge on bottom.
	Immerse cassette and sponge in transfer buffer. Wet 6 mm filter paper with transfer buffer, top with a thick sponge in landscape orientation.
	Place gel (quickly) onto filter paper, lane 1 on the right for the protein to end up on left side of membrane.
	Wet transfer membrane with methanol (use dry gloves and tweezers to avoid nonspecific blotching). Layer it onto gel. Prevent bubbles getting between the layers, especially between gel and membrane.
	Wet another filter paper with transfer buffer. Layer it onto membrane. Use test tube to gently roll out bubbles.
16 17	Place thin sponge onto filter paper. Roll out bubbles. Close cassette. Insert cassette into transfer unit (black side faces labeled "black", hinges face up). Put stir bar into bottom of transfer unit. Fill unit with transfer buffer. Bang unit to get rid of bubbles. Stir on stir plate in 4°C cold room.

19	_ Run unit at constant current 400 mAmp for 75 min (max 90 min, for bigger proteins).
While	transferring:
	Prepare 4% nonfat dry milk (Carnation)/PBS/Tween in a 50 ml tube (1.2 g/30 ml PBS/Tween).
21	After transfer, move cassette to a large tray. Open cassette. Clip off the top right corner of membrane to mark the first lane.
22	
at sol	hal: Stain gel by immersing gel in Coomassie Staining solution (reusable) on a shaker for 1 h RT. Rinse and de-stain with de-staining solution by shaking vigorously. Change de-staining lution every 15 min for as many times as needed. Look for any blue band or non-transferred otein.
23 24	Return "transfer buffer" from unit into original bottle for re-use. Prepare 4% nonfat milk/PBS/T (0.8 g/20 ml PBS/Tween). Immediately before adding transfer membrane, add 1°Ab. Mix gently. The membrane should always face up.
25	
Day 2	
26	_ Transfer membrane into tray with PBS/T for brief wash
27	Pour out PBS/T, add fresh PBS/T to tray. Shake vigorously for 10 min. Repeat once.
28	Prepare 4% nonfat milk/PBS/T in a 50 ml tube and pour into small tray. Immediately before immersing membrane, add 2°Ab. Mix gently.
29.	Transfer membrane into 2° Ab, shake gently at RT for 1 h.
30.	Transfer membrane into a tray with PBS/T for a brief wash.
31.	Pour off PBS/T and add fresh PBS/T to wash. Shake vigorously for 15 min. Repeat.
32	Mix Chemiluminescent Super Signal reagents A and B (0.5 ml each) into a 1.5 ml epi-tube. Mix by gentle inversion.
33	Using a pair of tweezers, place membrane in a folded acetate sheet, with protein side facing
34	up. Pipette reagent mixture from step 32 onto membrane. Gently lower acetate sheet to cover membrane with reagent mixture.
35	_ Incubate for 5 min at RT.
36	_ Transfer membrane to another new acetate sheet.
37	_ Immediately, read chemiluminescence on Fluor S-machine in room 435.
In rooi	m 435:
38	Open machine doors (top and bottom). Ensure camera is set at 1.4 aperture for maximum light exposure.
39	Software instructions: choose <i>Quality One</i> under <i>file</i> . Open a "new file" and select function "Fluor-S". Click on Select, Blotting chemiluminescent, and High Resolution.
40	Focus (at lower aperture, if too bright). Click once on the <i>focus</i> button until you hear
41	camera clicks. With a piece of paper with visible text, adjust focus on the camera. To properly position the membrane, make sure the numerical aperture is set at 1.4, close camera door (top). Next, click on position until you hear the camera clicks. Position
42	membrane Acquire for 1 sec to warm up camera. Refocus if necessary. Acquire image for up to 5 min (gradual increase).

43	On a pop up window, press and hold "Apple" and "B", and check "invert display" and "autoscale"
44	Save file and/or export in tif format.
	Protocol # A-14 Dual labeling of apoptotic/necrotic YAMC using Annexin-V FITC & Propidium Iodide (PI) Ref: Fan et al Am J Physiol (1999)
Pu	irpose: To quantify early and late apoptotic events in YAMC cultures; immunofluorescense technique(47).
	Caution: Sodium Azide and PI are poisonous and hazardous substances. Handle with care and spose of properly.
-	eagents/Materials: Annexin-V-FITC/PI Kit (Biosource, # PHN 1018): -Annexin V/FITC buffer saline: 1% BSA, 0.1% Sodium Azide, pH7.4 -PI: 50 μg/ml PBS
	-Annexin Binding Buffer (1x): 10 mM HEPES/NaOH, 140 mM NaCl, 2.5 mM CaCl ₂ , pH 7.4 PBS (sterile) (Invitrogen, 14190-B6) RPMI Media (Mediatech # 150400LV) with γ-IFN Sterile H ₂ O
	Sterile100 mm dish for cell culture <u>5 mM Butyrate</u> (Acros, #263190050) Dissolve 55 mg sodium butyrate in 1 ml RPMI media → [500 mM] Sodium butyrate stock
	Filter sterilize (0.2μm, tuffryn membrane-REF4192) Load 10 μl [500 mM] butyrate per 1 ml YAMC cell/media
Pr	eparation: Seed 100k YAMC cells in a 100 mm diameter dish (Falcon, # 353803) to grow for 72 h. Treat or induce apoptosis in cells with different treatments (e.g. 5 mM butyrate treatment)
Da	Tube 1: Dilute 10x Annexin-V binding buffer 1:10 in sterile distilled H ₂ O (10 μl in 90 μl). Tube 2: Mix 1x Binding buffer with AnnexinV/FITC and PI buffers (5 μl Annexin-V/FITC and 10 μl PI per 100 μl binding buffer) *** protect from light. Warm binding buffers and RPMI complete media (need ~6 ml per dish) in 37°C water bath. Turn on fluorescence microscope and make sure the system is working before beginning.
A)	ocedure: Floating cells: Collect old media with floating cells in a sterile 15 ml tube. Wash monolayer of cell with 6 ml sterile PBS. Collect sample into the same tube. ** Incubate adherent cell (for counting) in RPMI complete media/γ-IFN at 33°C while working on floating cells

3	Sediment floaters at > 200 x g (need review) for 5 min. Carefully pipette to remove
1	supernatant. Wash pellet in PBS (~6 ml). Repeat #3.
4 . —	Remove supernatant (~30 μl left). Gently tap to release cells. Resuspend pellet with 90 μl
J	of Tube 2 (Biosource suggest: 2-3x10\6 cells/ml of tube 2)
6	Remove $\sim 40 \mu l$ of the mixture to count cell using hemacytometer.
··	i) Without further delay,
	Resuspend (~ 50 μl "floaters" left) with 200 μl binding buffer from tube 1.
	Incubate at RT for 15 min in the dark. (Work on 2 nd tube while waiting)
ii	Mix well. Drop $\sim 40 \mu l$ of sample mixture on a microscope slide and cover with a
	cover slip, 24 x 60 mm. Invert slide before mounting it on the stage.
ii	Set magnification power at (10 x 20) X. Use fluorescence microscopy to capture 5
	fields per slide, 3 images per field: phase contrast, FITC and TX-Red filters.
i	v) Overlay images. Count positively stained cells and total cell per image.
D) A.	dherent cells only (-ve control):
	_ Aspirate old media.
2. —	Wash cells with 6 ml sterile HBSS.
3. —	Trypsinize adherent cells with 6 ml trypsin for 3 minutes.
4.	Resuspend cells with 12 ml complete media to stop reaction.
5	Transfer cell to a 50 ml falcon tubes & centrifuge at 200 x g for 5 min.
	Repeat step A) 5 & A) 6.
	Protocol # A-15 Agilant anti-active caspase-3 cell fluorescence assay Ref: Turner et al Can Lett (2002)
Purp	ose: To quantify apoptosis by active-caspase-3 immunostaining technique.
o ĵ	aration: YAMC cell growing and apoptosis induction, e.g. Positive control: 39°C w/o γ-IFN, 72 h, floaters only
	Segative control: 33°C w/ γ-IFN, adherent cells only
	Treatment control: 33 °C w/ γ -IFN, 6 h H ₂ O ₂ (150 μ M), adherent and floaters
	Test sample #1: 33°C, 72 h DHA (50 μ M) + 24 h Butyrate (5 mM), adherent and floaters
	Test sample #2: 33°C, 72 h LA (50 μ M) + 24 h Butyrate (5 mM), adherent and floaters
C4aal	
	k preparations: aining buffer (s.b.)
	5 μl SYT016 per 1 ml s.b [need:ml s.b, + μl SYT016]
	μg anti-caspase-3 (BD #559565) per 1 ml <i>BD perm/wash</i> solution.
	[need: ml soln + µg Ab]
D: 5	μg Cy5-conj. goat anti-mouse (Jackson IR #111-176-045) per 1 ml s.b.
	[needml s.b. + µg Ab]
	O Cyto-fix/Cyto-perm buffer (diluted in dH ₂ O) (BD, 554715) [need ml] BD perm/wash solution (dilute 10x in dH ₂ O) (BD, 554715)[(1x) needed ml]

Procedure:

 Harvest, count and resuspend cells at 5x10⁵/ml. Transfer 1 ml into a 2 ml epi-tube. Adjust cell density to 1x10⁶ cells/ml in stock B (0.5 ml stock B for 5x10⁵ cells). Incubate for 10 min, at 33°C. Wash cells in 2 ml s.b. and centrifuge at 200x g for 5 min. Resuspend 5x10⁵ cells in 250 μl stock E. Incubate for 20 min, at 4°C (in the fridge). Wash in 2 ml stock F. Centrifuge. Add 50 μl stock C: 1°Ab (to 5 x 10⁵ cells). Incubate on ice, 20-30 min. Wash cells in 2 ml s.b Centrifuge. Add 50 μl stock D: 2°Ab. Incubate on ice, 30 min, in dark. Wash cells in 2 ml s.b. 										
Instrument set up: Set Bioanalyzer (Agilent 2100) up for cell assay: insert pressure cartridge; set chip selector at position 2.										
 Sample loading: Centrifuge cells (200x g) to remove s.b. Dilute pellet with <i>cell buffer</i> (green) to 5x10⁵cells/250 μl. Check visually to make sure no cell clumps or agglomerates left. Pipett up and down to make sure cells are in suspension before loading. Load 10 μl of the chip <i>priming solution</i> (white) into the large <i>priming well</i> (PS). Wait for 60 sec for capillary force to fill all channels of the chip. 										
**Essential practices: To prevent the introduction of air bubbles: Insert tip of the pipette to the bottom & center of the well when dispensing. Use inverse pipetting: push slightly over the first resistant when filling; empty the pipette tip only to the first resistance.										
 ** Protect dye solution from light. 5 Load 10 μl of focusing dye solution (yellow) into the focusing well (FD) to adjust the bioanalyzer optics for each individual chip. 6 Load 30 μl of cell buffer (green) into wash of the 2 cell buffer wells (CB). 7 Homogenize sample through piptetting. Load 10 μl per sample well. ** Do not leave any well empty or the pressure cartridge may become clogged. Pipette 10 μl of cell buffer or a sample replicate into any empty sample well. 8 Loaded chip must be read in bioanalyzer within 5 min. 										

Appendix B: EXPERIMENTAL RESULTS

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Table B-1 Lipid oxidation induced by fatty acid treatments and the anti lipid-oxidation effect

of Vitamin E at 10 and 25 µM (figure 1&2)

		oi vitai	inn e at iv and	i 23 mivi (ligure	1&2)	
Butyrate	Vit E	Fatty acids	No bu	ityrate	5 mM	butyrate
incubation (h)	(μM)	_	¹ Fluorescence	² Standard Error	Fluorescence	Standard Error
			intensity		intensity	
12	0	DHA	546	38	689	38
12	0	LA	603	37	670	37
12	0	No FA	516	37	629	37
24	0	DHA	638	37	821	37
24	0	LA	687	37	751	37
24	0	No FA	595	37	692	37
12	10	DHA	597	84	605	86
12	10	LA	610	86	617	86
12	10	No FA	606	86	610	86
24	10	DHA	638	86	687	86
24	10	LA	688	86	662	86
24	10	No FA	664	84	633	81
12	25	DHA	474	55	495	59
12	25	LA	636	55	578	49
12	25	No FA	340	51	464	161
24	25	DHA	869	53	1006	54
24	25	LA	1027	47	1150	45
24	25	No FA	639	45	926	48

^TValues represent least squares means with (n) found in Table B-2, which is from 5 experiments.

Table B-2 The number of analyzed cells (n) per treatment group for each lipid-oxidation study

			511	uuy					
Date	well/txt	Treatment	VEs	No	FA	L	A	DI	HA
		Butyrate Incubation		12 h	24 h	12 h	24 h	12 h	24 h
2002.10.18	2	Control		99	91	84	69	77	72
		5 mM butyrate		76	100	73	42	63	67
2002.11.15	4	Control		25	19	23	22	23	20
		5 mM butyrate		21	23	21	21	22	21
2003.01.30	2	Control		16	12	12	12	16	12
		5 mM butyrate		14	12	14	13	12	12
2003.02.10	2	Control	-	14	12	13	14	12	12
(10 µM Vit E)			+	12	13	12	13	13	12
		5 mM butyrate	-	12	12	12	12	13	12
			+	12	14	12	12	12	12
2003.02.21	2	Control	-	126	96	96	85	96	109
(25 μM Vit E)			+	63	117	40	89	43	50
		5 mM butyrate	-	137	98	110	93	103	117
			+	82	77	64	106	34	46

Total	12	Control	-	280	230	228	202	224	225
			+	75	130	52	102	56	62
		5 mM butyrate	-	260	245	230	181	213	229
			+	94	91	76	118	46	58

Table B-3 Fluorescence intensities and number of analyzed cells (n) per treatment group in

butyrate dose response study

FA treatment	Fluor	escence	Intensi	ty	S	tandar	d Error	•	Cell analyzed (n)			
Butyrate (mM)	0	1	5	10	0	1	5	10	0	1	5	10
DHA	1690	1972	2071	2113	30.1	29.2	32.7	39.7	306	325	260	176
LA	1669	1887	1749	1802	27.8	27.2	24.4	22.9	360	375	466	531
No FA	1697	1842	1843	1868	21.6	26.0	24.6	28.6	594	409	459	340

Table B-4 Percentage change in DPPP-oxide intensity following 5 mM butyrate co-incubation

DHA	LA	No FA
*26.2%	11.1%	*21.9%
*28.7%	9.3%	16.3%
	*26.2%	*26.2% 11.1%

^{*} Indicates a significant change in lipid oxidation following butyrate incubation within a FA group.

Table B-5 Differences and statistical significance in the induction of lipid oxidation following butyrate co-treatment among different fatty acid treatment groups

Treatment groups:

A = without butyrate and Vitamin E

B = with 5 mM butyrate (no Vitamin E)

C = with 5mM butyrate and 10 μ M Vitamin E

To explain notation: DHA vs. LA: A - B 12 h means comparing the difference, A - B, for DHA vs. LA for the 12 h data.

				Standard				
Label	Estimate Error	DF	t Value	Pr > t				
DHA v. LA: A - B 12h	-76.2967	73.3530		33	-1.04	0.3058		
DHA v. no FA: A - B 12h -29.6799	72.8201		31.7	-0.41	0.6863			
LA v. no FA: A - B 12h	46.6168	72.4843	31	0.64	0.5249			
DHA v. no FA: A - B 24h -85.7469	972.7508 31.9	-1.18	0.2472					
LA v. no FA: A - B 24h	32.9323	73.0122		33.4	0.45	0.6549		
DHA v. LA: A - C 12h	-46.0405	129.75	220	-0.35	0.7231			
DHA v. no FA: A - C 12h 35.4460	129.66	219	0.27	0.7848				
LA v. no FA: A - C 12h	81.4865129.49	219	0.63	0.5298				
DHA v. LA: A - C 24h	-73.5267	129.82	221	-0.57	0.5717			
DHA v. no FA: A - C 24h -11.2017	7126.50 199	-0.09	0.9295					
LA v. no FA: A - C 24h	62.3250 126.52	202	0.49	0.6228				
		- '						
DHA v. LA: B - C 12h	30.2562	129.87	221	0.23	0.8160			
				_				

DHA v. no FA: B - C 12h	65.1259 129.66	219	0.50	0.6160	
LA v. no FA: B - C 12h	34.8697	129.64	217	0.27	0.7882
DHA v. LA: B - C 24h	45.1525	129.89	222	0.35	0.7285
DHA v. no FA: B - C 24h	74.5452 126.48	197	0.59	0.5563	
LA v. no FA: B - C 24h	29.3927 126.61	201	0.23	0.8167	

Table B-6 Analysis of ROS Data – Pooled Dates (8/11 & 9/1/04)

Compare Time, FA & Butyrate

This analysis was for both dates pooled together. The analysis was completed using a mixed model in SAS to account for the extra variance from subsampling from well. Factors of comparison are FA, time, and butyrate. Date was included as a factor to represent the two separate dates.

Type 3 Tests of Fixed Effects

Effect	Num DF	Den DF	F Value	Pr > F
Date	1	53.4	20.56	<.0001 *
FA	2	53.3	1.73	0.1862
butyrate	1	53.3	5.77	0.0198 *
time	2	53.3	27.46	<.0001 *
FA*butyrate	2	53.3	0.31	0.7322
FA*timé	4	53.3	0.81	0.5222
time*butyrate	2	53.3	0.21	0.8084
FA*time*butyrate	4	53.3	0.14	0.9674

Date, butyrate, and time are significant.

The following are the least squares means & standard errors.

Least Squares Means

Effect	FA	time	butyrate	Estimate	Standard Error	DF	t Value	Pr > t
FA FA FA	DHA LA No-FA		·	480.62 437.40 486.84	20.4197 20.4477 20.4460	53.2 53.5 53.5	23.54 21.39 23.81	<.0001 <.0001 <.0001
butyrate butyrate time time		12h 24h	OmM 5 mM	496.62 439.95 559.22 495.26	16.6727 16.7030 20.4351 20.4694	53.2 53.6 53.3 53.7	29.79 26.34 27.37 24.20	<.0001 <.0001 <.0001 <.0001
time FA*butyrate FA*butyrate	DHA DHA	6h	OmM 5mM	350.37 510.91 450.33	20.4088 28.8745 28.8797	53.1 53.2 53.2	17.17 17.69 15.59	<.0001 <.0001 <.0001
FA*butyrate FA*butyrate FA*butyrate FA*butyrate	LA LA NO-FA NO-FA		OmM 5 mM OmM 5 mM	453.44 421.36 525.52 448.16	28.8618 28.9710 28.8933 28.9353	53.1 53.8 53.3 53.6	15.71 14.54 18.19 15.49	<.0001 <.0001 <.0001 <.0001
FA*time FA*time FA*time	DHA DHA DHA	12h 24h 6h	J11114	581.32 482.34 378.19	35.3811 35.3936 35.3257	53.2 53.3 52.9	16.43 13.63 10.71	<.0001 <.0001 <.0001
FA*time FA*time FA*time FA*time	LA LA LA No-FA	12h 24h 6h 12h		522.58 503.03 286.60 573.78	35.4324 35.5000 35.3121 35.3664	53.5 53.9 52.8 53.2	14.75 14.17 8.12 16.22	<.0001 <.0001 <.0001 <.0001
FA*time FA*time time*butyrate time*butyrate	No-FA No-FA	24h 6h 12h 12h	OmM 5mM	500.40 386.33 585.14 533.31	35.4646 35.4060 28.8915 28.9060	53.8 53.4 53.3 53.4	14.11 10.91 20.25 18.45	<.0001 <.0001 <.0001 <.0001
time*butyrate time*butyrate time*butyrate		24h 24h 6h	OmM 5mM OmM	534.02 456.50 370.71	28.8921 29.0025 28.8460	53.3 54.1 52.9	18.48 15.74 12.85	<.0001 <.0001 <.0001
time*butyrate FA*time*butyrate FA*time*butyrate FA*time*butyrate	DHA DHA DHA	6h 12h 12h 24h	5mM OmM 5mM OmM	330.04 619.87 542.77 521.56	28.8774 50.0748 49.9970 50.0267	53.2 53.4 53.1 53.2	11.43 12.38 10.86 10.43	<.0001 <.0001 <.0001 <.0001
FA*time*butyrate FA*time*butyrate FA*time*butyrate	DHA DHA DHA	24h 6h 6h 12h	5mM OmM 5mM	443.12 391.29 365.09	50.0807 49.9324 49.9830 50.0324	53.4 52.8 53 53.2	8.85 7.84 7.30 10.40	<.0001 <.0001 <.0001
FA*time*butyrate FA*time*butyrate FA*time*butyrate FA*time*butyrate	LA LA LA	12h 24h 24h	OmM 5 mM OmM 5 mM	520.52 524.64 531.23 474.83	50.1842 50.0631 50.3441	53.9 53.3 54.5	10.45 10.61 9.43	<.0001 <.0001 <.0001 <.0001
FA*time*butyrate FA*time*butyrate FA*time*butyrate FA*time*butyrate	LA LA NO-FA NO-FA	6h 6h 12h 12h	OmM 5mM OmM 5mM	308.57 264.63 615.02 532.53	49.8718 50.0050 50.0145 50.0160	52.6 53.1 53.2 53.2	6.19 5.29 12.30 10.65	<.0001 <.0001 <.0001 <.0001
FA*time*butyrate FA*time*butyrate	No-FA No-FA	24h 24h	OmM 5mM	549.26 451.55	50.0359 50.2722	53.3 54.2	10.98 8.98	<.0001 <.0001

Now for pairwise comparisons. Estimates of differences, standard errors are given first, then p-values. I will highlight all <0.05 and for interaction terms, ones that have at least one factor in common:

The Mixed Procedure Differences of Least Squares Means

			Difference	3 01 LC	:ast 34	uares mean	3		
Effect	FA	time	butyrate	FA	time	butyrate	Estimate	Standard Error	DF
FA FA butyrate time time	DHA DHA LA	12h 12h	ОтМ	LA NO-FA NO-FA	24h 6h	5mM	43.2168 -6.2198 -49.4366 56.6707 63.9670 208.85	28.8960 28.8950 28.9146 23.5983 28.9222 28.8795	53.3 53.3 53.4 53.3 53.5 53.5
time FA*butyrate	DHA DHA DHA DHA DHA DHA DHA LA	24h	OmM OmM OmM OmM OmM 5mM 5mM 5mM OmM	DHA LA NO-FA NO-FA NO-FA NO-FA LA NO-FA	6h	5mM OmM 5mM OmM 5mM OmM 5mM OmM 5mM	144.88 60.5779 57.4690 89.5426 -14.6111 62.7495 -3.1089 28.9647 -75.1890 2.1716 32.0736 -72.0801	28.9038 40.8374 40.8247 40.9018 40.8767 40.8283 40.9054 40.8507 40.8803 40.8927 40.8380	53.4 53.2 53.5 53.2 53.4 53.5 53.2 53.4 53.4 53.2
FA*butyrate FA*butyrate FA*butyrate FA*butyrate FA*time	LA LA NO-FA DHA DHA DHA DHA DHA DHA DHA DHA DHA DH	12h 12h 12h 12h 12h 12h 12h 12h 12h 24h 24h	OmM 5mM 5mM OmM	NO-FA NO-FA DHA DHA LA LA NO-FA NO-FA NO-FA NO-FA DHA LA	24h 6h 12h 24h 6h 12h 24h 6h 12h 24h	5mM Отм 5mM 5mM	5.2805 -104.15 -26.7932 77.3606 98.9758 203.13 58.7414 78.2937 294.72 7.5444 80.9173 194.99 104.15 4.0.2344 -20.6822	40.8676 40.9151 40.9446 40.8900 50.0444 49.9964 50.0715 50.1193 49.9868 50.0252 50.0946 50.0530 50.0054 50.0886 50.1283	53.3 53.6 53.7 53.3 53.3 53.5 53.5 53.5 53.5 53.5 53.3 53.5 53.5 53.5 53.5 53.5 53.6
FA*time	DHA	24h 24h 24h 24h 6h 6h 6h 12h 12h 12h 12h 12h 12h 24h		LA NO-FA NO-FA LA LA NO-FA NO-FA NO-FA LA NO-FA NO-FA NO-FA NO-FA NO-FA NO-FA	6h 12h 24h 6h 12h 24h 6h 12h 24h 6h 12h 6h 12h 6h		195.75 -91.4314 -18.0585 96.0096 -144.39 -124.84 91.5913 -195.59 -122.21 -8.1452 19.5523 235.98 -51.1970 22.1759 136.24 216.43	49.9958 50.0342 50.1036 50.0620 50.0804 49.9477 49.9862 50.0557 50.0140 50.1554 50.0230 50.0614 50.1308 50.0892 50.0708	53.1 53.2 53.5 53.4 53.2 53.4 52.9 53.3 53.2 53.7 53.2 53.4 53.5 53.4
FA*time time*butyrate	LA LA LA LA NO-FA NO-FA	24h 24h 24h 6h 6h 12h 12h 12h 12h 12h 12h 12h 12h 12h 12	OmM OmM OmM OmM SmM 5mM 5mM OmM	NO-FA NO-FA NO-FA NO-FA NO-FA NO-FA NO-FA	12h 24h 12h 64h 64h 64h 64h 24h 66h 24h 66h 24h 66h	5mM OmM 5mM OmM 5mM OmM 5mM OmM 5mM 5mM	-70.7492 2.6237 116.69 -287.18 -213.80 -99.7365 73.3729 187.44 114.07 51.8229 51.1205 128.64 214.43 255.10 -0.7024 76.8135 162.60 203.28 77.5160 163.31 203.98	50.1092 50.1785 50.1369 49.9766 50.0045 50.0844 50.0122 40.8678 40.8582 40.8255 40.8477 40.8685 40.9462 40.8358 40.8358 40.8366 40.8261 40.8483	53.5 53.87 53.31 53.35 53.35 53.37 5
time*butyrate time*butyrate time*butyrate	24h	24h 6h	5mM 5mM OmM		6h 6h 6h	OmM 5mM 5mM	85.7891 126.46 40.6733	40.9040 40.9261 53 40.8156	53.5 .6 53

The Mixed Procedure

Differences of Least Squares Means

## FA* time butyrate			Dif	ferences o	f Least	Squar	es Means			
### FA*Time*butyrate	Effoct	ΕΛ.	+imo	hutyrato	ΕΛ.	+imo	butyrato	Estimato		DE
FA*Time*butyrate	ETTECL	FA	time	butyrate	FA	t ille	butyrate	ESCIIIALE	EIIOI	DF
FA*Time*butyrate										
FA*Time*butyrate	FA*time*hutvrate	DHA	12h	OmM	DHA	12h	5mM	77 0993	70 7607	53 2
FA*time*butyrate DHA										
FA*Time*butyrate DHA										
FA*time*butyrate DHA 12h 0mM		DHA	12h	OmM	DHA	6h	OmM			53.1
FA*time*butyrate DHA 12h										
FA*Time*butyrate DHA										
FA*time*butyrate DHA										
FA*time*butyrate PhA*time*butyrate PhA*time*buty										
FA*time*butyrate DHA 12h 0mm No-FA 12h 0mm 4,8493 70,7732 53.3 FA*time*butyrate DHA 12h 0mm No-FA 12h 0mm 4,8493 70,7732 53.3 FA*time*butyrate DHA 12h 0mm No-FA 12h 0mm 7,3383 70,7784 53.3 FA*time*butyrate DHA 12h 0mm No-FA 6h 0mm 207.59 70,8204 53.4 FA*time*butyrate DHA 12h 0mm No-FA 6h 0mm 207.59 70,8204 53.4 FA*time*butyrate DHA 12h 0mm No-FA 6h 0mm 21,2060 70,7270 53.1 FA*time*butyrate DHA 12h 5mm DHA 24h 0mm 21,2060 70,7270 53.1 FA*time*butyrate DHA 12h 5mm DHA 24h 0mm 21,2060 70,7270 53.1 FA*time*butyrate DHA 12h 5mm DHA 24h 0mm 121,2060 70,7270 53.1 FA*time*butyrate DHA 12h 5mm DHA 6h 0mm 12,768 70,6959 53.4 FA*time*butyrate DHA 12h 5mm LA 12h 5										
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	FA*time*butyrate	DHA	6h	5mM	No-FA	24h	5mM	-86.4558	70.8906	53.6

FA*time*butyrate FA*tim	DHA LA	6612121212121212121222444hhhhhhhhhhhhhhh	5 mm 5 mm 5 mm 0 mm 5 mm 5 mm 5 mm 5 mm 0 mm	NO-FA NO-FA LA LA LA LA LA NO-FA NO-FA NO-FA NO-FA NO-FA LA	661224hh hhhhh hhhhh hhhhhhhhhhhhhhhhhhhhh	OMM 5 MM	-47.1837	70.7556 70.7411 70.8630 70.7774 70.9763 70.6425 70.7365 70.7432 70.7432 70.77586 70.9254 70.77904 70.7759 70.8834 70.8659 71.0324 70.88518 70.8659 71.0324 70.7661 70.7582 70.7661 70.7683 70.9651 70.9651 70.9651 70.9653 70.9651 70.9653 70.9651 70.9653 70.9651 70.7649 70.7797 70.7975 70.9638 70.9651 70.7975 70.9638 70.9651 70.7975 70.7979 70.7979 70.7964 70.77240 70.77240 70.77567 70.77319 70.77567 70.77319 70.77646 70.77567 70.77567 70.77567 70.77567 70.77567 70.77567 70.77646 70.9283 70.97638 70.9653	22.53.9.9.2.2.2.7.33.6.2.2.5.5.64.6.6.9.5.2.2.33.8.4.3.5.8.8.8.9.4.4.9.8.9.9.9.4.33.1.1.2.7.3.2.2.2.7.3.3.2.7.3.3.2.7.3.3.8.8.4.5.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3
P-VALUES:		···				Squares Me		.010200	33
Effect	FA	t ⁻		rate FA		•	ate t Value	Pr >	t
Effect of FA:									
FA FA FA	DHA DHA LA			LA No- No-			1.50 -0.22 -1.71	0.14 0.83 0.09	304
None significant at a	ιlpha=0.	05							
Effect of butyrate:			Отм			5mM	2.40	0.01	L98 *
Butyrate Significant.			UIIM			MIIIC	2.40	0.01	190
Effect of time:									
time time			2h 2h			4h ih	2.21 7.23	0.03 <.00	313 * 001 *

	time		24h			6h		5.01	<.0001 *
All tir	mes are significan	tly dif	ferent						
Effect	of Fa*butyrate:								
	FA*butyrate	DHA DHA DHA DHA DHA DHA DHA LA LA LA LA LA LA LA		OmM OmM OmM OmM 5mM 5mM 5mM 0mM OmM OmM OmM	DHA LA NO-FA NO-FA LA NO-FA NO-FA NO-FA NO-FA NO-FA NO-FA		5 mM OmM 5 mM 5 mM OmM 5 mM OmM 5 mM 5 mM 5 mM 0 mM 5 mM 5 mM	1.48 1.41 2.19 -0.36 1.54 -0.08 0.71 -1.84 0.05 0.78 -1.77 0.13 -2.55 -0.65 1.89	0.1439 0.1650 0.0330 0.7220 0.1307 0.9396 0.4820 0.0713 0.9578 0.4363 0.0833 * 0.8977 0.0138 0.5157 0.0639 *
				Difference	s of Le	ast Sq	uares Means		
	Effect	FA	time	butyrate	FA	time	butyrate t	Value	Pr > t
Effect	of FA*time:								
	FA*time	DHA	12h 12h 12h 12h 12h 12h 12h 24h 24h 24h 24h 66h 66h 12h 12h 12h 12h 12h 12h 12h 12h 12h 12		DHA DHA LA LA NO-FA NO-FA LA LA LA NO-FA NO-FA LA LA NO-FA LA NO-FA LA NO-FA	24h 124h 124h 124h 6h 6h 124h 6h 6h 6h 6h 6h 6h 6h 6h 6h 6h 6h 6h 6h		1.98 4.06 1.17 5.90 0.15 1.62 3.90 2.08 -0.80 1.92 -1.83 -2.49 1.83 -3.91 1.92 -2.49 1.83 -3.91 1.92 -2.49 1.83 -3.91 1.92 -1.41 0.32 -1.41 0.33 -5.75 -4.27 -1.41 0.33 -5.75 -4.27 -1.46 3.72	0.0531 * 0.0002 * 0.2459 0.1241 <.0001 0.8807 0.1121 0.0003 0.0421 * 0.4253 0.6816 0.0003 0.0732 0.7199 0.0605 0.0056 0.0056 0.0158 0.0723 * 0.0003 0.0180 0.8712 0.6982 <.0001 * 0.3111 0.6600 0.0088 <.0001 * 0.1638 0.9585 0.0237 <.0001 <.0001 <.0001 <.0004 * 0.01488 0.0004 * 0.0269 *
Effect	of time*butyrate:								
	time*butyrate		12h 12h 12h 12h 12h 12h 12h 12h 24h 24h 24h 24h 24h	Omm Omm Omm Omm SmM 5mM 5mM Omm Omm Omm Omm Omm Omm		12h 24h 6h 6h 24h 6h 24h 6h 6h 6h 6h	5 mM 0 mM 5 mM 5 mM 5 mM 5 mM 5 mM 5 mM 5 mM 0 mM 5 mM 0 mM 5 mM 0 mM	1.27 1.25 3.14 5.25 6.25 -0.02 1.88 3.98 4.98 1.89 4.00 4.99 2.10 3.09	0.2103 0.2163 0.0027 <.0001 <.0001 0.9864 0.0661 0.0002 <.0001 * 0.0037 * 0.0002 * 0.00407 0.0032 * 0.3235
Effect	of FA*time*butyra	te							
	FA*time*butyrate FA*time*butyrate FA*time*butyrate FA*time*butyrate	DHA DHA DHA	12h 12h 12h 12h	OmM OmM OmM OmM Differenc	DHA DHA DHA DHA es of L	12h 24h 24h 6h east S	5mM OmM 5mM OmM quares Means	1.09 1.39 2.50 3.23	0.2808 0.1707 0.0157 0.0021 *

Effect	FA	time	butyrate	FA	time	butyrate	t Value	Pr > t
FA*time*butyrate	DHA	12h	OmM	DHA	6h	5mM	3.60	0.0007
FA*time*butyrate	DHA	12h	OmM	LA	12h	OmM	1.40	0.1663
FA*time*butyrate	DHA	12h	OmM	LA	12h	5mM	1.34	0.1848
FA*time*butyrate	DHA	12h	OmM	LA	24h	OmM	1.25	0.2161
FA*time*butyrate	DHA	12h	OmM	LA	24h	5mM	2.04	0.0460
FA*time*butyrate	DHA	12h	OmM	LA	6h	OmM	4.40	<.0001
FA*time*butyrate	DHA	12h	OmM	LA	6h	5mM	5.02	<.0001
FA*time*butyrate	DHA	12h	OmM	No-FA	12h	OmM	0.07	0.9456
FA*time*butyrate	DHA	12h	OmM	No-FA	12h	5mM	1.23	0.2226
FA*time*butyrate	DHA	12h	OmM	No-FA	24h	OmM	1.00	0.3230
FA*time*butyrate	DHA	12h	OmM	No-FA	24h	5mM	2.37	0.0213
FA*time*butyrate	DHA	12h	OmM	No-FA	6h	OmM	2.93	0.0050
FA*time*butyrate	DHA	12h	OmM	No-FA	6h	5mM	3.66	0.0006
FA*time*butyrate	DHA	12h	5mM	DHA	24h	OmM	$0.30 \\ 1.41$	0.7655
FA*time*butyrate	DHA	12h	5mM	DHA	24h	5mM		0.1649
FA*time*butyrate	DHA	12h	5mM	DHA	6h	OmM	2.14	0.0367
FA*time*butyrate	DHA	12h	5mM	DHA	6h	5mM	2.51	0.0150 *
FA*time*butyrate	DHA	12h	5mM	LA	12h	OmM	0.31	0.7543
FA*time*butyrate	DHA	12h	5mM	LA	12h	5 mM	0.26	0.7990
FA*time*butyrate	DHA	12h	5mM	LA	24h	OmM	0.16	0.8710
FA*time*butyrate	DHA	12h	5mM	LA	24h	5mM	0.96	0.3425
FA*time*butyrate	DHA	12h	5mM	LA	6h	OmM	3.32	0.0017
FA*time*butyrate	DHA	12h	5mM	LA	6h	5mM	3.93	0.0002
FA*time*butyrate	DHA	12h	5mM	No-FA	12h	OmM	-1.02	0.3116
FA*time*butyrate	DHA	12h	5mM	No-FA	12h	5mM	0.14	0.8854
FA*time*butyrate	DHA	12h	5mM	No-FA	24h	OmM	-0.09	0.9273
FA*time*butyrate	DHA	12h	5mM	No-FA	24h	5mM	1.29	0.2037
FA*time*butyrate	DHA	12h	5mM	No-FA	6h	OmM	1.84	0.0707
FA*time*butyrate	DHA	12h	5mM	No-FA	6h	5mM	2.58	0.0127
FA*time*butyrate	DHA	24h	OmM	DHA	24h	5mM	1.11	0.2728
FA*time*butyrate	DHA	24h	OmM	DHA	6h	OmM	1.84	0.0709 *
FA*time*butyrate	DHA	24h	OmM	DHA	6h	5mM	2.21	0.0312
FA*time*butyrate	DHA	24h	OmM	LA	12h	OmM	0.01	0.9883
FA*time*butyrate	DHA	24h	OmM	LA	12h	5mM	-0.04	0.9656
FA*time*butyrate	DHA	24h	OmM	LA	24h	OmM	-0.14	0.8919
FA*time*butyrate	DHA	24h	OmM	LA	24h	5mM	0.66	0.5130
FA*time*butyrate	DHA	24h	OmM	LA	6h	OmM	3.02	0.0039
FA*time*butyrate	DHA	24h	OmM	LA	6h	5mM	3.63	0.0006
FA*time*butyrate	DHA	24h	OmM	No-FA	12h	OmM	-1.32	0.1921
FA*time*butyrate	DHA	24h	OmM	No-FA	12h	5mM	-0.16	0.8774
FA*time*butyrate	DHA	24h	OmM	No-FA	24h	OmM	-0.39	0.6971
FA*time*butyrate	DHA	24h	OmM	No-FA	24h	5mM	0.99	0.3280
FA*time*butyrate	DHA	24h	OmM	No-FA	6h	OmM	1.54	0.1285
FA*time*butyrate	DHA	24h	OmM	No-FA	6h	5mM	2.28	0.0268
FA*time*butyrate	DHA	24h	5mM	DHA	6h	OmM	0.73	0.4668
FA*time*butyrate	DHA	24h	5mM	DHA	6h	5mM	1.10	0.2751
FA*time*butyrate	DHA	24h	5mM	LA	12h	OmM	-1.09	0.2792
FA*time*butyrate	DHA	24h	5mM	LA	12h	5mM	-1.15	0.2553
FA*time*butyrate	DHA	24h	5mM	LA	24h	OmM	-1.24	0.2189
FA*time*butyrate	DHA	24h	5mM	LA	24h	5mM	-0.45	0.6571
FA*time*butyrate	DHA	24h	5mM	LA	6h	OmM	1.90	0.0624
FA*time*butyrate	DHA	24h	5mM	LA	6h	5mM	2.52	0.0147
FA*time*butyrate	DHA	24h	5mM	No-FA	12h	OmM	-2.43	0.0186
FA*time*butyrate	DHA	24h	5mM	No-FA	12h	5mM	-1.26	0.2120
FA*time*butyrate	DHA	24h	5mM	No-FA	24h	OmM	-1.50	0.1397
FA*time*butyrate	DHA	24h	5mM	No-FA	24h	5mM	-0.12	0.9059
FA*time*butyrate	DHA	24h	5mM	No-FA	6h	OmM	0.44	0.6649
FA*time*butyrate	DHA	24h	5mM	No-FA	6h	5mM	1.17 0.37	0.2479
FA*time*butyrate	DHA	6h	OmM	DHA	6h	5mM	-1.83	0.7123
FA*time*butyrate	DHA	6h	OmM	LA	12h	OmM		0.0731
FA*time*butyrate	DHA	6h	OmM	LA	12h	5mM	-1.88	0.0651
FA*time*butyrate	DHA	6h	OmM	LA	24h	OmM	-1.98	0.0530
FA*time*butyrate	DHA	6h	OmM	LA	24h	5mM	-1.18	0.2439
FA*time*butyrate	DHA	6h	OmM	LA	6h	OmM	1.17	0.2464
FA*time*butyrate	DHA	6h	OmM	LA	6h	5mM	1.79	0.0788
FA*time*butyrate	DHA	6h	OmM	No-FA	12h	OmM	-3.17	0.0026
FA*time*butyrate	DHA	6h	OmM	No-FA	12h	5mM	-2.00	0.0508
FA*time*butyrate	DHA	6h	OmM	No-FA	24h	OmM	-2.23	0.0297
FA*time*butyrate	DHA	6h	OmM	No-FA	24h	5mM	-0.85	0.3988
FA*time*butyrate	DHA	6h	OmM	No-FA	6h	OmM	-0.30	0.7678
FA*time*butyrate	DHA	6h	OmM	No-FA	6h	5mM	0.44	0.6639
FA*time*butyrate	DHA	6h	5mM	LA	12h	OmM	-2.20	0.0323
FA*time*butyrate	DHA	6h	5mM	LA	12h	5mM	-2.25	0.0284
FA*time*butyrate	DHA	6h	5mM	LA	24h	OmM	-2.35	0.0226
FA*time*butyrate	DHA	6h	5mM	LA	24h	5mM	-1.55	0.1278
FA*time*butyrate	DHA	6h	5mM	LA	6h	OmM	0.80	0.4270
FA*time*butyrate	DHA	6h	5mM	LA	6h	5mM	1.42	0.1612
FA*time*butyrate	DHA	6h	5mM	No-FA	12h	OmM	-3.53	0.0009
FA*time*butyrate	DHA	6h	5mM	No-FA	12h	5mM	-2.37	0.0216
FA*time*butyrate	DHA	6h	5mM	No-FA	24h	OmM	-2.60	0.0119
FA*time*butyrate	DHA	6h	5mM	No-FA	24h	5mM	-1.22	0.2280
FA*time*butyrate	DHA	6h	5mM	No-FA	6h	OmM	-0.67	0.5077
FA*time*butyrate	DHA	6h	5mM	No-FA	6h	5mM	0.07	0.9473
FA*time*butyrate	LA	12h	OmM	LA	12h	5mM	-0.06	0.9539
FA*time*butyrate	LA	12h	OmM	LA	24h	OmM	-0.15	0.8803
FA*time*butyrate	LA	12h	OmM	LA	24h	5 mM	0.64	0.5225
FA*time*butyrate	LA	12h	OmM	LA	6h	0 mM	3.00	0.0041 *
FA*time*butyrate	LA	12h	OmM	LA	6h	5mM	3.62	0.0007
FA*time*butyrate	LA	12h	OmM	No-FA	12h	OmM	-1.34	0.1873

FA*+imo*bu+vra+o	1.4	12h	OmM	No FA	12h	5mM	-0.17	0.8658
FA*time*butyrate	LA		OmM	No-FA				
FA*time*butyrate	LA	12h	OmM	No-FA	24h	OmM	-0.41	0.6863
FA*time*butyrate	LA	12h	OmM	No-FA	24h	5mM	0.97	0.3352
FA*time*butyrate	LA	12h	OmM	No-FA	6h	OmM	1.53	0.1322
		12h			6h		2.26	0.0278
FA*time*butyrate	LA		OmM	No-FA		5mM		
FA*time*butyrate	LA	12h	5mM	LA	24h	OmM	-0.09	0.9263
FA*time*butyrate	LA	12h	5mM	LA	24h	5mM	0.70	0.4865
FA*time*butyrate	LA	12h	5mM	LA	6h	OmM	3.05	0.0035
FA*time*butyrate	LA	12h	5mM	LA	6h	5mM	3.67	0.0006
FA*time*butyrate	LA	12h	5mM	No-FA	12h	OmM	-1.28	0.2076
FA*time*butyrate	LA	12h	5mM	No-FA	12h	5mM	-0.11	0.9117
FA*time*butyrate	LA	12h	5mM	No-FA	24h	OmM	-0.35	0.7297
FA*time*butyrate	LA	12h	5 mM	No-FA	24h	5 mM	1.03	0.3081
FA*time*butyrate	LA	12h	5mM	No-FA	6h	OmM	1.58	0.1189
FA*time*butyrate	LA	12h	5mM	No-FA	6h	5mM	2.32	0.0243
FA*time*butyrate	LA	24h	OmM	LA	24h	5mM	0.79	0.4305
FA*time*butyrate	LA	24h	OmM	LA	6h	OmM	3.15	0.0027 *
FA*time*butyrate	LA	24h	OmM	LA	6h	5mM	3.77	0.0004
FA*time*butyrate	LA	24h	OmM	No-FA	12h	OmM	-1.18	0.2416
FA*time*butyrate	LA	24h	OmM	No-FA	12h	5mM	-0.02	0.9854
FA*time*butyrate	LA	24h	OmM	No-FA	24h	OmM	-0.25	0.7999
FA*time*butyrate	LA	24h	OmM	No-FA	24h	5mM	1.12	0.2664
FA*time*butyrate	LA	24h	OmM	No-FA	6h	OmM	1.68	0.0988
FA*time*butyrate	LA	24h	OmM	No-FA	6h	5mM	2.41	0.0193
FA*time*butyrate	LA	24h	5mM	LA	6h	OmM	2.35	0.0227
	LA	24h	5mM	LA	6h		2.96	0.0045 *
FA*time*butyrate						5mM		
FA*time*butyrate	LA	24h	5mM	No-FA	12h	OmM	-1.98	0.0533
FA*time*butyrate	LA	24h	5mM	No-FA	12h	5mM	-0.81	0.4197
FA*time*butyrate	LA	24h	5mM	No-FA	24h	OmM	-1.05	0.2990
FA*time*butyrate	LA	24h	5mM	No-FA	24h	5mM	0.33	0.7448
FA*time*butyrate	LA	24h	5mM	No-FA	6h	OmM	0.88	0.3823
FA*time*butyrate	LA	24h	5mM	No-FA	6h	5mM	1.61	0.1128
FA*time*butyrate	LA	6h	OmM	LA	6h	5mM	0.62	0.5365
FA*time*butyrate	LA	6h	OmM	No-FA	12h	OmM	-4.34	<.0001
FA*time*butyrate	LA	6h	OmM	No-FA	12h	5mM	-3.17	0.0025
FA*time*butyrate	LA	6h	OmM	No-FA	24h	OmM	-3.41	0.0013
FA*time*butyrate	LA	6h	OmM	No-FA	24h	5mM	-2.02	0.0485
FA*time*butyrate	LA	6h	OmM	No-FA	6h	OmM	-1.47	0.1482
FA*time*butyrate	LA	6h	OmM	No-FA	6h	5mM	-0.73	0.4666
FA*time*butyrate	LA	6h	5mM	No-FA	12h	OmM	-4.95	<.0001
FA*time*butyrate	LA	6h	5mM	No-FA	12h	5mM	-3.79	0.0004
FA*time*butýrate	LA	6h	5mM	No-FA	24h	OmM	-4.02	0.0002
FA*time*butyrate	LA	6h	5 mM	No-FA	24h	5mM	-2.64	0.0109
							-2.09	
FA*time*butyrate	LA	6h	5mM	No-FA	6h	OmM		0.0418
FA*time*butyrate	LA	6h	5mM	No-FA	6h	5mM	-1.35	0.1816
FA*time*butyrate	No-FA	12h	OmM	No-FA	12h	5mM	1.17	0.2487
FA*time*butyrate	No-FA	12h	OmM	No-FA	24h	OmM	0.93	0.3568
FA*time*butyrate	No-FA	12h	OmM	No-FA	24h	5mM	2.31	0.0250
FA*time*butyrate	No-FA	12h	OmM	No-FA	6h	OmM	2.86	0.0060 *
FA*time*butyrate	No-FA	12h	OmM	No-FA	6h	5mM	3.60	0.0007
FA*time*butyrate	No-FA	12h	5mM	No-FA	24h	OmM	-0.24	0.8140
FA*+imo*butyrate		12h	5mM		24h	5mM	1.14	0.2585
FA*time*butyrate	No-FA			No-FA				
FA*time*butyrate	No-FA	12h	<u>5</u> mM	No-FA	6h	<u>O</u> mM	1.70	0.0951
FA*time*butyrate	No-FA	12h	5mM	No-FA	6h	5mM	2.43	0.0184 *
FA*time*butyrate	No-FA	24h	OmM	No-FA	24h	5mM	1.38	0.1740
FA*time*butyrate	No-FA	24h	OmM	No-FA	6h	OmM	1.93	0.0583 *
							2.67	
FA*time*butyrate	No-FA	24h	QmM	No-FA	6h	5mM		0.0101
FA*time*butyrate								
	No-FA	24h	5mM	No-FA	6h	OmM	0.55	0.5823
FA*time*butyrate	No-FA No-FA	24h 24h	5mM 5mM	No-FA No-FA	6h 6h	OmM 5mM	0.55 1.28	0.5823 0.2043

Analysis of Controls for Both Days

DF

Error

t Value

Pr > |t|

For 8/11/03, the procedure was run in proc mixed using a subsampling term for well. Some problems with the data – groups are very variable.

Type 3 Tests of Fixed Effects Den Effect F Value Pr > FDF cont_2 443 657.44 <.0001 Least Squares Means Standard Error Effect cont_2 Estimate DF t Value Pr > |t|cont_2 cont_2 cont_2 H2O250 cum100 cum25 667.96 1677.22 130.97 28.0391 30.0775 31.6323 443 443 443 <.0001 <.0001 <.0001 Differences of Least Squares Means Standard

Estimate

Effect

cont_2

cont_2

cont_2	H2O250	cum100	-1009.26	41.1199	443	-24.54	<.0001
cont_2	H2O250	cum25	536.99	42.2705	443	12.70	<.0001
cont_2	cum100	cum25	1546.25	43.6493	443	35.42	<.0001

All significantly different.

9/1/03, since only one well, anova was run. Data set very variable – not confident about the results.

Type 3 Tests of Fixed Effects

Effect	Num DF	Den DF	F Value	Pr > F
control	2	712	257.48	<.0001

Least Squares Means

Effect	control	Estimate	Standard Error	DF	t Value	Pr > t
control	DH2O2	459.35	18.3452	712	25.04	<.0001
control	H2O2	1123.52	23.7868	712	47.23	<.0001
control	Cum	509.58	35.8173	712	14.23	<.0001

Differences of Least Squares Means

Effect	control	control	Estimate	Standard Error	DF	t Value	Pr > t
control control control	DH2O2-50uM DH2O2-50uM H2O2-5uM		-664.18 -50.239 613.94	30.0393 40.2421 42.9964	712 712 712	-22.11 -1.25 14.28	<.0001 0.2123 <.0001

Dh202 and cum not significantly different from each other.

Table B-7. Summary of Fatty Acid and Butyrate Treatment Effects

Treatments	Results	Mitochondrion
		West of the second
	(Compared to LA treatment)	
	• Did not induce lipid oxidation	
DHA incubation	• Greater effect: ↑ basal MP (50 µl	M)
(72 h)	• ↑ basal MP (dose dependent)	
Butyrate	(In the absence of fatty acid)	
incubation	• ↑ lipid oxidation (no dose effect)
	(Compared to LA and butyrate co	trantment)
DHA and butyrate co-incubation	• ↑ mitochondrial lipid oxidation (
co-mediation	• Greater effect: ↑ lipid oxidation	
	• \dime dependent)	(S HEVI BURYTURE) CUIT
	• Greater effect: ↓ MP (24 h)	· · ·

		Apoptosis (?)
		Apoptosis (1)
		THE STREET, ST

Table B-8. Outline of research proposal

Hypot proces:	Hypothesis: The incorporation of n-3 PUFA increases cell susceptibility to oxidative damage, and subsequently perturbs the integrity of mitochondrial membranes, a process potentiated by butyrate.	ell suscep	ribility to oxidative damage, and subsequen	ntly pertu	rbs the	integrity of mitochondrial membranes, a
Objective mitochon colonocyt (FA only)	Objective 1: To quantify total cellular and mitochondrial membrane lipid oxidation in colonocytes following n-3 and n-6 PUFA incubation. (FA only)	Objective between mitochond (FA only)	Objective 2: To determine the association between membrane lipid oxidation and mitochondrial function. (FA only)	Objective 3: To production, mitocoxidation in coloand n-6 PUFA. (FA and Butyrate)	ive 3: ion, n on in PUF ₂	Objective 3 : To determine if butyrate potentiates ROS production, mitochondrial functional alteration and lipid oxidation in colonocyte membranes enriched with n-3 and n-6 PUFA. (FA and Butyrate)
444C :noitabixo t	Global membrane effects: • Does lipid oxidation response to the change in FA dose? • Does VEs prevent oxidation? Mitochondrial membrane effects: • Does MitoO prevent oxidation?	1123	Mitochondrial membrane effects: Does FA or butyrate treatment	:noitabixo biqiJ	• •	Does butyrate/isobutyrate treatment induce lipid oxidation in global and mitochondrial membranes? Does butyrate/FA co-treatment induce lipid oxidation in global and mitochondrial membranes?
DCFDA Lipid	Does prevention of ROS prevent lipid oxidation? Global membrane effects: Does ROS production response to the change in FA dose?	odA : 4M To noitsedies	dissipate MP? Is the dissipation of MP inhibitable by Cyclosporin A? How and why would MP changes with FA treatment at different doses? Does the prevention of lipid oxidation by VEs and MitoO prevent MP	Cytosolic ROS	• •	Does butyrate/isobutyrate treatment induce ROS production in mitochondrial matrix? Does butyrate/FA co-treatment induce ROS production in mitochondrial matrix?
Cytosolic ROS:	 Does VEs prevent ROS production? Mitochondrial membrane effects: TPPB prevents ROS production? 	рiQ	dissipation?	dМ	• •	Does butyrate/isobutyrate treatment dissipate MP? Does butyrate/FA co-treatment dissipate MP?
Abbrev MitoQ (oxygen	Apoptotic Phenotype Abbreviations used: CMH ₂ -DCFDA (5-(and-6)-chloromethyl-2',7'- dichlorodihydrofluorescein diacetate, acetyl ester), DPPP (diphenyl-1-pyrenylphosphine), FA (fatty acids), MitoQ (10-(6'-ubiquinoyl) decyltriphenylphosphonium bromide), MP (mitochondrial membrane potential), PUFA (polyunsaturated fatty acids), Rhd 123 (rhodamine 123), ROS (reactive oxygen species), TPPB (2-(2-(triphenylphosphonio))ethyl)-dihydro-H-benzopyran-6-ol bromide), VEs (vitamin E succinate)	omethyl-2'), MP (m iro-H-ben:	Apoptotic Phenotype ,7- dichlorodihydrofluorescein diacetate, acetyl e itochondrial membrane potential), PUFA (polyun zopyran-6-ol bromide), VEs (vitamin E succinate	ester), DP nsaturated	PP (dij	ohenyl-1-pyrenylphosphine), FA (fatty acids), cids), Rhd 123 (rhodamine 123), ROS (reactive

APPENDIX C

RESULTS FROM PRELIMINARY EXPERIMENTS

Result C-1: Mitochondrial Membrane Potential Dissipation in Viable Cells

Purpose: To determine the utility of 2 fluorescence probes, TMRM and calcein AM, to measure mitochondrial membrane potential (MP) in viable YAMC cells.

Rationale:

The tetramethylrhodamine methyl ester perchlorate (TMRM) (Molecular Probes, #T668) is a lipophilic cationic fluorescent probe, which accumulates in the mitochondria in proportion to MP (77). In comparison, calcein AM (Molecular probes, #C3100) is a cell-permeant dye that can be used to determine cell viability in most eukaryotic cells. In viable cells, the nonfluorescent calcein AM is converted to a green-fluorescent calcein after acetoxymethyl ester hydrolysis by intracellular esterases (Molecular probe).

To determine the validity of TMRM and calcein AM to measure the change in MP in viable YAMC cells, following butyrate incubation with or without 50 μ M DHA or LA pretreatment, cells were loaded with TMRM and calcein AM as indicated in the protocol.

TMRM and Calcein dye loading protocol:

- 1. Add 1 ml RPMI complete media with γ-IFN into each well of YAMC cells
- 2. Add 1.0 µl TMRM (150 nM final conc.)
- 3. Incubate at 33°C, at least 10 min
- 4. Add 1.0 μl calcein-AM (1 μM final conc.)
- 5. Incubate at 33°C, for another 10 min
- 6. Aspirate off media with dyes
- 7. Add 1 ml fresh Leibovitz media (without phenol red)
- 8. Image for 0, 5, 10 and 15 min.

Butyrate, isobutyrate (negative control), DHA and LA were added to the media as described in protocol # A-5. Each experiment was designed to optimize the protocol, hence the order and duration of butyrate incubations varied. Magnification power was set at 200 or 400, images of cells (about 30 cells per field) were captured by a phase-contrast, fluorescent, widefield inverted Nikon microscope. The fluorescence excitation/emission wavelengths were 562/624 nm for TMRM, and 477/536 nm for calcein AM. Ten representative cells from each captured image were randomly selected for fluorescence intensity quantification. Data were plotted using mean (+/-SEM) average pixel intensity of all cells.

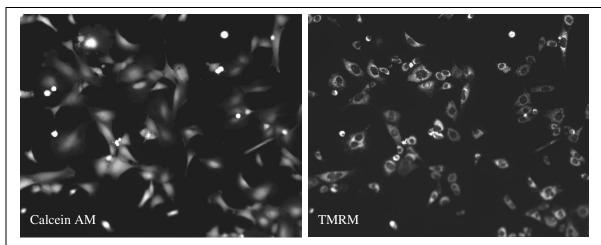


Fig. C-1. Representative photomicrographs of YAMC cells illuminated following loading with TMRM at 150 nM for 10 min and subsequent incubation with calcein AM at 1 μ M for 10 min. Two images of each same field were captured using different excitation and emission filters.

Results:

- 1) A kinetic study (2002.02.15) was also performed to look at the effect of short duration butyrate treatment at 5 mM for up to 45 min, post DHA or LA treatments, on MP and cell viability. Butyrate incubation for up to 45 min did not induce MP dissipation, using TMRM and calcein AM as two fluorescence probes.
- 2) A representative set of data (2002.03.07) is presented to show that DHA- and LA-BSA pretreatment, at 50 μ M, did not alter MP, using TMRM as a fluorescent probe. In this specific set of experiments, 5 mM butyrate treatment was incubated for 2, 6, 12 or 24 h, with or without 50 μ M DHA or LA pre-treatment in YAMC cells for 72 h. See table Appendix C-1 for results.
- 3) To optimize imaging outcome, YAMC cells treated with PUFA and butyrate were analyzed using confocal microscopy and TMRM (2002.05.10). The results from 2003.03.07 (butyrate treatment dissipate MP) were not reproducible.

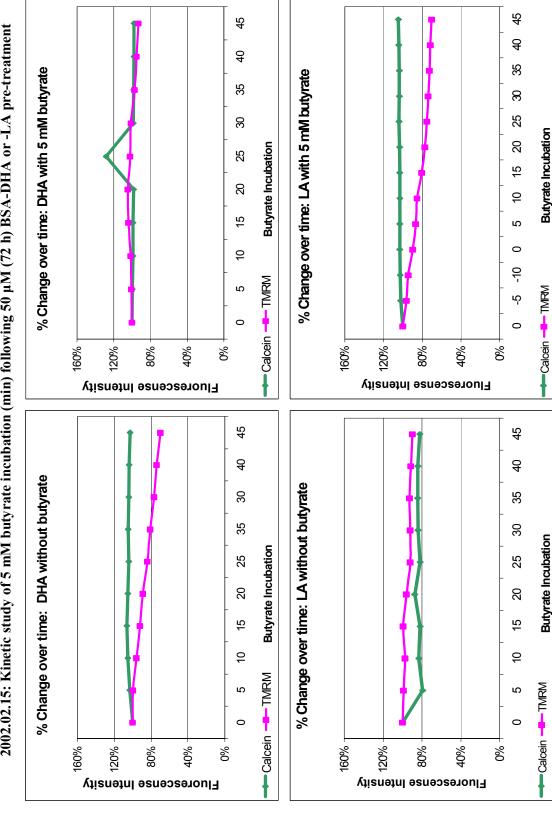
Conclusion:

We have previously demonstrated that MP was dissipated by butyrate incubation (5 mM) for 24 h (47). This experiment using TMRM could not consistently reproduce the effect. We conclude that TMRM might not be a suitable probe to measure MP dissipation in YAMC cells.

References:

Molecular probes

http://www.probes.com (last accessed: May 20, 2004)



2002.02.15: Kinetic study of 5 mM butyrate incubation (min) following 50 μM (72 h) BSA-DHA or -LA pre-treatment

2002.03.07 Preliminary study: YAMC-Fatty acid +/- 2 h butyrate

	A DHA+/- 2 h Butyrate		110%Calcein - But			100% — Calcein + But			%06 BB	, sə			0 5 10 15		Time (min)		B. LA +/- Z n Butyrate		110% Calcein - But			International Calcain + But	<u></u>	mo TMRM + But	%06 expenses			80%	0 5 10 15		Time (min)	
%FI/OI	100%	103%					. 85%		100%			102%	100%	94%	91%	87%	100%	103%		103%		94%		81%		101%			100%	%56	%28	81%
Average Pixel Intensity	312.14	320.016	322.486	320.842	272.967	255.065	231.654	222.497	143.146	144.588	145.658	146.554	191.015	179.494	174.476	166.627	172.629	176.994	179.153	177.298	230.78	216.967	201.692	186.983	146.798	148.567	149.463	149.543	258.878	246.137	225.107	209.154
Probe	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM
Dye loading	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min
Butyrate	+	+	+	+	+	+	+	+	1	ı	ı	ı	ı	ı	ı	1	+	+	+	+	+	+	+	+	ı	1	ı	1	ı	1	,	1
FA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	ΓA	LA	LA	LA	LA	LA	LA	LA	LA	ΓA	LA	LA	LA	ΓA	LA	LA

2002.03.07 Preliminary study: YAMC-Fatty acid +/- 6 h butyrate

A DHA+/- 6 h Butyrate	100	- Calcein - But	110% Them. But		0	8/80	GC TWRM+ But	%06 uə	cose		- 1 m 80% + 1 m 1 m 1 m 1 m 1 m 1 m 1 m 1 m 1 m 1	H 0 5 10 15		Time (min)			B. LA+/- 6 h Butyrate		110%	sities ————————————————————————————————————		100%	Ce TMRM + But	»õõ		ore	%08 pnl=		2	Time (min)		
%FI/OI	100%	100%	102%	101%	100%	111%	%88	%08	100%	102%	102%	%66	100%	84%	71%	%59	100%	106%	107%	110%	100%	%86	%26	%68	100%	%86	100%	%26	100%	%06	87%	82%
Average Pixel Intensity	325.248	324.587	332.729	329.499	242.144	269.925	212.458	192.886	197.385	200.628	201.122	194.639	185.385	156.332	132.222	120.055	182 758	193.796	196.027	200.191	159.511	155.825	147.487	142.15	289.723	283.177	288.949	281.283	216.907	195.97	189.7	177.575
Probe	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM
Dye loading	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min
Butyrate	+	+	+	+	+	+	+	+	1	1	1	ı		1	1	1	+	+	+	+	+	+	+	+	1	ı	1	1	ı	ı	1	1
FA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	T A	ΓĄ	LA	ΓA	LA	ΓA	ΓA	LA	LA	ΓA	ΓA	LA	ΓA	LA	ΓA	LA

2002.03.07 Preliminary study: YAMC-Fatty acid +/- 12 h butyrate

	A. DHA +/- 12 h Butyrate		110% Telephone T			100% — Calcein + But			%06 909	ore:		0.00	0 5 10 15	I the state of the	IIme (min)		B. LA +/- 12 h Butyrate	•	110% - Calcein - But			7000/	In 100% ——————————————————————————————————	90		%06 pse	nor		_	0 5 10 15	Time (min)		
%FI/OI	100%	102%	104%	103%	100%	93%	85%	83%	100%	101%	102%	102%	100%	94%	%06	%88		100%	103%	105%	102%	100%	%96	%06	83%	100%	100%	101%	101%	100%	%26	91%	%88
Average Pixel Intensity	228.672	233.828	238.943	236.062	319.004	297.336	272.185	263.499	123.337	124.625	125.209	125.717	146.152	137.4	131.274	128.661		170.98	175.947	178.935	175.217	262.932	251.291	236.55	218.683	130.154	130.143	131.584	131.537	127.332	116.839	115.256	112.434
Probe	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM		Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM	Calcein	Calcein	Calcein	Calcein	TMRM	TMRM	TMRM	TMRM
Dye loading	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min		0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min	0 min	5 min	10 min	15 min
Butyrate	. +	+	+	+	+	+	+	+	ı	1	ı	1	ı	1	1	1		+	+	+	+	+	+	+	+	ı	1	1		ı	1	1	
FA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA	DHA		LA	LA	LA	LA	LA	LA	ΓA	LA	LA	LA	LA	LA	ΓA	ΓA	LA	LA

Calcein + But Calcein + But ■Calcein - But -Calcein - But - TMRM + But — TMRM + But -TMRM - But -- TMRM - But B. LA +/- 24 h Buty A. DHA +/- 24 h Buty 5 5 10 9 Time (min) Time (min) S Ŋ 0 0 %08 110% 100% %06 110% 100% %06 80% Fluorescence Intensity Fluorescence Intensity 100% 100% 102% 100% 100% 102% 103% 102% %001 103% 104% 105% %001 104% 101% %86 94% 95% 90% %66 93% 94% 90% %06 %06 Average Pixel Intensity 326.143 319.082 293.546 228.884 226.219 168.629 171.265 226.366 211.36 204.639 215.939 161.088 159.664 172.299 306.915 432.223 412.704 389.536 211.454 326.685 306.024 294.734 225.307 224.871 369.504 223.651 208.225 214.025 Calcein TMRM TMRM Calcein Calcein TMRM Calcein Calcein TMRM Calcein Calcein Calcein Calcein TMRM Calcein TMRM Calcein **IMRM** IMRM Calcein **FMRM FMRM** Calcein Calcein IMRM **FMRM FMRM FMRM** Probe Dye loading 10 min 15 min 10 min 15 min 10 min 15 min 10 min 10 min 15 min 10 min 15 min 5 min 5 min 10 min 15 min 5 min 5 min 0 min 0 min 15 min 0 min 5 min 0 min 5 min 0 min 5 min 0 min Butyrate DHA

2002.03.07 Preliminary study: YAMC-Fatty acid +/- 24 h butyrate

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EDUCATION

Combined Graduate Degree – Dietetic Internship, Texas A&M University, College Station, TX

- ☐ General-emphasis Dietetic Internship, July 2004
- □ Master of Science, Nutrition, August 2004
 - o GPR: 3.6/4.0
 - Thesis: To investigate the mechanism by which docosahexaenoic acids and butyrate prevent colon cancer development

Bachelor of Human Resource and Family Sciences-Dietetics

- □ University of Nebraska, Lincoln, 2001
 - o GPR: 4.0/4.0

EXPERIENCES

<u>Volunteer Nutrition Educator</u>, part-time, Scott and White Renal Dialysis Center, Temple, TX, 2004

Teaching Assistant, Texas A&M University, College Station, TX 2002-03

Teaching Assistant, University of Nebraska, Lincoln, NE 2001

<u>Peer Educator</u>, Students Helping Students Nurture Esteem, University of Nebraska, Lincoln, NE 2000-01

Research Assistant, University of Nebraska, Lincoln, NE 2000

<u>Assistant Nutrition Therapist</u>, part-time, Madonna Rehabilitation Hospital, Lincoln, NE, 2000

PROFESSIONAL/HONORS

- □ Malaysian Student Organization, Treasurer, Texas A&M University, 2003-2004
- ☐ American Dietetic Association, member, 2003-present
- ☐ American Society for Nutritional Sciences, 2003-present
 - First Prize student poster competition, 2004
- □ Academic Excellence Award, Kojadi, Malaysia, 2003
- □ Malaysia Embassy Award, U.S.A., 2001